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The influence of nanoclays on the mechanical and thermal properties of rigid PIR and PUR foams

RUSLAN R. VLASOV¹*, DARIA I. RYABOVA¹, SAKINA Z. ZEYNALOVA², DMITRY V. SOKOLOV¹ and SERGEI A. RYABOV¹

¹Lobachevsky State University – Faculty of Chemistry, Department of Macromolecular Compounds and Colloidal Chemistry, Gagarina av. 23, 603098 Nizhny Novgorod, Russia and ²Tre Tau Engineering srl, Pietro Colletta 85, 10153 Torino, Italy

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Abstract: The effect of small amounts of chemically modified nanosized clays (from 0.05 to 1 %) on the morphological, physical-mechanical and thermophysical characteristics of rigid polyurethane–polyisocyanurate (PIR) and polyurethane (PUR) foams has been studied. The effect of these additives on the structure of the resulting material, the change in its compressive strength, Young's modulus, mass loss during combustion, and thermal conductivity are evaluated. Based on the results obtained, it is noted that the addition of small amounts (up to 0.2 %) of chemically modified Cloisite 30B nanoclay effectively reduces the average cell size of nanocomposite foams, which leads to an improvement in their performance.

Keywords: polyurethane; polyurethane–polyisocyanurate foam; nanosized fillers; compressive strength; cell morphology; thermal conductivity.

INTRODUCTION

Currently, one of the largest molar weight polymers is polyurethane – a polymeric material, traditionally obtained by polycondensation (polyaddition) of aromatic or aliphatic di- and polyisocyanates with polyol resins.¹ Due to the distinctive features of the material, such as insulation ability from high temperature and sound in the case of polyurethane foams, and exceptional mechanical properties, polyurethanes have now become an integral part of everyday life. One of the main industrially significant types of polyurethane is rigid polyurethane (PUR) and polyurethane–polyisocyanurate (PIR) foam. PIR foams are obtained using isocyanate trimerization catalysts as a result of technological processes similar to those in the case of PUR foams. It should be noted that PIR foams have improved thermal insulation properties and are less flammable than conventional



^{*} Corresponding author. E-mail: vlasovruslan.hs@yandex.ru https://doi.org/10.2298/JSC221103089V

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polyurethane foams. Reactions of the formation of urethane and isocyanurate groups are shown in Fig. $1.^2$ It is important to note that in the framework of the synthesis of both polyurethane and, mainly, polyurethane–polyisocyanurate foams, a number of secondary chemical processes occur, the consideration of which is beyond the scope of this publication.^{3,4}



Fig. 1. Reactions of urethane and isocyanurate formation.

It is important to note that the annually growing demand for polymeric materials inevitably leads to ever greater requirements of functional (physical-mechanical, thermophysical, *etc.*) characteristics of the resulting polymers, which the existing ones, for a very wide range of applications, no longer meet. One of the most promising ways to solve this problem is the synthesis of composite materials based on these polymers. Today, the filled polyurethane systems are used everywhere, while the mass fraction of the filler in the formulation can traditionally vary from 1 to 50 % by weight of the composition. The most common fillers currently used in the polyurethane industry are chalk, carbon black,⁵ sand,⁶ expandable graphite,⁷ titanium dioxide⁸ and a wide range of other oxides and inorganic salts.

It should be noted that the introduction of such inorganic additives in acceptable amounts into systems for producing foams contributes to a decrease in the average cell size of the formed foam,⁹ since fillers act as a nucleating agent, creating many smaller cells.^{10,11} In addition, these additives prevent the additional growth of bubbles due to an increase in the viscosity of the reaction medium.¹² Thus, the use of nanosized fillers to a large extent contributes to the improvement of the morphological characteristics of foams and, as a result, to an increase in a number of performance characteristics, increasing strength indicators^{13,14} to a significant extent and preventing a gradual increase in the thermal conductivity of foams during aging.^{15,16}

