

SUPPLEMENTARY MATERIAL TO
**Structural and electronic effects of cation binding
(Li⁺, Na⁺, K⁺, Mg²⁺ and Ca²⁺) to the π system of
the η^6 -benzene–Cr(CO)₃ complex: a theoretical study**

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J. Serb. Chem. Soc. 82 (10) (2017) 1123–1134

TABLE S-I. Computed λ_{\max} (maximum absorption wavelength), nm, f (oscillator strength), TDM (transition dipole moments), root mean square value, RMS , e-bohr, between the ground and n^{th} excited state of all the systems under study; Data corresponding to the maximum absorption wavelength are in bold letters

System	State transition	λ / nm	f	MB			RMS e-bohr
				Transition dipole moment			
				x	y	z	
BC	$S_0 \rightarrow S_3$	168.03	0.5587	-0.1278	-0.4628	0.0000	0.480
	$S_0 \rightarrow S_4$	168.03	0.5588	-0.4629	0.1278	0.0000	0.480
	$S_0 \rightarrow S_8$	155.08	0.0056	0.0000	0.0000	-0.0087	0.009
Li ⁺ -BC	$S_0 \rightarrow S_2$	193.20	0.0202	0.0085	0.0635	0.0000	0.064
	$S_0 \rightarrow S_3$	193.19	0.0202	-0.0636	0.0084	0.0000	0.064
	$S_0 \rightarrow S_6$	165.65	0.5223	-0.1315	0.4292	0.0000	0.449
Na ⁺ -BC	$S_0 \rightarrow S_7$	165.65	0.5222	-0.4292	-0.1315	0.0000	0.449
	$S_0 \rightarrow S_{10}$	163.62	0.0066	0.0000	0.0001	-0.0277	0.028
	$S_0 \rightarrow S_1$	230.59	0.0111	-0.0067	0.0494	0.0001	0.050
	$S_0 \rightarrow S_2$	230.57	0.0111	0.0495	0.0069	0.0001	0.050
	$S_0 \rightarrow S_5$	168.36	0.5276	-0.0938	0.444	0.0009	0.454
K ⁺ -BC	$S_0 \rightarrow S_6$	168.35	0.5276	0.4441	0.0938	0.001	0.454
	$S_0 \rightarrow S_1$	230.94	0.0085	0.0000	0.0002	-0.0395	0.040
	$S_0 \rightarrow S_2$	230.93	0.0085	0.0000	-0.0395	-0.0002	0.040
	$S_0 \rightarrow S_5$	172.58	0.0001	-0.0055	0.0000	0.0001	0.006
	$S_0 \rightarrow S_7$	171.21	0.1136	0.1767	-0.0001	0.0000	0.177
	$S_0 \rightarrow S_9$	169.78	0.3383	0.0000	-0.2698	0.2307	0.355
Mg ²⁺ -BC	$S_0 \rightarrow S_{10}$	169.76	0.3436	-0.0001	0.233	0.2718	0.358
	$S_0 \rightarrow S_1$	334.30	0.0014	0.0098	0.0108	0/0000	0.015
	$S_0 \rightarrow S_2$	334.28	0.0014	-0.0107	0.0098	-0.0001	0.015
	$S_0 \rightarrow S_8$	174.03	0.0109	0.0000	0.0000	-0.0519	0.052

