

SUPPLEMENTARY MATERIAL TO
Green synthesis of silver nanoparticles utilizing *Gardenia latifolia* extract, characterization and *in vitro* antibacterial activity

SATYAM SRIVASTAVA¹, PRASHANSHA SRIVASTAVA², GAURAV RAJ DWIVEDI²,
VENKATNARAYAN RAMANATHAN³ and HARISH CHANDRA UPADHYAY^{1*}

¹Laboratory of Chemistry, Department of Applied Sciences, Rajkiya Engineering College (Affiliated with Dr. A.P.J. Abdul Kalam Technical University, Lucknow), Churk, Sonbhadra-231206, India, ²Department of Microbiology, ICMR - Regional Medical Research Centre, Gorakhpur-273013, India and ³Department of Chemistry, Indian Institute of Technology (BHU), Varanasi-221005, India

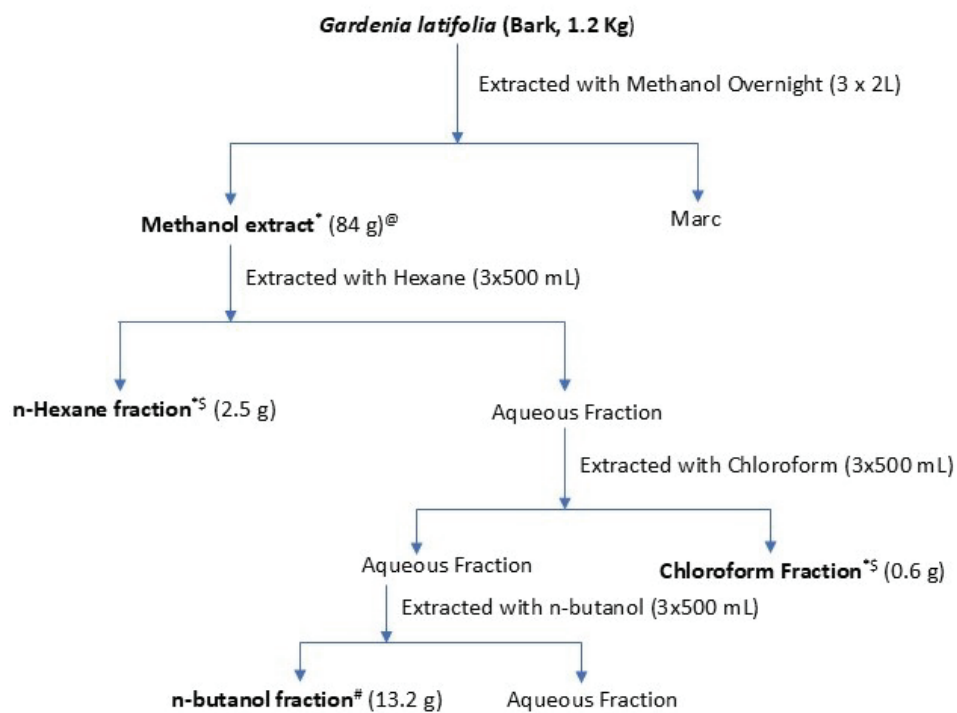
J. Serb. Chem. Soc. 90 (6) (2025) 709–725

PHYTOCHEMICAL INVESTIGATIONS OF CRUDE EXTRACT AND FRACTIONS

Table S-I. Preliminary phytochemical investigation of the bark of Indian plant *G. latifolia*