The trend towards the addition of micro-sized clays (mainly montmorillonite and vermiculite) in polyurethane foam compositions arose about 20 years ago.^{17–19} Nanoclays, on the other hand, are finely dispersed material, which improves the mechanical and thermal insulation properties of polyurethane composite foams due to a significantly increased specific surface area.²⁰

It is extremely necessary to select the optimal amount of nanofiller, since at an excessively high content of these additives, a significant deterioration in the morphology of the material and degradation of performance characteristics are noted.^{21,22} This deterioration in morphology can manifest itself both as an increase in the average cell size¹³ and as a decrease in their density,²³ as well as an increase in the anisotropy coefficient of the foam.^{24,25}

Thus, in the case of polyurethane and polyurethane–polyisocyanurate foams, nanoclays are of significant interest. In this regard, it is expedient to develop formulations of new nanocomposite materials based on polyurethane–polyisocyanurate and polyurethane foams, using the additives described above, which possess some improved performance characteristics.

EXPERIMENTAL

Materials

For the synthesis of polyurethane and polyurethane-polyisocyanurate foams, a mixture of polyether polyols (H6007 (HongBaoLi PU, China) and HF-310 (Zhejiang Hengfeng New Material, China)), tris(2-chloropropyl)phosphate (TCPP, Shijiazhung Hejia Chemical Products, China) as a flame retardant, a mixture of 1,1,4,7,7-pentamethyldiethylenetriamine (PMDETA, Evonik, Germany) and dibenzylamine (Evonik, Germany) as catalysts for urethane and urea formation, a solution of potassium acetate in diethylene glycol in a ratio of 30:70 (Evonik, Germany) as trimerization catalyst, as well as water and *n*-pentane (for synthesis, Ekos-1, Russia) as blowing agents, were used. A highly functional polymeric MDI – Lupranat M50 (BASF, Germany) – with dynamic viscosity of 557 mPa s, determined at 25 °C in accordance with ISO 3219-2:2021,²⁶ and NCO content of 31.5 %, determined in accordance with ISO 148696:2009,²⁷ was used as the isocyanate component. The composition of the systems used is shown in Table I.

| Common out | Fo | am |
|--|------|-------|
| Component | PUR | PIR |
| Polyether polyols blend | 50.0 | 50.0 |
| TCPP | 7.0 | 7.0 |
| PMDETA | 0.8 | 0.8 |
| Dibenzylamine | 2.1 | 2.1 |
| Potassium acetate in diethylene glycol (30:70) | _ | 2.5 |
| Water | 2.4 | 2.4 |
| <i>n</i> -Pentane | 2.7 | 2.7 |
| Polymeric MDI (Lupranat M 50) | 62.3 | 155.8 |

TABLE I. The composition (amount in g) of PIR/PUR foam systems

Commercially available chemically modified organoclays Cloisite 30B (surface layer modified with a quaternary salt of bis-2-hydroxyethylmethylammonium, modifier content <90 meq/100 g) and Cloisite 25A (surface layer modified with a quaternary salt of diethylmethyl-ammonium, content modifier <95 meq/100 g) manufactured by BYK Additives (Germany).

Nanoclay suspensions

The insertion of fillers into the isocyanate component to obtain PUR and PIR foams was carried out by ultrasonic dispersion using an Inlab 1100-6/4 unit equipped with an 110-2.0 ultrasonic generator at a frequency of 22.5 kHz with an output power of 2 kW.

Dynamic viscosity measurements

The dynamic viscosity of the suspensions obtained was measured using a Lamy RM200 CP4000 Plus rotary rheometer using R-6 disc (\emptyset 14.62 mm) at shear rate of 60 rpm in accordance with ISO 3219-2:2021²⁶ at 25 °C. Further in the work, the average viscosity values determined by analyzing three samples are given.

