

SUPPLEMENTARY MATERIAL TO
**Spectroscopic and structural characterization of
hexaamminecobalt(III) dibromide permanganate**

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Table S-I. Crystal data and structure refinement

| | [Co(NH ₃) ₆]Br ₂ (MnO ₄) | [Co(NH ₃) ₆]Cl ₂ (MnO ₄) ¹² |
|--------------------------|--|--|
| Empirical formula | Br ₂ H ₁₈ CoMnN ₆ O ₄ | C ₁₂ H ₁₈ CoMnN ₆ O ₄ |
| Formula weight | 439.89 | 350.97 |
| Temperature | 103(2) | 163(2) |
| Radiation and wavelength | Mo-Kα, λ=0.71073 Å | Mo-Kα, λ=0.71075 Å |
| Crystal system | monoclinic | Monoclinic |
| Space group | P2 ₁ /c | P2 ₁ /c |
| Unit cell dimensions | a = 13.9533(6) Å b = 7.4499(4) Å c = 12.3766(7) Å α = 90° β = 108.453(8)° γ = 90° | a = 13.6133(7) Å b = 7.3658(5) Å c = 12.3682(6) Å α = 90° β = 108.547(8)° γ = 90° |
| Volume | 1220.41(12) Å ³ | 1175.78(13) Å ³ |
| Z | 4 | 4 |
| Density (calculated) | 2.394 Mg/m ³ | 1.983 Mg/m ³ |

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| | | |
|--|--|--|
| Absorption coefficient, μ | 8.944 mm ⁻¹ | 2.941 mm ⁻¹ |
| $F(000)$ | 856 | 712 |
| Crystal color | red | Red |
| Crystal description | prism | Platelet |
| Crystal size | 0.45 x 0.43 x 0.11 mm | 0.56 x 0.41 x 0.11 mm |
| Absorption correction | numerical | Numerical |
| Max. and min. transmission | 0.4820.889 | 0.8680.987 |
| θ - range for data collection | 3.078 $\leq \theta \leq$ 30.504° | 3.157 $\leq \theta \leq$ 27.473° |
| Index ranges | -19 $\leq h \leq$ 19; -10 $\leq k \leq$ 10; -17 $\leq l \leq$ 17 | -17 $\leq h \leq$ 17; -9 $\leq k \leq$ 9; -16 $\leq l \leq$ 15 |
| Reflections collected | 39290 | 18320 |
| Completeness to 2 θ | 1.000 | 1.000 |
| Independent reflections | 3727 [$R(\text{int})=0.1132$] | 2684 [$R(\text{int})=0.0588$] |
| Reflections $I > 2\sigma(I)$ | 3309 | 2351 |
| Refinement method | full-matrix least-squares on F^2 | full-matrix least-squares on F^2 |
| Data/restraints/parameters | 3727 / 0 / 136 | 2684 / 0 / 136 |
| Goodness-of-fit on F^2 | 1.188 | 1.152 |
| Final R indices [$I > 2\sigma(I)$] | $R_1 = 0.0438$, $wR_2 = 0.0759$ | $R_1 = 0.0422$, $wR_2 = 0.0854$ |
| R indices (all data) | $R_1 = 0.0531$, $wR_2 = 0.0783$ | $R_1 = 0.0528$, $wR_2 = 0.0892$ |

Table S-II. Bond lengths [\AA] of compound **1**.

| | | | | |
|----------|----------|--|----------|----------|
| Mn1-O1 | 1.608(3) | | Mn1-O4 | 1.614(3) |
| Mn1-O3 | 1.620(3) | | Mn1-O2 | 1.627(3) |
| Co1-N1#1 | 1.958(3) | | Co1-N1 | 1.958(3) |
| Co1-N2#1 | 1.960(3) | | Co1-N2 | 1.960(3) |
| Co1-N3#1 | 1.965(3) | | Co1-N3 | 1.965(3) |
| Co2-N4#2 | 1.957(3) | | Co2-N4 | 1.957(3) |
| Co2-N5#2 | 1.962(3) | | Co2-N5 | 1.962(3) |
| Co2-N6 | 1.982(3) | | Co2-N6#2 | 1.982(3) |

