



SUPPLEMENTARY MATERIAL TO  
**Synthesis, spectroscopic characterization and DFT analysis of  
dichlorido( $\eta^6$ -*p*-cymene)ruthenium(II) complexes with  
isonicotinate-polyethylene glycol ester ligands**

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ANALYTICAL AND SPECTRAL DATA OF THE SYNTHESIZED COMPOUNDS

*2-(2-Methoxyethoxy)ethyl iso-nicotinate hydrochloride (L2·HCl)*

Properties: Viscous, hygroscopic oil; soluble in water, methanol, ethanol, *isopropanol*, insoluble in diethyl ether.

<sup>1</sup>H-NMR (400 MHz, D<sub>2</sub>O):  $\delta$  3.33 (s, 3H, OCH<sub>3</sub>), 3.63, 3.74 (m, 4H, CH<sub>3</sub>OCH<sub>2</sub>CH<sub>2</sub>), 3.95 (m, 2H, COOCH<sub>2</sub>CH<sub>2</sub>), 4.63 (m, 2H, COOCH<sub>2</sub>), 8.59 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>3</sup>), 9.03 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>2</sup>). <sup>13</sup>C-NMR (100 MHz, D<sub>2</sub>O):  $\delta$  58.0 (CH<sub>3</sub>O), 66.0 (CH<sub>2</sub>OOC), 68.1, 69.5, 70.9 (CH<sub>2</sub>O), 126.9 (C<sup>3</sup>), 142.6 (C<sup>2</sup>), 145.9 (C<sup>4</sup>), 163.2 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>11</sub>H<sub>16</sub>NO<sub>4</sub>]<sup>+</sup> 226.10738, *m/z* 226.10738 [M+H]<sup>+</sup>. IR:  $\nu$  (cm<sup>-1</sup>) 3068(w), 2877(w), 2400(b), 2100(w), 2022(w), 1916(w), 1731(s), 1635(w), 1605(m), 1501(w), 1450(w), 1282(s), 1239(m), 1198(w), 1100(s), 998(m), 845(m), 754(s), 684(m), 359(w). Yield: 88%.

*2-[2-(2-Methoxyethoxy)ethoxy]ethyl iso-nicotinate hydrochloride (L3·HCl)*

Properties: Viscous, hygroscopic oil; soluble in water, methanol, ethanol, *isopropanol*, chloroform, and dichloromethane, insoluble in diethyl ether.

<sup>1</sup>H-NMR (400 MHz, D<sub>2</sub>O):  $\delta$  3.35 (s, 3H, OCH<sub>3</sub>), 3.60, 3.70, 3.72, 3.79 (m, 8H, OCH<sub>2</sub>CH<sub>2</sub>O), 3.98 (m, 2H, COOCH<sub>2</sub>CH<sub>2</sub>), 4.66 (m, 2H, COOCH<sub>2</sub>), 8.62 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>3</sup>), 9.04 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>2</sup>). <sup>13</sup>C-NMR (100 MHz, D<sub>2</sub>O):  $\delta$  58.0 (CH<sub>3</sub>O), 66.1 (CH<sub>2</sub>OOC), 68.2, 69.4, 69.5, 69.7, 70.9 (CH<sub>2</sub>O), 127.0 (C<sup>3</sup>), 142.6 (C<sup>2</sup>), 145.9 (C<sup>4</sup>), 163.3 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>13</sub>H<sub>20</sub>NO<sub>5</sub>]<sup>+</sup> 270.13360, *m/z* 270.13351 [M+H]<sup>+</sup>. IR:  $\nu$  (cm<sup>-1</sup>) 3073(w), 2875(w), 2400(b), 2100(w), 2026(w), 1916(w), 1730(s), 1635(w), 1607(m), 1500(w), 1450(w), 1282(s), 1239(m), 1198(w), 1098(s), 998(m), 845(m), 758(s), 685(m), 363(w). Yield: 95%.

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*2-(2-Hydroxyethoxy)ethyl isonicotinate hydrochloride (L4-HCl)*

Properties: Viscous, colourless oil; soluble in water, methanol, ethanol, isopropanol, acetone, chloroform, and dichloromethane, insoluble in diethyl ether.