Functional groups	Test	Impression
Alkaloids	Dragendrof's reagent: Extract + dilute Hydrochloric acid + Dragendroff's reagent (potassium bismuth iodide solution). Red ppt. obtained indicated the presence of alkaloids	-
Flavonoids	Lead acetate test: Extract + 2-3 drops of lead acetate solution. The yellow color ppt. obtained indicated the presence of flavonoids.	+
Anthraquinones	Bontrager's test: Boil the extract + dilute sulfuric acid+ filter + chloroform. The organic layer obtained + 2-3 drops of strong ammonia solution. The lower ammoniacal layer takes on a pink or red color indicating the presence of anthraquinones	+
Tannins	Gelatin test: Extract + 1% gelatin solution containing NaCl. White ppt. obtained shows the presence of Tannins.	+
Saponins	Extract + 20 mL water. The formation of stable foam indicates the presence of saponins.	+
Steroids	Libermann-Burchard test: Extract + 2 mL of chloroform + ten drops of acetic anhydride and two drops of concentrated sulfuric acid. If the solution turns red, then blue, and finally bluish-green, a steroidal nucleus is present. If the solution turns purple or red, a triterpenoid nucleus is present.	+
Glycosides/Sugars	0.5 mL of crude extract +1 mL of distilled water and NaOH. The yellow tint obtained indicates the presence of glycosides/reducing sugars.	+

* Corresponding author. E-mail: harishcu@recsonbhadra.ac.in



@A part (24g) was taken out.

sWashed with water and dried over Na_2SO_4

*Solvent was completely removed under vacuum at 60-70°C on Buchi Rota Vapor.

#Solvent removed under vacuum by making azeotrope with water.

Fig. S-1. Flow chart of fractionation procedure of crude extract

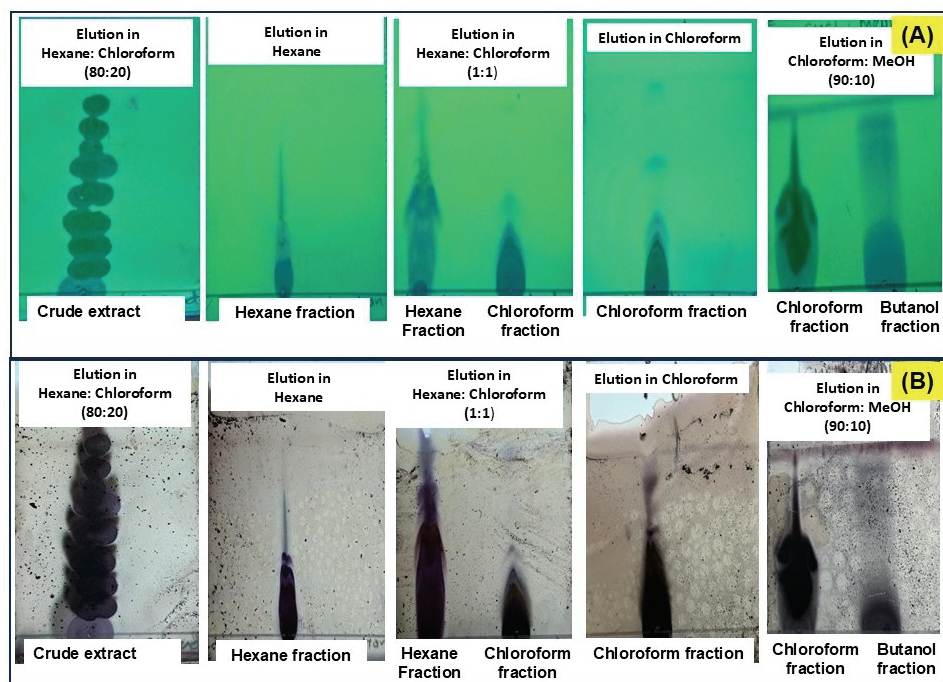


Fig. S-2. Thin layer chromatography profile of extract and fractions visualized under UV-254 nm (A), and on treated with a vanillin-sulphuric acid (1: 5, w/v) solution in ethanol, and subsequently heated at 95 °C for 5 minutes (B).