PUR and PIR foams preparation

The isocyanate and polyol components were thermostated at 20 °C. To obtain foam samples, the required weights of the blended polyether component were placed in plastic cylinders with a volume of 1000 cm³, the required amount of *n*-pentane was added, after which it was actively mixed with the component for 2 min. After adding the required weight of isocyanate (or nanoclay suspensions), the system was mixed for 5 s using an automatic mixer at a speed of 3000 rpm. Foaming was carried out in the above-described plastic containers or wooden cubes with a volume of 15625 cm³. Ten min after mixing the components, fragments were cut out from the core of the formed foams, which were subsequently used as analysed samples.

Gel time

The so called "gel time" was tested as the time elapsed from the start of mixing polyol and isocyanate components to the point at which soft threads are formed when an ordinary wire yarn is touched to the surface of the foaming material and withdrawn.

Isocyanate group content measurements

The content of isocyanate groups in the polyisocyanate used was determined using the potentiometric titration method on an 848 Titrino plus Metrohm automatic titrator in accordance with ISO 148696:2009.²⁷

Apparent density

The determination of the apparent density of the synthesized foams was carried out in accordance with ASTM D1622-20,²⁸ for which cubic samples of $50 \times 50 \times \text{mm}^2$ in size were cut from the obtained materials. Further in the work, the average apparent density values, determined by analysing five samples, are given.

Mechanical tests

The determination of compressive strength and Young's modulus was carried out in the direction of foaming on Roell/Zwick Z005 universal testing machine at a strain rate of 10 mm min⁻¹ at 10 % linear strain in accordance with the EN 826:2013 method.²⁹ Within this work, only the relative compressive strength and Young's modulus are considered, numerically equal to the ratio of the compressive strength and Young's modulus to the square of the apparent density of the foam sample. The data shown in the work are averages calculated from analyses of five samples of each of the investigated formulations of polyurethane–polyiso-cyanurate and polyurethane foams.

Open cell content

The content of open cells (OC / %) in the analysed polyurethane and polyurethane–polyisocyanurate foams was analysed using an AccuPyc II 1340 gas pycnometer in accordance

with ASTM D6226-05.³⁰ Further, the paper presents the average value of the content of open cells in the studied samples, determined from the results of three independent measurements.

Thermal conductivity

The thermal conductivity of the analysed polyurethane-polyisocyanurate and polyurethane foams was determined at 24 °C using LaserComp Fox600 HFM and Fox200 SN instruments in accordance with EN 12667:2001.³¹ The size of each sample was 200 mm×200 mm×25 mm. Further, the paper presents the average values of thermal conductivity determined by analysing three samples.

Morphological characteristics analysis

The morphological characteristics of the closed-cell structure of the foam samples were studied using the method of scanning electron microscopy (SEM). A Hitachi TM4000 Plus II instrument was used as a scanning electron microscope. Using a blade, 0.5 mm thick layers of the analysed foam were cut out, and they were subsequently placed in the sample compartment of the above-described photomicrograph instrument.

The average cell size (Φ) and the anisotropy coefficient ($A_{y/x}$) were estimated using the ImageJ software with the help of the intersection method, in accordance with the methods proposed by Brondi *et al.*³² Each analysed micrograph was overlaid with a grid composed of equidistant perpendicular lines (*m* vertical lines of length *h* and *l* horizontal lines of length *w*). For each of the lines, the number of crossed cells (n_i , n_j) was counted. Dividing the length of each individual line by a certain number of intersections, we found the lengths of the chords of the cells (Φ_i and Φ_j , *i.e.*, one-dimensional values). This operation was carried out for each of the lines of the constructed grid, after which the average cell size was determined in accordance with the Eq. (1) and the anisotropy coefficient in accordance with the Eq. (2). Further in the paper, the averaged values of all these quantities, determined by analyzing five samples, are given:

$$\overline{\Phi} = \frac{\sum_{i,j=1}^{m,l} \Phi_{i,j}}{m+l}$$
(1)

where Φ is average cell size, $\Phi_{i,j}$ are the lengths of cell chords, *m* is the number of vertical grid lines, *l* is the number of horizontal grid lines.

$$A_{y/x} = \left(\left(\sum_{i=1}^{m} \boldsymbol{\Phi}_{i} \right) / m \right) \left(\left(\sum_{j=1}^{l} \boldsymbol{\Phi}_{j} \right) / l \right)$$
(2)

where $A_{y/x}$ is the anisotropy coefficient, $\Phi_{i,j}$ are the lengths of cell chords, *m* is the number of vertical grid lines, *l* is the number of horizontal grid lines.