<sup>1</sup>H-NMR (400 MHz, D<sub>2</sub>O): δ 3.77 (m, 4H, HOCH<sub>2</sub>CH<sub>2</sub>), 4.00 (m, 2H, COOCH<sub>2</sub>CH<sub>2</sub>), 4.69 (m, 2H, COOCH<sub>2</sub>), 8.60 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>3</sup>), 9.04 (d, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 2H, H<sup>2</sup>). <sup>13</sup>C-NMR (100 MHz, D<sub>2</sub>O): δ 60.3 (CH<sub>2</sub>OH), 66.1 (CH<sub>2</sub>OOC), 68.1 (CH<sub>2</sub>CH<sub>2</sub>OOC), 71.8 (CH<sub>2</sub>CH<sub>2</sub>OH), 126.9 (C<sup>3</sup>), 142.7 (C<sup>2</sup>), 145.9 (C<sup>4</sup>), 163.4 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>10</sub>H<sub>14</sub>NO<sub>4</sub>]<sup>+</sup> 212.09173, *m/z* 212.09176 [M+H]<sup>+</sup>. IR: ν (cm<sup>-1</sup>) 3350(b), 3080(w), 2872(w), 2440(b), 2100(w), 1730(s), 1637(w), 1604(w), 1502(w), 1451(w), 1404(w), 1282(s), 1119(s), 1068(m), 1002(m), 925(w), 889(w), 839(m), 754(s), 678(m), 520(w), 357(w), 255(w).

*2-[2-(2-Methylamino-benzoyloxy)ethoxy]ethyl iso-nicotinate (L5)*

Properties: Viscous oil, which crystallized upon standing, soluble in methanol, ethanol, isopropanol, chloroform, and dichloromethane, and insoluble in diethyl ether and water.

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>): δ 2.89 (s, 3H, NCH<sub>3</sub>), 3.87 (m, 4H, OCH<sub>2</sub>), 4.43, 4.53 (t, <sup>3</sup>J<sub>H,H</sub> = 6.8 Hz, 4H, COOCH<sub>2</sub>), 6.53 (t, <sup>3</sup>J<sub>H,H</sub> = 7.6 Hz, 1H, H<sup>9</sup>), 6.65 (d, <sup>3</sup>J<sub>H,H</sub> = 8.5 Hz, 1H, H<sup>7</sup>), 7.37 (ddd, <sup>3</sup>J<sub>H<sub>8</sub>,H<sub>7</sub></sub> = 7.8 Hz, <sup>3</sup>J<sub>H<sub>8</sub>,H<sub>9</sub></sub> = 7.1 Hz, <sup>4</sup>J<sub>H<sub>8</sub>,H<sub>10</sub></sub> = 1.7 Hz, 1H, H<sup>8</sup>), 7.84 (d, <sup>3</sup>J<sub>H,H</sub> = 6.0 Hz, 2H, H<sup>3</sup>), 7.88 (dd, <sup>3</sup>J<sub>H<sub>10</sub>,H<sub>9</sub></sub> = 8.0 Hz, <sup>4</sup>J<sub>H<sub>10</sub>,H<sub>8</sub></sub> = 1.6 Hz, 1H, H<sup>10</sup>), 8.71 (d, <sup>3</sup>J<sub>H,H</sub> = 6.1 Hz, 2H, H<sup>2</sup>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>): δ 29.7 (NCH<sub>3</sub>), 63.2, 65.0 (CH<sub>2</sub>OOC), 69.0, 69.4 (CH<sub>2</sub>CH<sub>2</sub>OOC), 109.7 (C<sup>5</sup>), 110.9 (C<sup>7</sup>), 114.5 (C<sup>9</sup>), 123.4 (C<sup>3</sup>), 131.7 (C<sup>10</sup>), 135.0 (C<sup>8</sup>), 138.1 (C<sup>4</sup>), 149.8 (C<sup>2</sup>), 152.2 (C<sup>6</sup>), 164.9, 168.5 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>18</sub>H<sub>21</sub>N<sub>2</sub>O<sub>5</sub>]<sup>+</sup> 345.14450, *m/z* 345.14451 [M+H]<sup>+</sup>. IR: ν (cm<sup>-1</sup>) 3372(m), 2950(w), 2885(w), 2825(w), 1718(s), 1684(s), 1603(w), 1580(m), 1518(m), 1454(w), 1431(m), 1408(m), 1375(m), 1322(m), 1278(m), 1261(s), 1243(s), 1180(s), 1134(s), 1117(s), 1093(s), 1063(m), 1040(m), 1021(m), 952(m), 894(w), 877(w), 867(w), 850(w), 753(s), 700(s), 675(m), 605(m), 575(m), 533(m), 512(m), 275(m), 217 (w). UV-vis (1.74 × 10<sup>-4</sup> M, CHCl<sub>3</sub>, 298 K): λ<sub>max</sub> 358 nm (ε = 5747 M<sup>-1</sup> cm<sup>-1</sup>).