CHARACTERIZATION OF THE AGNPS FORMULATIONS

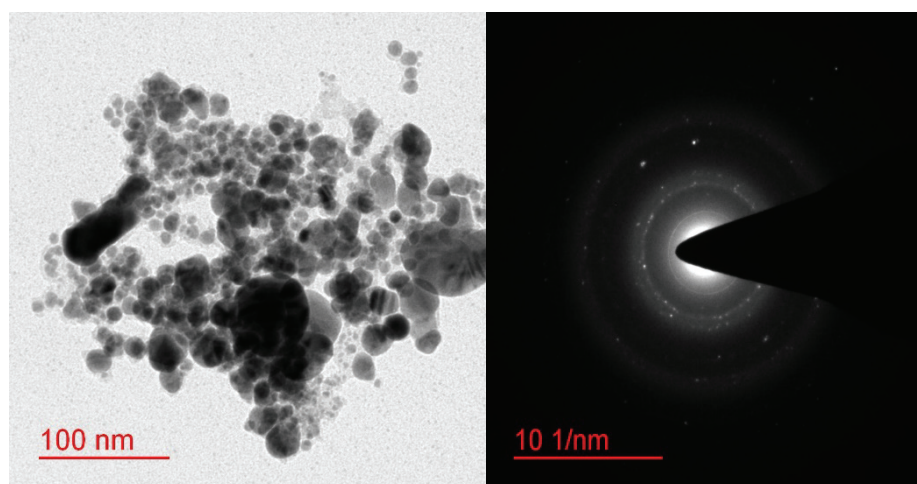


Fig. S-3. HR-TEM and SAED images of the AgNPs formulation F2

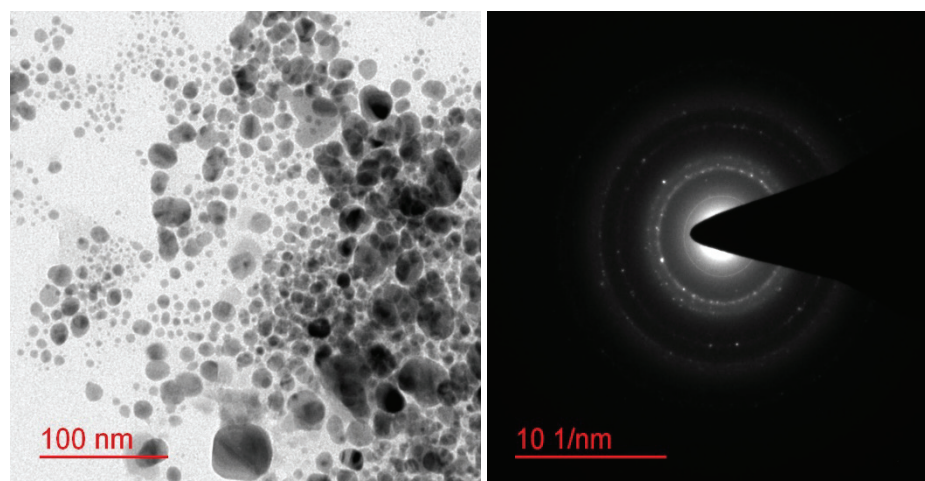


Fig. S-4. HR-TEM and SAED images of the AgNPs formulation **F4**