Table S-III. Bond angles [°] of compound **1**.

| | | | | |
|---------------|----------|--|---------------|-----------|
| O1-Mn1-O4 | 109.3(1) | | O1-Mn1-O3 | 110.3(1) |
| O4-Mn1-O3 | 109.9(1) | | O1-Mn1-O2 | 110.1(1) |
| O4-Mn1-O2 | 108.5(1) | | O3-Mn1-O2 | 108.8(2) |
| N1#1-Co1-N1 | 180.0 | | N1#1-Co1-N2#1 | 89.6(1) |
| N1-Co1-N2#1 | 90.4(1) | | N1#1-Co1-N2 | 90.4(1) |
| N1-Co1-N2 | 89.6(1) | | N2#1-Co1-N2 | 180.0 |
| N1#1-Co1-N3#1 | 90.3(1) | | N1-Co1-N3#1 | 89.7(1) |
| N2#1-Co1-N3#1 | 89.0(1) | | N2-Co1-N3#1 | 91.0(1) |
| N1#1-Co1-N3 | 89.7(1) | | N1-Co1-N3 | 90.3(1) |
| N2#1-Co1-N3 | 91.0(1) | | N2-Co1-N3 | 89.0(1) |
| N3#1-Co1-N3 | 180.0(2) | | N4#2-Co2-N4 | 180.0(2) |
| N4#2-Co2-N5#2 | 89.9(1) | | N4-Co2-N5#2 | 90.1(1) |
| N4#2-Co2-N5 | 90.1(1) | | N4-Co2-N5 | 89.9(1) |
| N5#2-Co2-N5 | 180.0(2) | | N4#2-Co2-N6 | 89.2(1) |
| N4-Co2-N6 | 90.8(1) | | N5#2-Co2-N6 | 88.9(1) |
| N5-Co2-N6 | 91.1(1) | | N4#2-Co2-N6#2 | 90.8(1) |
| N4-Co2-N6#2 | 89.2(1) | | N5#2-Co2-N6#2 | 91.1(1) |
| N5-Co2-N6#2 | 88.9(1) | | N6-Co2-N6#2 | 180.00(9) |

Symmetry codes to generate equivalent atoms:

1. -x+1,-y+1,-z+1

2. -x,-y+1,-z

Table S-IV. Analysis of Potential Hydrogen Bonds and Schemes with $d(D...A) < R(D)+R(A)+0.50$, $d(H...A) < R(H)+R(A)-0.12$ Ang., $D-H...A > 100.0$ Deg in compound **1**.

| Nr | Donor | --- H...Acceptor | Symm. op. | D - H | H...A | D...A | D - H...A |
|----|-------|------------------|------------------|-------|-------|----------|-----------|
| 1 | N1 | --H1A ..Br2 | x,y,z | 0.91 | 2.87 | 3.402(3) | 119 |
| 2 | N1 | --H1A ..O1 | x,1+y,z | 0.91 | 2.45 | 2.937(4) | 113 |
| 3 | N1 | --H1B ..Br1 | 1-x,1/2+y,1/2-z | 0.91 | 2.57 | 3.460(3) | 165 |
| 4 | N1 | --H1C ..Br1 | x,3/2-y,1/2+z | 0.91 | 2.70 | 3.518(3) | 149 |
| 5 | N2 | --H2A ..Br1 | x,y,z | 0.91 | 2.66 | 3.539(3) | 164 |
| 6 | N2 | --H2B ..O3 | x,y,z | 0.91 | 2.59 | 3.273(4) | 132 |
| 7 | N2 | --H2C ..Br1 | 1-x,-1/2+y,1/2-z | 0.91 | 2.70 | 3.513(3) | 149 |
| 8 | N3 | --H3A ..Br2 | x,y,z | 0.91 | 2.86 | 3.555(3) | 134 |
| 9 | N3 | --H3A ..O1 | x,1/2-y,1/2+z | 0.91 | 2.49 | 2.930(4) | 110 |
| 10 | N3 | --H3B ..Br1 | 1-x,1-y,1-z | 0.91 | 2.62 | 3.484(3) | 159 |
| 11 | N3 | --H3C ..Br1 | x,1/2-y,1/2+z | 0.91 | 2.88 | 3.703(3) | 152 |
| 12 | N4 | --H4A ..O2 | -x,1-y,-z | 0.91 | 2.37 | 3.179(4) | 149 |
| 13 | N4 | --H4A ..O4 | -x,1-y,-z | 0.91 | 2.30 | 3.070(4) | 142 |
| 14 | N4 | --H4B ..O2 | x,1+y,z | 0.91 | 2.03 | 2.922(4) | 166 |
| 15 | N4 | --H4C ..O3 | -x,1/2+y,1/2-z | 0.91 | 2.14 | 3.028(4) | 165 |