*Dichlorido(η<sup>6</sup>-p-cymene)(diethylene glycol monomethyl ether isonicotinate-κN)ruthenium(II) (2)*

Properties: Orange powder; soluble in chloroform, dichloromethane, dimethylformamide, acetone, and acetonitrile, moderately soluble in methanol, ethanol, isopropanol, and tetrahydrofuran, insoluble in diethyl ether and toluene.

EA: Anal. Found: C, 47.00; H, 5.18; N, 2.65. Calcd for C<sub>21</sub>H<sub>29</sub>Cl<sub>2</sub>NO<sub>4</sub>Ru (531.43): C, 47.46; H, 5.50; N, 2.64. <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>): δ 1.31 (d, <sup>3</sup>J<sub>H,H</sub> = 7.0 Hz, 6H, CH(CH<sub>3</sub>)<sub>2</sub>), 2.09 (s, 3H, CCH<sub>3</sub>), 2.98 (sept, <sup>3</sup>J<sub>H,H</sub> = 7.0 Hz, 1H, CH(CH<sub>3</sub>)<sub>2</sub>), 3.38 (s, 3H, OCH<sub>3</sub>), 3.56, 3.68 (m, 4H, OCH<sub>2</sub>), 3.83 (t, <sup>3</sup>J<sub>H,H</sub> = 4.7 Hz, 2H, COOCH<sub>2</sub>CH<sub>2</sub>), 4.53 (t, <sup>3</sup>J<sub>H,H</sub> = 4.6 Hz, 2H, COOCH<sub>2</sub>), 5.23 (d, <sup>3</sup>J<sub>H,H</sub> = 5.9 Hz, 2H, CHCCH<sub>3</sub>), 5.45 (d, <sup>3</sup>J<sub>H,H</sub> = 5.8 Hz, 2H, CHCHCCH<sub>3</sub>), 7.85 (d, <sup>3</sup>J<sub>H,H</sub> = 6.5 Hz, 2H, H<sup>3</sup>), 9.23 (d, <sup>3</sup>J<sub>H,H</sub> = 6.5 Hz, 2H, H<sup>2</sup>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>): δ 18.4 (CCH<sub>3</sub>), 22.4 (C(CH<sub>3</sub>)<sub>2</sub>), 30.8 (C(CH<sub>3</sub>)<sub>2</sub>), 59.2 (CH<sub>3</sub>O), 65.1 (CH<sub>2</sub>OOC), 70.3 (CH<sub>2</sub>O), 82.5, 83.3 (CH), 97.5, 103.9 (CCH), 123.7 (C<sup>3</sup>), 138.7 (C<sup>4</sup>), 155.9 (C<sup>2</sup>), 163.9 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>21</sub>H<sub>29</sub>ClNO<sub>4</sub><sup>96</sup>Ru]<sup>+</sup> 490.08556, *m/z* 490.08544 [M-Cl]<sup>+</sup>. IR: ν (cm<sup>-1</sup>) 3062(w), 2964(w), 2878(w), 1728(s), 1556(w), 1468(w), 1451(w), 1414(m), 1377(w), 1323(w), 1275(s), 1228(w), 1202(w), 1107(s), 1056(m), 1027(m), 863(m), 769(m), 695(m), 659(w), 451(w), 371(w), 284(s), 232(s).