Cell density was estimated using ImageJ software according to Kumar's theoretical approximation principle.^{33,34} After opening the microphotograph in the ImageJ program, the area for analysis was selected. The number of cells and the square of the analysed area of the micrograph were determined. The density of cells in the volume was estimated in accordance with the Eq. (3) proposed by Kumar, which can be expressed in terms of the number of cells located per cubic centimetre of the analysed foam (cells cm⁻³). Further in the work, the average values of cell density determined by analysing five samples are given:

$$N_{\rm f} = \left(\frac{n}{A}\right)^{3/2} \tag{3}$$

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where $N_{\rm f}$ is the density of cells in the foam, *n* is the number of cells in the analysed area of the microphotograph, *A* is the square of the analysed area, cm².

Combustibility

In order to assess the change in the combustibility of the analysed polyurethane–polyisocyanurate foams, samples (30 mm×30 mm×15 mm) were vertically fixed in a tripod, after which the edge (30 mm×30 mm) was blown over for 30 s by the flame of a gas burner located at a distance of 5 cm from the sample, so that the entire surface of the facet was covered in flames. The gas flow rate was 0.36 ml s^{-1} . Further, the paper presents the average values of the degree of damage by mass (S_m) determined by analysing five samples, calculated taking into account the change in the mass of the analysed samples before and after the test.

RESULTS AND DISCUSSION

In accordance with the ISO 3219-2:2021 method,²⁶ the dynamic viscosity of the resulting organoclay suspensions in the polyisocyanate component used was determined. The trend in dynamic viscosity versus filler content is shown in Fig. 2.





Thus, there is a gradual increase in the dynamic viscosity of the isocyanate component as the content of the filler in it increases.

In order to study the effect of small additions of organoclays on the physicalmechanical, thermophysical and morphological properties, the samples of rigid closed-cell polyurethane and polyurethane–polyisocyanurate foams were obtained using the suspensions of the filler in polyisocyanate. It is noted that the addition of the considered fillers into the composition slightly shifted the parameters of the system, slowing down the so called "gel time" by 1-2 s in the case of organoclay concentration of 0.05 and 0.1 % and 2-3 s in the case of using these additives in higher concentrations. These minor changes in the technological parameters of the system are the result of a clear effect exerted by the additives on the physics of the foaming process.

The values of the physical-mechanical, morphological and thermophysical characteristics of the obtained composite foams were determined using the methods described above are presented in Tables II and III, and in a graphic form as the Supplementary material to this paper.