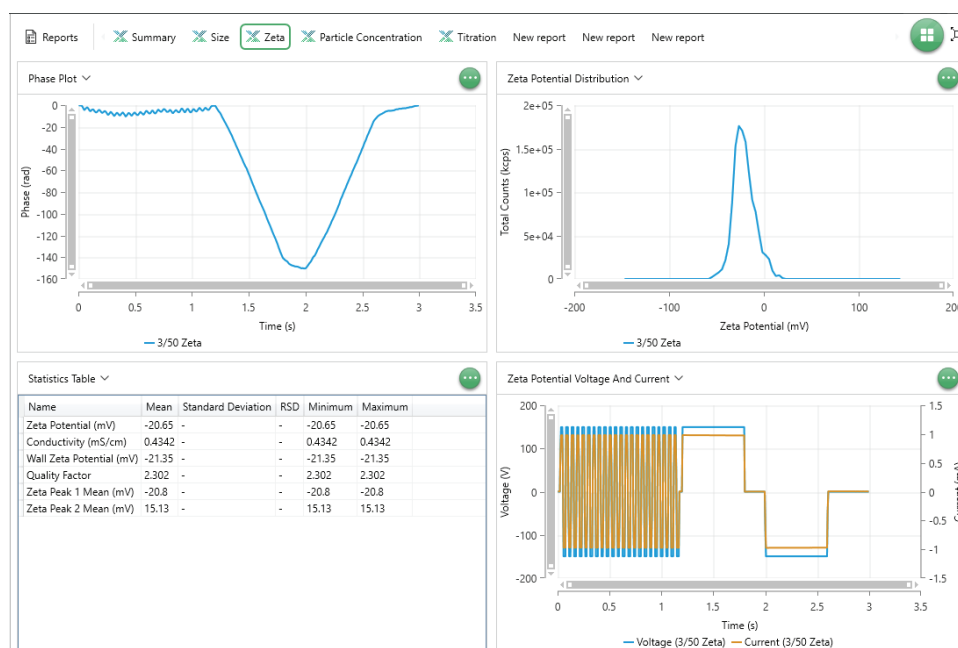


Fig. S-5. Zeta potential of the AgNPs formulation **F3**

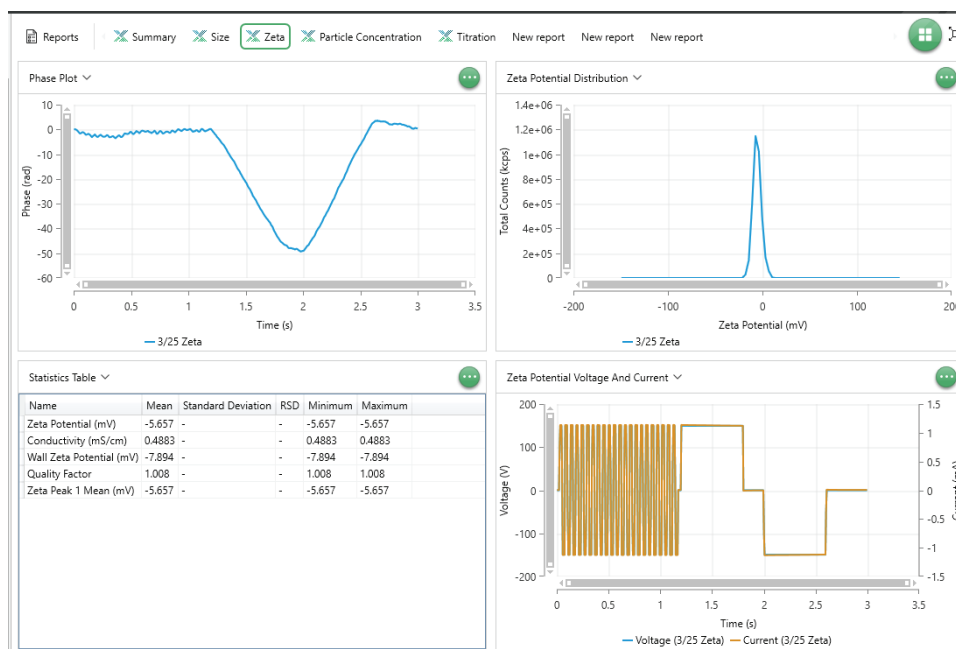


Fig. S-6. Zeta potential of the AgNPs formulation F4