*Dichlorido( $\eta^6$ -p-cymene)(triethylene glycol monomethyl ether iso-nicotinate- $\kappa$ N)ruthenium(II) (3)*

Properties: Orange powder; soluble in chloroform, dichloromethane, dimethylformamide, acetone, and acetonitrile, moderately soluble in methanol, ethanol, isopropanol, and tetrahydrofuran, insoluble in diethyl ether and toluene.

EA: Anal. Found: C, 47.58; H, 5.31; N, 2.47. Calcd for  $C_{23}H_{33}Cl_2NO_5Ru$  (575.49): C, 48.00; H, 5.78; N, 2.43.  $^1H$ -NMR (400 MHz,  $CDCl_3$ ):  $\delta$  1.31 (d,  $^3J_{H,H} = 6.9$  Hz, 6H,  $CH(CH_3)_2$ ), 2.09 (s, 3H,  $CCH_3$ ), 2.98 (sept,  $^3J_{H,H} = 6.9$  Hz, 1H,  $CH(CH_3)_2$ ), 3.37 (s, 3H,  $OCH_3$ ), 3.54, 3.64, 3.67, 3.68 (m, 8H,  $OCH_2$ ), 3.83 (t,  $^3J_{H,H} = 4.6$  Hz, 2H,  $COOCH_2CH_2$ ), 4.52 (t,  $^3J_{H,H} = 4.6$  Hz, 2H,  $COOCH_2$ ), 5.23 (d,  $^3J_{H,H} = 5.9$  Hz, 2H,  $CHCCH_3$ ), 5.45 (d,  $^3J_{H,H} = 5.9$  Hz, 2H,  $CHCHCCH_3$ ), 7.85 (d,  $^3J_{H,H} = 6.5$  Hz, 2H,  $H^3$ ), 9.22 (d,  $^3J_{H,H} = 6.5$  Hz, 2H,  $H^2$ ).  $^{13}C$ -NMR (100 MHz,  $CDCl_3$ ):  $\delta$  18.4 ( $CCH_3$ ), 22.4 ( $C(CH_3)_2$ ), 30.8 ( $C(CH_3)_2$ ), 59.2 ( $CH_3O$ ), 65.5 ( $CH_2OOC$ ), 69.0, 70.7, 70.8, 70.8, 72.1 ( $CH_2O$ ), 82.5, 83.2 (CH), 97.5, 103.9 (CCH), 123.7 ( $C^3$ ), 138.8 ( $C^4$ ), 155.9 ( $C^2$ ), 163.9 (COO). ESI-HRMS ( $CH_3OH$ ), positive mode: Calcd for  $[C_{23}H_{33}ClNO_5^{96}Ru]^+$  534.11178,  $m/z$  534.11174  $[M-Cl]^+$ . IR:  $\nu$  ( $cm^{-1}$ ) 3063(w), 2964(w), 2876(w), 1732(s), 1556(w), 1475(w), 1451(w), 1414(m), 1367(w), 1319(w), 1279(s), 1202(w), 1144(m), 1104(s), 1090(s), 1053(m), 1028(m), 947(w), 867(m), 849(m), 802(w), 769(m), 695(m), 659(w), 518(w), 455(w), 371(w), 283(s), 232(s).

*Dichlorido( $\eta^6$ -p-cymene)((3-oxa-5-hydroxy)-pentyl isonicotinate- $\kappa$ N)ruthenium(II) (4)*

Properties: Orange powder; soluble in chloroform, dichloromethane, dimethylformamide, acetone, and acetonitrile, moderately soluble in methanol, ethanol, isopropanol, and tetrahydrofuran, insoluble in diethyl ether and toluene.