| Foom comple | | ω | γ | σ | $\sigma_{\rm rel.}$ | Ε | $E_{\rm rel.}$ |
|-------------|--------------|-------|--------------------|--------|-------------------------------------|---------|-------------------------------------|
| гоаш | sample | wt. % | kg m ⁻³ | kPa | kPa m ⁶ kg ⁻² | kPa | kPa m ⁶ kg ⁻² |
| PUR | Standard | 0.00 | 40.35 | 209.71 | 0.129 | 1974.10 | 1.21 |
| | Cloisite 30B | 0.05 | 40.15 | 229.51 | 0.142 | 1981.01 | 1.23 |
| | | 0.10 | 39.95 | 238.80 | 0.150 | 1990.68 | 1.25 |
| | | 0.20 | 39.81 | 246.51 | 0.156 | 1993.45 | 1.26 |
| | | 0.50 | 39.72 | 245.13 | 0.155 | 1979.59 | 1.25 |
| | | 1.00 | 39.49 | 242.05 | 0.155 | 1976.67 | 1.27 |
| | Cloisite 25A | 0.05 | 40.17 | 224.79 | 0.139 | 1983.54 | 1.23 |
| | | 0.10 | 40.02 | 231.46 | 0.145 | 1981.21 | 1.24 |
| | | 0.20 | 39.79 | 235.02 | 0.148 | 1981.60 | 1.25 |
| | | 0.50 | 39.71 | 234.58 | 0.149 | 1977.65 | 1.25 |
| | | 1.00 | 39.54 | 232.07 | 0.148 | 1992.46 | 1.27 |
| PIR | Standard | 0.00 | 40.32 | 266.09 | 0.164 | 3319.21 | 2.04 |
| | Cloisite 30B | 0.05 | 40.12 | 295.49 | 0.184 | 3328.50 | 2.07 |
| | | 0.10 | 39.92 | 309.30 | 0.194 | 3333.48 | 2.09 |
| | | 0.20 | 39.78 | 320.13 | 0.202 | 3331.49 | 2.11 |
| | | 0.50 | 39.69 | 318.70 | 0.202 | 3350.41 | 2.13 |
| | | 1.00 | 39.51 | 316.05 | 0.202 | 3325.85 | 2.13 |
| | Cloisite 25A | 0.05 | 40.14 | 288.41 | 0.179 | 3374.97 | 2.09 |
| | | 0.10 | 39.99 | 298.31 | 0.187 | 3360.37 | 2.10 |
| | | 0.20 | 39.76 | 303.58 | 0.192 | 3328.50 | 2.11 |
| | | 0.50 | 39.68 | 302.11 | 0.192 | 3339.46 | 2.12 |
| | | 1.00 | 39.61 | 301.00 | 0.192 | 3334.81 | 2.13 |

TABLE II. Physical and mechanical characteristics of PUR and PIR foams

As a result of the introduction of the considered additives into the composition for the synthesis of PIR and PUR foams, there is a slight decrease in the apparent density of the resulting materials. Similar results, confirming the decrease in the foam density with the introduction of small amounts of carbon filler, were obtained by Pikhurov *et al.*³⁵ It is noted that the proven decrease in the apparent density of the materials under consideration can be explained by the increasing degree of stabilization of the three-dimensional polymer network.³⁶

There is gradual increase in the compressive strength index of the obtained composite foams until the mass concentration of the additive is equal to 0.2 %, over that value this index slightly decreases. In the case of Young's modulus, no such trend is observed: this value remains almost unchanged when the fillers considered in this work are introduced, regardless of their concentration. This pattern was previously revealed by Gibson *et al.*,³⁷ who proved that Young's modulus does not change significantly even with a significant decrease or inc-

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rease in the average size of the foam cells (the trends of parameter is described below). On the contrary, the compressive strength index is closely related to the morphology of the cellular structure, increasing with homogeneity of the spatially cross-linked polymer network.