| | | | | | | | | |
|----|----|-------|-------|-----------------|------|------|----------|-----|
| 16 | N5 | --H5A | ..O2 | x,y,z | 0.91 | 2.34 | 3.181(4) | 154 |
| 17 | N5 | --H5A | ..Br2 | -x,-1/2+y,1/2-z | 0.91 | 2.89 | 3.417(3) | 118 |
| 18 | N5 | --H5B | ..O2 | -x,1/2+y,1/2-z | 0.91 | 2.45 | 3.120(4) | 131 |
| 19 | N5 | --H5B | ..O3 | -x,1/2+y,1/2-z | 0.91 | 2.38 | 3.243(4) | 159 |
| 20 | N5 | --H5C | ..Br2 | x,y,z | 0.91 | 2.53 | 3.423(3) | 167 |
| 21 | N6 | --H6A | ..Br2 | x,3/2-y,-1/2+z | 0.91 | 2.72 | 3.619(3) | 169 |
| 22 | N6 | --H6B | ..O4 | x,y,z | 0.91 | 2.47 | 3.163(4) | 133 |
| 23 | N6 | --H6C | ..Br2 | x,y,z | 0.91 | 2.86 | 3.704(2) | 155 |
| 24 | N6 | --H6C | ..O1 | x,1+y,z | 0.91 | 2.58 | 3.063(4) | 114 |

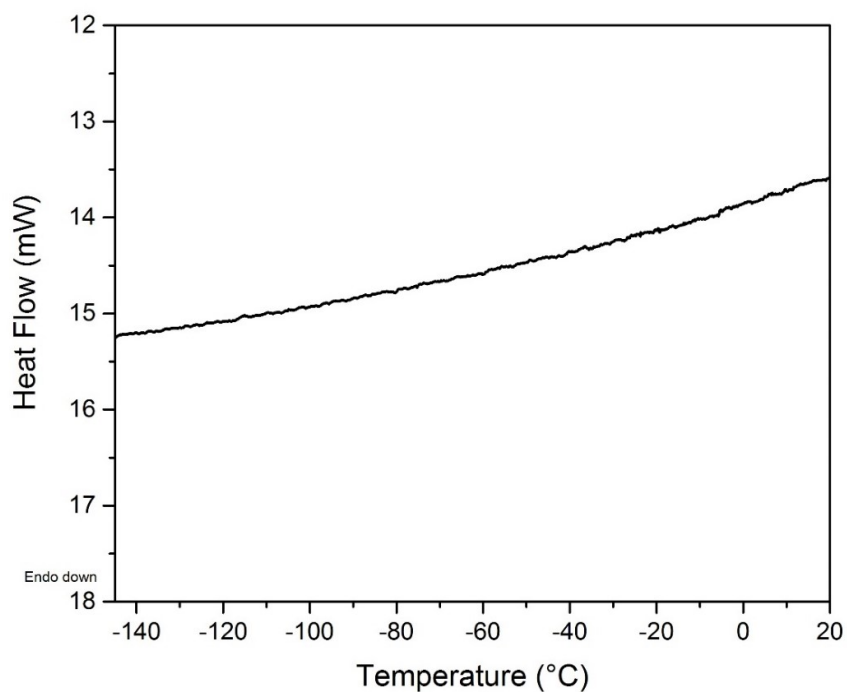


Fig. S-1. The low temperature (between -140 — 25 °C) DSC curve of compound **1**.