EA: Anal. Found: C, 45.98; H, 5.16; N, 2.70. Calcd for  $C_{20}H_{27}Cl_2NO_4Ru$  (518.21): C, 46.36; H, 5.25; N, 2.70.  $^1H$ -NMR (400 MHz,  $CDCl_3$ ):  $\delta$  1.32 (d,  $^3J_{H,H} = 6.9$  Hz, 6H,  $CH(CH_3)_2$ ), 2.10 (s, 3H,  $CCH_3$ ), 2.99 (sept,  $^3J_{H,H} = 6.9$  Hz, 1H,  $CH(CH_3)_2$ ), 3.64, 3.76, 3.85 (m, 6H,  $OCH_2$ ), 4.54 (m, 2H,  $COOCH_2$ ), 5.24 (d,  $^3J_{H,H} = 5.9$  Hz, 2H,  $CHCCH_3$ ), 5.46 (d,  $^3J_{H,H} = 5.8$  Hz, 2H,  $CHCHCCH_3$ ), 7.85 (d,  $^3J_{H,H} = 6.4$  Hz, 2H,  $H^3$ ), 9.24 (d,  $^3J_{H,H} = 6.4$  Hz, 2H,  $H^2$ ).  $^{13}C$ -NMR (100 MHz,  $CDCl_3$ ):  $\delta$  18.4 ( $CCH_3$ ), 22.4 ( $C(CH_3)_2$ ), 30.8 ( $C(CH_3)_2$ ), 61.9, 65.4, 68.9, 72.6 ( $CH_2O$ ), 82.5, 83.2 (CH), 97.6, 104.0 (CCH), 123.7 ( $C^3$ ), 138.7 ( $C^4$ ), 156.0 ( $C^2$ ), 164.0 (COO). ESI-HRMS ( $CH_3OH$ ), positive mode: Calcd for  $[C_{20}H_{27}ClNO_4^{96}Ru]^+$  476.06991,  $m/z$  476.06973  $[M-Cl]^+$ . IR:  $\nu$  ( $cm^{-1}$ ) 3450(b), 3063(w), 2962(w), 2878(w), 1728(s), 1471(w), 1414(m), 1382(w), 1323(w), 1278(s), 1228(m), 1120(s), 1061(s), 932(w), 866(m), 800(w), 771(s), 698(m), 659(w), 451(w), 400(w), 371(w), 355(w), 286(s), 233(s).

*Dichlorido( $\eta^6$ -p-cymene)[2-{2-(2-methylamino-benzoyloxy)-ethoxy}-ethyl isonicotinate- $\kappa$ N]ruthenium(II) (5)*

Properties: Yellow powder; soluble in chloroform, dichloromethane, moderately soluble in methanol, ethanol, isopropanol, insoluble in diethyl ether.

EA: Anal. Found: C, 51.83; H, 4.99; N, 4.09. Calcd for  $C_{28}H_{34}Cl_2N_2O_5Ru$  (650.56): C, 51.70; H, 5.27; N, 4.31.  $^1H$ -NMR (400 MHz,  $CDCl_3$ ):  $\delta$  1.30 (d,  $^3J_{H,H} = 6.9$  Hz, 6H,  $CH(CH_3)_2$ ), 2.07 (s, 3H,  $CCH_3$ ), 2.90 (s, 3H,  $NCH_3$ ), 2.97 (sept,  $^3J_{H,H} = 6.9$  Hz, 1H,  $CH(CH_3)_2$ ), 3.86 (m, 4H,  $OCH_2$ ), 4.43, 4.52 (m, 4H,  $COOCH_2CH_2$ ), 5.21 (d,  $^3J_{H,H} = 5.5$  Hz, 2H,  $CHCCH_3$ ), 5.44 (d,  $^3J_{H,H} = 5.5$  Hz, 2H,  $CHCHCCH_3$ ), 6.59 (t,  $^3J_{H,H} = 7.5$  Hz, 1H,  $H^9$ ), 6.69 (d,  $^3J_{H,H} = 8.3$  Hz, 1H,  $H^7$ ), 7.40 (t,  $^3J_{H,H} = 7.8$  Hz, 1H,  $H^8$ ), 7.79 (d,  $^3J_{H,H} = 5.8$  Hz, 2H,  $H^3$ ), 7.90 (d,  $^3J_{H,H} = 8.0$  Hz, 1H,  $H^{10}$ ), 9.17 (d,  $^3J_{H,H} = 5.8$  Hz, 2H,  $H^2$ ).  $^{13}C$ -NMR (100 MHz,  $CDCl_3$ ):  $\delta$  18.3 ( $CCH_3$ ), 22.4 ( $C(CH_3)_2$ ), 29.8 ((b),  $NCH_3$ ), 30.8 ( $C(CH_3)_2$ ), 63.2, 65.3 ( $CH_2OOC$ ), 68.9, 69.4 ( $CH_2OCH_2$ ), 82.5, 83.2 ( $CH_{cym}$ ), 97.5, 103.9 (CCH<sub>cym</sub>), 109.8 (b),