| Foam sample | | ω | $ar{\phi}$ | $A_{\rm u/v}$ | $N_{\rm f} \times 10^{-3}$ | λ | OC | S _m |
|-------------|--------------|-------|------------|---------------|----------------------------|--------------------|------|----------------|
| | | wt. % | μm | 11y/x | cell cm ⁻³ | $mW m^{-1} K^{-1}$ | % | % |
| PUR | Standard | 0.00 | 568.32 | 1.12 | 5.58 | 26.41 | 5.62 | - |
| | Cloisite 30B | 0.05 | 528.54 | 1.04 | 16.13 | 25.58 | 5.97 | |
| | | 0.10 | 500.12 | 1.00 | 20.26 | 25.34 | 6.01 | |
| | | 0.20 | 454.66 | 0.97 | 33.87 | 25.13 | 6.13 | |
| C | | 0.50 | 471.71 | 0.92 | 30.79 | 25.13 | 6.24 | |
| | | 1.00 | 511.49 | 0.88 | 29.46 | 25.19 | 6.41 | |
| | Cloisite 25A | 0.05 | 539.90 | 1.06 | 15.23 | 25.61 | 6.01 | |
| | | 0.10 | 505.80 | 0.99 | 18.47 | 25.40 | 6.13 | |
| | | 0.20 | 471.71 | 0.98 | 32.48 | 25.22 | 6.18 | |
| | | 0.50 | 488.76 | 0.91 | 29.46 | 25.23 | 6.29 | |
| | | 1.00 | 522.85 | 0.86 | 27.44 | 25.27 | 6.52 | |
| PIR | Standard | 0.00 | 681.07 | 1.07 | 3.27 | 24.17 | 6.24 | 57.03 |
| | Cloisite 30B | 0.05 | 627.20 | 1.02 | 9.78 | 23.40 | 6.63 | 48.35 |
| | | 0.10 | 588.72 | 0.95 | 12.36 | 23.17 | 6.68 | 46.22 |
| | | 0.20 | 527.15 | 0.93 | 20.79 | 22.96 | 6.81 | 42.03 |
| | | 0.50 | 550.24 | 0.88 | 18.89 | 22.97 | 6.93 | 42.05 |
| | | 1.00 | 604.11 | 0.85 | 18.06 | 23.03 | 7.12 | 44.78 |
| | Cloisite 25A | 0.05 | 642.59 | 1.03 | 9.25 | 23.41 | 6.68 | 49.35 |
| | | 0.10 | 596.41 | 0.94 | 11.23 | 23.23 | 6.81 | 47.01 |
| | | 0.20 | 550.24 | 0.93 | 19.93 | 23.07 | 6.87 | 44.27 |
| | | 0.50 | 573.32 | 0.88 | 18.06 | 23.06 | 6.99 | 44.61 |
| | | 1.00 | 619.50 | 0.82 | 16.82 | 23.11 | 7.24 | 47.04 |

TABLE III. Morphological and thermophysical characteristics of PUR and PIR foams

The observed difference between the relative strengths of composite foams obtained using Cloisite 30B and Cloisite 25A can be explained by the presence of hydroxyl groups on the surface of Cloisite 30B particles, which contribute to a better distribution of nanosized filler particles over the emerging polymer matrix.

It is known that the strength characteristics of both PIR and PUR foams largely depend on the morphology of the cellular structure of the material.³⁸ Within the framework of this research, the change in the morphological characteristics of the obtained foams was analysed using the method of scanning electron microscopy. SEM microphotographs examples are presented in Fig. 3: the standard (non-filled) polyurethane and polyurethane–polyisocyanurate foams are shown in Fig. 3A and C, respectively, and microphotographs of nanocomposite foam samples (comprising 0.2 % Cloisite 30B) are shown in Fig. 3B and D.

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Fig. 3. Microphotographs of the obtained non-composite and composite PUR (A, B) and PIR (C, D) foams.

It should be noted that the introduction of the additives under consideration significantly affects the uniformity of the structure of the final polymer.

It is important to note that the introduction of a filler significantly affects both the average cell size and their density, contributing to the formation of the cells closest in size. It is assumed that such a modification of the cellular structure of the emerging foam makes a significant contribution to the identified increase in the functional characteristics of the foam.

Based on the information presented in Table III, we failed to identify a clear dependence of the anisotropy coefficient of PIR and PUR foams on the type and concentration of the introduced fillers. Nevertheless, it is known that the introduction of such additives can lead to an increase of the anisotropy coefficient of composite foams,³⁹ however, an increase or decrease in this index has only a slight effect on the performance characteristics of composite foams, which was proved in literature.³⁵

It is important to note that in a number of studies, for example,^{13,35} as a consequence of the introduction of a filler into the system, an increase in the average cell size is noted. Apparently, this trend is associated with an insufficient degree of dispersion of the solid phase during the preparation of the initial suspensions.

