



SUPPLEMENTARY MATERIAL TO
**Structure and DNA/BSA binding study of zinc(II) complex with
4-ethynyl-2,2'-bipyridine**

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ANALYTICAL DATA FOR EBPY LIGAND AND THE COMPLEX

Anal. Calcd. for $C_{12}H_8N_2$ (MW 180.21): C, 79.98; H, 4.47; N, 15.55 %. Found: C, 79.82; H, 4.52; N, 15.68 %. 1H NMR (200 MHz, DMSO- d_6): δ = 4.69 (*s*, 1H, CCH), 7.52 (*m*, 2H, C5H and C5'H), 7.98 (*dd*, *J* = 8.6, 6.9 Hz, 1H, C4'H), 8.39 (*d*, *J* = 6.9 Hz, 2H, C3H and C3'H), 8.72 (*d*, *J* = 5.0 Hz, C6H and C6'H) ppm. IR (KBr, ν , cm $^{-1}$): 3224m (ν (Calkyne-H)), 3059w, 3029w, 3011w (ν (C_{ar}-H)), 2108m (ν (C≡C)), 1584vs, 1564m, 1540s, 1457vs (ν (C_{ar}=C_{ar}) and ν (C_{ar}=N)), 794s (γ (C_{ar}-H)), 626m (γ (Calkyne-H)). UV-Vis (DMSO, λ_{max} , nm): 283 (ε = 1.0 \times 10 4 M $^{-1}$ cm $^{-1}$), 308 (*sh*, ε = 5.3 \times 10 3 M $^{-1}$ cm $^{-1}$). A_M (DMSO): 1.50 $\Omega^{-1}\text{cm}^2\text{mol}^{-1}$.

Anal. Calcd. for $C_{12}H_8Cl_2N_2Zn$ (MW 316.47): C, 45.54; H, 2.55; N, 8.85 %. Found: C, 45.41; H, 2.52; N, 8.92 %. 1H NMR (200 MHz, DMSO- d_6): δ = 4.84 (*s*, 1H, CCH), 7.67 (*m*, 2H, C5H and C5'H), 8.12 (*t*, *J* = 7.4 Hz, 1H, C4'H), 8.57 (*s*, 2H, C3H and C3'H), 8.72 (*d*, *J* = 4.8 Hz, C6H and C6'H) ppm. IR (KBr, ν , cm $^{-1}$): 3215s (ν (Calkyne-H)), 3114w, 3080w, 3057w (ν (C_{ar}-H)), 2109m (ν (C≡C)), 1604vs, 1571w, 1542m, 1480m, 1440s, 1406s (ν (C_{ar}=C_{ar}) and ν (C_{ar}=N)), 792s (γ (C_{ar}-H)), 623m (γ (Calkyne-H)). UV-Vis (DMSO, λ_{max} , nm): 283 (ε = 9.3 \times 10 3 M $^{-1}$ cm $^{-1}$), 308 (*sh*, ε = 4.8 \times 10 3 M $^{-1}$ cm $^{-1}$). A_M (DMSO): 4.82 $\Omega^{-1}\text{cm}^2\text{mol}^{-1}$.

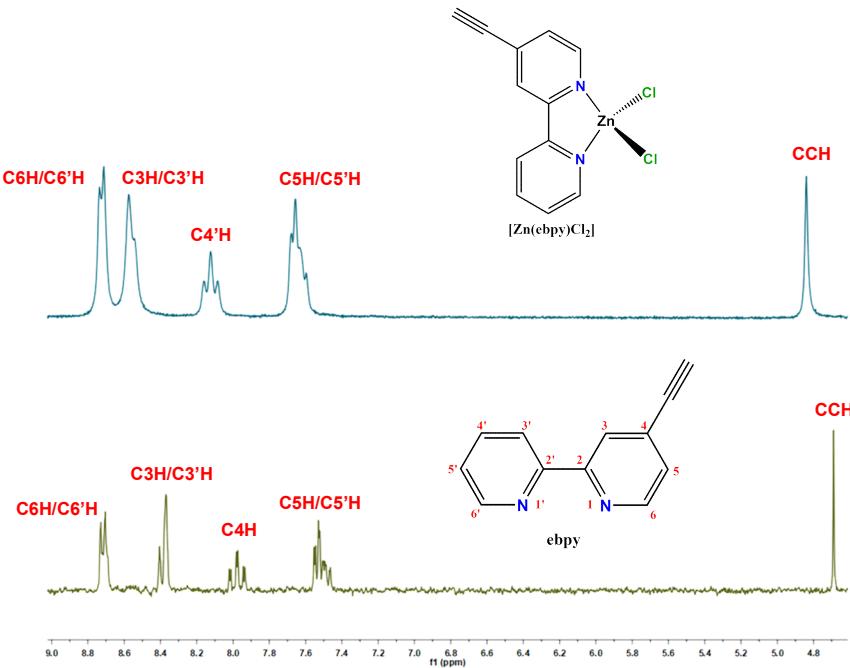


Fig. S-1. The parts of 1H NMR spectrum of $[Zn(\text{ebpy})\text{Cl}_2]$ measured in DMSO- d_6 in comparison to that of ebpy.