$C^5$ ), 111.2 ((b),  $C^7$ ), 114.7 ((b),  $C^9$ ), 123.6 ( $C^3$ ), 131.6 ( $C^{10}$ ), 135.0 ( $C^8$ ), 138.6 ( $C^4$ ), 152.2 ((b),  $C^6$ ), 155.9 ( $C^2$ ), 163.9, 168.4 (COO). ESI-HRMS (CH<sub>3</sub>OH), positive mode: Calcd for [C<sub>28</sub>H<sub>34</sub>ClN<sub>2</sub>O<sub>5</sub><sup>96</sup>Ru]<sup>+</sup> 609.12267, *m/z* 609.12286 [M-Cl]<sup>+</sup>. IR:  $\nu$  (cm<sup>-1</sup>) 3372(w), 3059(w), 2964(w), 2872(w), 2819(w), 1734(s), 1676(s), 1604(w), 1576(m), 1516(m), 1442(w), 1413(m), 1378(w), 1322(w), 1278(s), 1236(s), 1120(s), 1084(s), 1057(m), 948(w), 858(m), 762(s), 702(s), 662(w), 528(m), 283(s), 234(s). UV-vis (7.01×10<sup>-5</sup> M, CHCl<sub>3</sub>, 298 K):  $\lambda_{\max}$  351 nm ( $\epsilon = 11689 \text{ M}\times\text{cm}^{-1}$ ). Fluorescence (7.01×10<sup>-5</sup> M, CHCl<sub>3</sub>, 298 K):  $\lambda_{\max, \text{excitation}}$  367 nm,  $\lambda_{\max, \text{emission}}$  418 nm.

**Table S-I.** Second order perturbation theory parameters and occupancies for the selected interactions within investigated ligands and complexes

Donor (i)	L1	
	Acceptor (j)	E2/ kJ mol <sup>-1</sup>
$\pi(\text{C-C})$	$\pi^*(\text{C-C})$	76
$\pi(\text{C-C})$	$\pi^*(\text{C-C})$	95
$\pi(\text{C-N})$	$\pi^*(\text{C-C})$	60
$\pi(\text{C-C})$	$\pi^*(\text{C-N})$	109
LP(N)	$\pi^*(\text{C-C})$	41
LP(O)	$\pi^*(\text{C-C})$	75
LP(O)	$\pi^*(\text{C-O})$	135
	1	
$\pi(\text{C-C})$	$\pi^*(\text{C-C})$	46
$\pi(\text{C-C})$	$\pi^*(\text{C-C})$	55
$\pi(\text{C-C})$	LP*(Ru)	60
$\pi(\text{C-C})$	LP*(Ru)	80
LP(N)	$\sigma^*(\text{Ru-Cl})$	27
$\pi(\text{C-N})$	LP*(Ru)	50
LP(Cl)	LP*(Ru)	34
LP(Cl)	LP*(Ru)	75
LP(Cl)	$\sigma(\text{Ru-Cl})$	5
	L4	
LP(O)	$\sigma(\text{C-C})$	4.8
	L5	
LP(O)	$\sigma(\text{C-C})$	71
LP(O)	$\sigma(\text{C-O})$	132
LP(O)	$\pi^*(\text{C-O})$	105

TABLE S-II. Experimental and theoretical  $^{13}\text{C}$ -NMR chemical shifts (in ppm) of **L1** and **1**