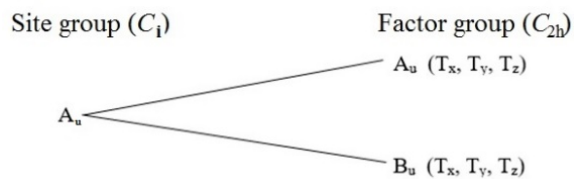


Fig. S-2. The correlation analysis for Co^{III} ions in compound **1**.

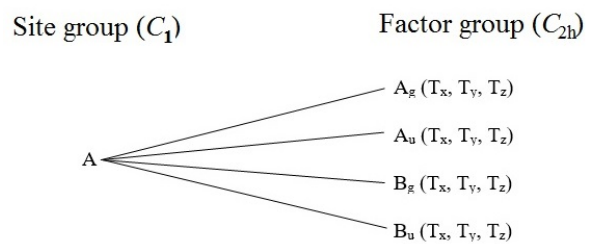


Fig. S-3. The correlation analysis for Br ions in compound **1**.

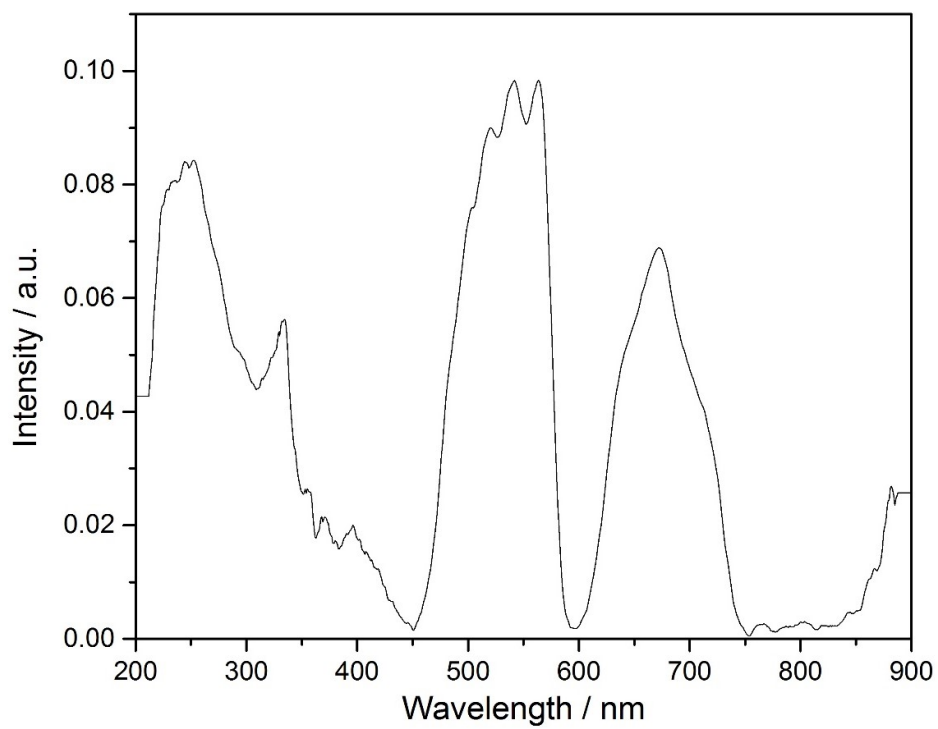


Fig. S-4. The UV-VIS spectra of compound **1**.