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When small amounts of the considered additives are introduced into the composition, a noticeable decrease in their thermal conductivity index is noted. The most noticeable decrease in this characteristic was observed in the case of PIR foams, which may be due to the better distribution of the filler in the polymer matrix. Such an effect is quite natural, since when the filler was dispersed directly in the isocyanate component, the excess of it is shown during the synthesis of PIR foams.

Espadas-Escalante *et* al.⁴⁰ noted an increase in the thermal conductivity of composite foams, probably associated with an insufficient degree of homogeneity of the final suspension used for the synthesis of foams.

Since one of the most important characteristics of PIR foams is combustibility, the effect of fillers introduced into the composition on the stability of the final foams was evaluated by assessing the degree of damage by mass during combustion, $S_{\rm m}$. The combustibility of the analysed PIR foams significantly decreases when small additives of the considered fillers are introduced into the composition.

The content of open cells, *OC*, in the analysed composite foam plastics naturally increases with the content of filler particles. However, there is an extremely low influence of this parameter on the other properties of PIR and PUR foams considered and described above. It is expected that such an increase in the content of open cells will primarily negatively affect the thermal conductivity of the synthesized foams, however, the thermal insulation properties of the obtained foams only increase, which leads to a conclusion that the degree of influence of such a small content of opened cells is negligible.

The relative compressive strength, $\overline{\Phi}$, directly depends on the average cell size of the foams, increasing with a decrease in average cell size of the resulting foamed material. We conclude that in order to improve the strength characteristics of foam plastics, it is extremely advisable to obtain foam plastics with the best morphology of the cellular structure, including the lowest average cell size.

CONCLUSION

Thus, when using small amounts of chemically modified nanoclays (Cloisite 30B, Cloisite 25A), two series of composite polyurethane and polyurethane–polyisocyanurate foams were obtained. The main physical-mechanical, thermophysical and morphological characteristics of the synthesized materials have been studied. It is shown that the introduction of small additives of the considered fillers leads to an improvement in the main performance characteristics of PIR and PUR foams: an increase in relative compressive strength, a decrease in thermal conductivity and a damage of mass during combustion. Based on the data obtained, it is expected that the use of the nanosized fillers described in the

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work will be appropriate in the development of new industrially used heat-insulating foams.

SUPPLEMENTARY MATERIAL

Additional data and information are available electronically at the pages of journal website: https://www.shd-pub.org.rs/index.php/JSCS/article/view/12125, or from the corresponding author on request.

ИЗВОД

УТИЦАЈ НАНОГЛИНЕ НА МЕХАНИЧКА И ТОПЛОТНА СВОЈСТВА КРУТИХ PIR И PUR ПЕНА

RUSLAN R. VLASOV¹, DARIA I. RYABOVA¹, SAKINA Z. ZEYNALOVA², DMITRY V. SOKOLOV¹ и SERGEI A. RYABOV¹

¹Lobachevsky State University – Faculty of Chemistry, Department of Macromolecular Compounds and Colloidal Chemistry, Gagarina av. 23, 603098 Nizhny Novgorod, Russia u ²Tre Tau Engineering srl, Pietro Colletta 85, 10153 Torino, Italy

Проучаван је ефекат додавања малих количина (од 0,05 до 1 мас. %) наночестичних глина на морфологију, физичко-механичке и термофизичке карактеристике крутих полиуретан-полиизоцијануратних (PIR) и полиуретанских (PUR) пена. Процењени су ефекти ових адитива на структуру добијеног материјала, промену његове компресивне чврстоће, Јангов модуо еластичности, губитак масе током сагоревања и топлотну проводљивост. На основу добијених резултата закључено је да додавање малих количина (до 0,2 мас. %) хемијски модификоване Cloisite 30В наноглине ефикасно смањује просечну величину ћелија нанокомпозитних пена, што доводи до побољшања својстава пена.

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