Carbon atom	Ligand <b>L1</b> · <b>HCl</b>		Complex <b>1</b>	
	Experimental	Theoretical	Experimental	Theoretical
C <sup>2</sup>	142.7	140.8	155.9	148.8
C <sup>3</sup>	126.9	127.5	123.7	118.0
C <sup>4</sup>	145.7	145.0	138.8	136.2
CH <sub>3</sub> O	58.2	62.8	59.2	58.0
CH <sub>2</sub> OOC	65.9	73.9	65.5	67.8
CH <sub>2</sub> O	69.6	73.7	69.0	69.7
COO	163.4	158.6	163.9	159.2
CCH <sub>3</sub>			18.4	18.9
C(CH <sub>3</sub> ) <sub>2</sub>			22.4	22.7
C(CH <sub>3</sub> ) <sub>2</sub>			30.8	35.5
C <sub>arm</sub>			70.8	83.2
C <sub>arm</sub>			72.1	83.6
C <sub>arm</sub>			82.5	84.5
C <sub>arm</sub>			83.3	86.9
CH			97.5	99.9
CCH			103.9	111.4
	R	0.99	R	0.98
	MAE [ppm]	3.5	MAE [ppm]	4.6

Table S-III. Experimental and theoretical  $^1\text{H}$  NMR chemical shifts (in ppm) of **L1**·**HCl** and **1**

Hydrogen atom	Ligand <b>L1</b> · <b>HCl</b>		Complex <b>1</b>	
	Experimental	Theoretical	Experimental	Theoretical
OCH <sub>3</sub>	3.48	3.28	3.42	3.39
OCH <sub>2</sub>	3.92	3.60	3.72	3.55
COOCH <sub>2</sub>	4.67	4.31	4.52	4.17
C <sup>3</sup> -H	8.62	8.61	7.86	7.62
C <sup>2</sup> -H	9.05	8.61	9.23	8.64
CH(CH <sub>3</sub> ) <sub>2</sub>			1.31	1.31
CCH <sub>3</sub>			2.09	1.67
CH(CH <sub>3</sub> ) <sub>2</sub>			2.98	3.00
C <sub>arm</sub> -H			5.23	4.69
C <sub>arm</sub> -H			5.45	4.95
	R	0.97	R	0.99
	MAE [ppm]	0.27	MAE [ppm]	0.44

**Table S-IV.** Experimental and theoretical  $^{13}\text{C}$  NMR chemical shifts (in ppm) of **L4·HCl** and **4**

Carbon atom	<b>L4·HCl</b>		<b>4</b>	
	Experimental	Theoretical	Experimental	Theoretical
C <sup>2</sup>	142.7	155.2	156.0	150.9
C <sup>3</sup>	126.9	127.5	138.7	119.5
C <sup>4</sup>	145.9	141.1	123.7	134.3
CH <sub>2</sub> CH <sub>2</sub> OH	71.8	67.4	72.6	73.4
CH <sub>2</sub> CH <sub>2</sub> OOC	68.1	63.8	65.4	66.7
CH <sub>2</sub> OOC	66.1	62.9	68.9	68.5
CH <sub>2</sub> OH	60.3	59.2	61.9	60.4
COO	163.4	161.7	164.0	161.4
CCH <sub>3</sub>			18.4	18.8
C(CH <sub>3</sub> ) <sub>2</sub>			22.4	27.2
C(CH <sub>3</sub> ) <sub>2</sub>			30.8	35.9
C <sub>arm</sub>			82.5	84.4
C <sub>arm</sub>			82.5	85.34
C <sub>arm</sub>			83.2	85.9
C <sub>arm</sub>			83.2	87.0
CH			97.6	100.6
CCH			104.0	114.1
	R	0.98	R	0.98
	MAE [ppm]	4.1	MAE [ppm]	4.5