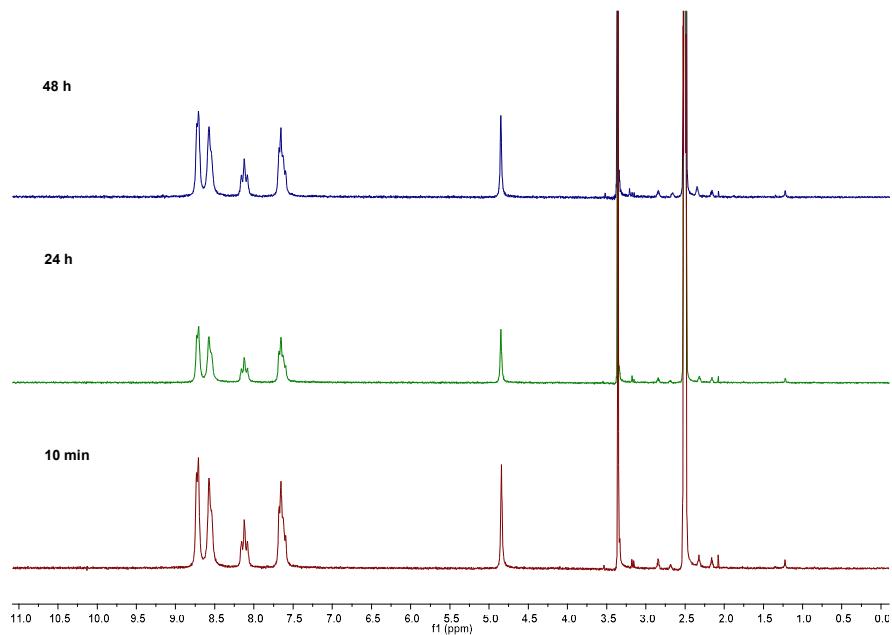


Fig. S-2. Time-dependent ¹H NMR spectra of $[\text{Zn}(\text{ebpy})\text{Cl}_2]$ measured in $\text{DMSO}-d_6$ at ambient temperature over 48 h.

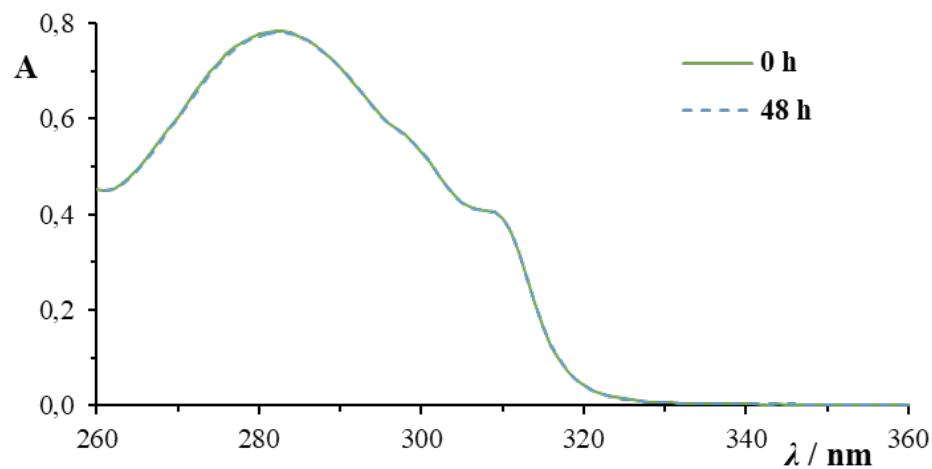


Fig. S-3. UV-Vis spectra of $[\text{Zn}(\text{ebpy})\text{Cl}_2]$ measured in DMSO right after its dissolution and after 48 h standing at ambient temperature.

TABLE S-I. Details of the crystal structure determination of $[\text{Zn}(\text{ebpy})\text{Cl}_2]$

	[$\text{Zn}(\text{ebpy})\text{Cl}_2$]
Empirical formula	$\text{C}_{12}\text{H}_8\text{Cl}_2\text{N}_2\text{Zn}$
Molecular weight	316.47
Crystal system, space group	monoclinic, $P2_1/m$
a (Å)	7.9738(7)
b (Å)	7.2991(8)
c (Å)	11.0462(10)
β (°)	98.749(7)
V (Å ³)	635.43(11)
F_{000}	316
Z	2
X-radiation, λ / Å	Mo- K_α 0.71073
data collect. temperat. / K	250(2)
Calculated density (g cm ⁻³)	1.654
Absorption coefficient (mm ⁻¹)	2.329
Crystal size (mm ³)	0.19 × 0.143 × 0.08
2θ range (°)	3.73 to 52.644
index ranges h, k, l	-9 ... 9, -9 ... 9, -13 ... 13
No. of collected and independent reflections	5319, 1363
R_{int}	0.0361
Data / restraints / parameters	1363 / 0 / 100
Goodness-on-fit on F^2	1.079
Final R indices [$I \geq 2\sigma(I)$]	0.0280, 0.0660
Final R indices (all data)	0.0332, 0.0702
Difference density: max, min (e Å ⁻³)	0.27, -0.33
CCDC number	2266337

TABLE S-II. Selected bond distances (Å) and valence angles (°) in $[\text{Zn}(\text{ebpy})\text{Cl}_2]$

[$\text{Zn}(\text{ebpy})\text{Cl}_2$]
Zn1—N1 2.045(3)
Zn1—N2 2.048(3)
Zn1—Cl1 2.1981(7)
Cl1 ⁱ —Zn1—Cl1 116.77(4)
N1—Zn1—Cl1 ⁱ 113.93(3)
N1—Zn1—Cl1 113.93(3)
N1—Zn1—N2 80.46(10)
N2—Zn1—Cl1 ⁱ 113.25(3)
N2—Zn1—Cl1 113.25(3)
C1—N1—Zn1 126.5(2)
C5—N1—Zn1 114.5(2)
C6—N2—Zn1 114.2(2)
C10—N2—Zn1 126.8(2)

Symmetry code: (i) $x, 1/2-y, z$