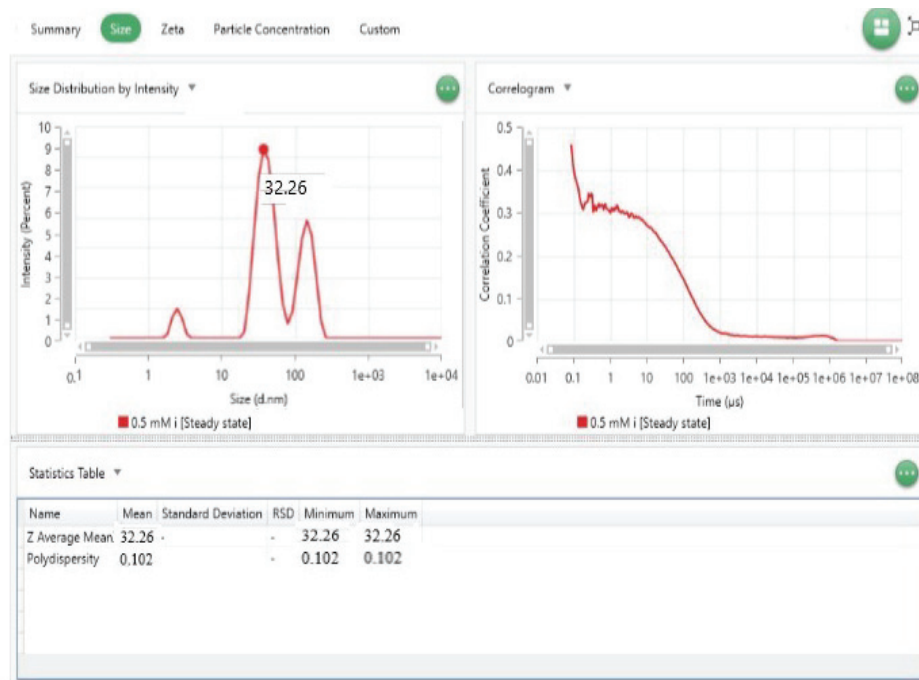


Fig. S-7. Hydrodynamic particle size of the AgNPs formulation F3

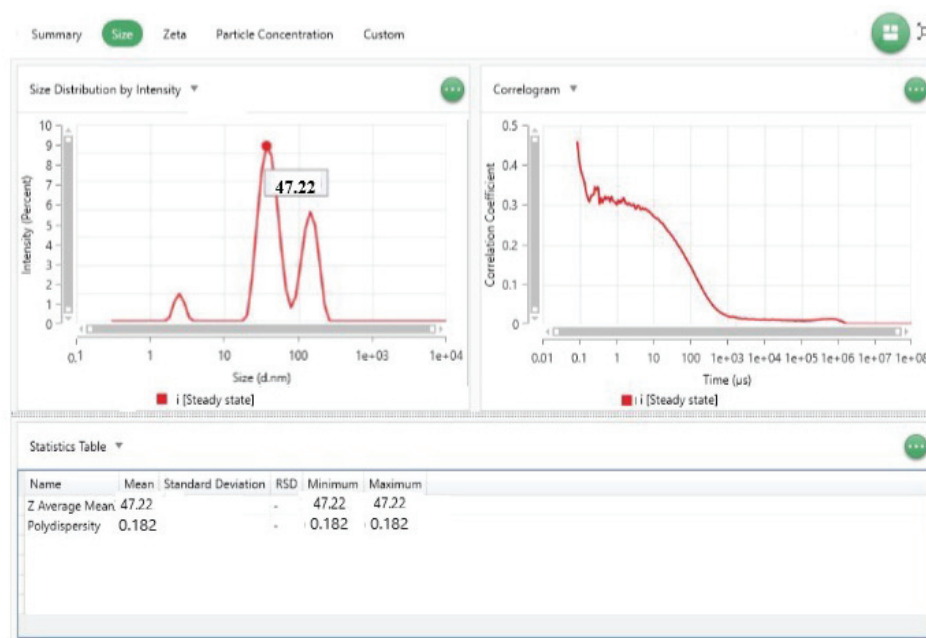


Fig. S-8. The hydrodynamic particle size of the AgNPs formulation F4.