**Table S-V.** Experimental and theoretical  $^1\text{H}$  NMR chemical shifts (in ppm) of **L4·HCl** and **4**

Hydrogen atom	<b>L4·HCl</b>		<b>4</b>	
	Experimental	Theoretical	Experimental	Theoretical
OCH <sub>2</sub>	3.77	3.71	3.85	3.96
HOCH <sub>2</sub> CH <sub>2</sub>	3.77	3.55	3.64	3.93
COOCH <sub>2</sub> CH <sub>2</sub>	4.00	3.74	3.76	3.87
COOCH <sub>2</sub>	4.69	44.26	4.54	4.45
C <sup>3</sup> -H	8.60	8.64	9.24	9.13
C <sup>2</sup> -H	9.04	9.45	7.85	8.09
CH(CH <sub>3</sub> ) <sub>2</sub>			1.32	1.38
CCH <sub>3</sub>			2.10	1.81
CH(CH <sub>3</sub> ) <sub>2</sub>			2.99	3.21
C <sub>arm</sub> -H			5.24	5.01
C <sub>arm</sub> -H			5.46	5.21
	R	0.99	R	0.99
	MAE [ppm]	0.24	MAE [ppm]	0.18

**Table S-VI.** Experimental and theoretical  $^{13}\text{C}$  NMR chemical shifts (in ppm) of **L5** and **5**

Carbon atom	<b>L5</b>		<b>5</b>	
	Experimental	Theoretical	Experimental	Theoretical
C <sup>2</sup>	149.8	148.5	155.9	151.1
C <sup>3</sup>	123.4	123.4	123.6	119.7
C <sup>4</sup>	138.1	136.2	138.6	134.4
C <sup>5</sup>	109.7	127.6	109.8	128.7
C <sup>6</sup>	152.2	150.9	152.2	126.0
C <sup>7</sup>	110.9	110.3	111.2	125.1
C <sup>8</sup>	135.0	135.7	135.0	127.7
C <sup>9</sup>	114.5	130.8	114.7	123.9
C <sup>10</sup>	131.7	129.6	131.6	122.7
NCH <sub>3</sub>	29.7	29.1	29.8	29.7
CH <sub>2</sub> CH <sub>2</sub> OOC	69.4	67.2	69.4	68.6
CH <sub>2</sub> OOC	65.0	64.0	65.3	67.3
CH <sub>2</sub> CH <sub>2</sub> OOC	69.0	62.8	68.9	72.3
CH <sub>2</sub> OOC	63.2	61.1	63.2	69.3
COO	168.5	164.2	168.4	169.4
COO	164.9	156.2	163.9	161.7
CCH <sub>3</sub>			18.3	19.1
C(CH <sub>3</sub> ) <sub>2</sub>			22.4	22.8
C(CH <sub>3</sub> ) <sub>2</sub>			30.8	36.3
C <sub>am</sub>			82.5	85.2
C <sub>am</sub>			82.5	85.2
C <sub>am</sub>			83.2	85.8
C <sub>am</sub>			83.2	88.3
CH			97.5	114.3
CCH			103.9	102.5
	R	0.96	R	0.95
	MAE [ppm]	4.2	MAE [ppm]	6.0

**Table S-VII.** Experimental and theoretical  $^1\text{H}$  NMR chemical shifts (in ppm) of **L5** and **5**

Hydrogen atom	<b>L5</b>		<b>5</b>	
	Experimental	Theoretical	Experimental	Theoretical
$\text{NCH}_3$	2.89	2.53	2.90	2.69
$\text{OCH}_2$	3.87	3.88	3.86	2.85
$\text{COOCH}_2\text{CH}_2$	4.43	3.44	4.43	3.27
$\text{COOCH}_2\text{CH}_2$	4.53	3.51	4.52	3.40
$\text{C}^3\text{-H}$	7.84	8.23	7.79	7.07
$\text{C}^2\text{-H}$	8.71	9.05	9.17	8.16
$\text{C}^7\text{-H}$	6.65	6.81	6.69	6.74
$\text{C}^8\text{-H}$	7.37	7.68	7.40	6.89
$\text{C}^9\text{-H}$	6.53	6.81	6.59	6.74
$\text{C}^{10}\text{-H}$	7.88	7.73	7.90	6.74
$\text{CH}(\text{CH}_3)_2$			1.30	0.30
$\text{CCH}_3$			2.07	0.79
$\text{CH}(\text{CH}_3)_2$			2.97	2.09
	R	0.95	R	0.94
	MAE [ppm]	0.40	MAE [ppm]	0.79