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TABLE S-I. Continued

System	State transition	λ / nm	f	MB			RMS e-bohr	
				Transition dipole moment				
				x	y	z		
Ca ²⁺ -BC	S ₀ →S ₁	296.74	0.0027	0.0000	0.0000	0.0157	0.016	
	S ₀ →S ₂	296.72	0.0027	0.0000	0.0157	0.0000	0.016	
	S₀→S₁₀	173.55	0.0069	0.0335	-0.0001	0.0000	0.034	
MBC								
BC	S ₀ →S ₂	361.16	0.0004	-0.0011	-0.0122	0.0661	0.067	
	S ₀ →S ₃	361.10	0.0004	-0.0006	0.0657	0.0121	0.067	
	S ₀ →S ₄	346.01	0.0002	0.0323	0.0405	-0.0043	0.052	
	S ₀ →S ₅	345.91	0.0010	-0.0093	0.0202	-0.1022	0.105	
	S ₀ →S ₆	345.80	0.0009	-0.0117	0.0954	0.0238	0.099	
	S ₀ →S ₇	344.98	0.0005	0.0720	0.0005	-0.0060	0.072	
	S₀→S₈	328.93	0.0016	-0.0002	-0.1242	-0.0418	0.131	
	S ₀ →S ₉	328.89	0.0016	0.0008	-0.0419	0.1247	0.132	
	S ₀ →S ₁₀	308.11	0.0001	0.0006	-0.0353	-0.0010	0.035	
	Li ⁺ -BC	S ₀ →S ₄	390.76	0.0006	-0.0051	-0.0323	0.0820	0.088
S ₀ →S ₅		390.42	0.0006	0.0013	-0.0824	-0.0336	0.089	
S₀→S₆		352.21	0.0025	0.1700	0.0058	0.0026	0.170	
S ₀ →S ₇		349.04	0.0013	0.0925	-0.0814	0.0041	0.123	
S ₀ →S ₈		348.72	0.0012	-0.0476	-0.0692	-0.0820	0.117	
S ₀ →S ₉		348.49	0.0013	0.0531	0.0593	-0.0911	0.121	
S ₀ →S ₁₀		322.44	0.0006	0.0060	0.0061	-0.0785	0.079	
Na ⁺ -BC		S ₀ →S ₂	398.04	0.0006	0.0000	-0.0337	0.0837	0.090
		S ₀ →S ₃	397.86	0.0006	0.0018	0.0831	0.0338	0.090
		S ₀ →S ₄	391.39	0.0001	-0.0007	-0.0035	0.0404	0.041
	S ₀ →S ₅	391.27	0.0001	0.0007	0.0414	0.0036	0.042	
	S ₀ →S ₈	367.94	0.0012	-0.1217	-0.0008	-0.0002	0.122	
	S₀→S₉	351.84	0.0015	0.0012	-0.1334	0.0118	0.134	
	S ₀ →S ₁₀	351.64	0.0015	0.0001	-0.0119	-0.1333	0.134	
	K ⁺ -BC	S ₀ →S ₄	379.73	0.0007	0.0019	0.0929	0.0145	0.094
		S ₀ →S ₅	379.42	0.0007	-0.0026	-0.0143	0.0924	0.094
		S ₀ →S ₆	377.77	0.0001	-0.0001	0.0292	-0.0173	0.034
S ₀ →S ₇		377.69	0.0001	-0.0002	-0.0152	-0.0304	0.034	
S ₀ →S ₉		341.11	0.0001	-0.0332	-0.0051	-0.0049	0.034	
S₀→S₁₀		339.76	0.0013	-0.0026	-0.0183	0.1175	0.119	
Mg ²⁺ -BC		S ₀ →S ₁	1096.59	0.0002	0.0018	0.0651	0.0407	0.077
		S ₀ →S ₂	1095.48	0.0002	0.0018	0.0401	-0.0655	0.077
		S₀→S₃	792.33	0.0484	-1.1238	-0.0011	0.0001	1.124
		S ₀ →S ₅	402.55	0.0001	0.0020	-0.0285	-0.0024	0.029
	S ₀ →S ₆	402.43	0.0001	-0.0014	0.0018	-0.0282	0.028	
	S ₀ →S ₇	395.43	0.0001	-0.0007	0.0356	0.0053	0.036	
	S ₀ →S ₈	395.14	0.0001	0.0001	-0.0057	0.0363	0.037	
	S ₀ →S ₉	374.08	0.0016	0.0000	0.1391	-0.0113	0.140	
	S ₀ →S ₁₀	373.93	0.0016	0.0010	0.0112	0.1395	0.140	

TABLE S-I. Continued

System	State transition	λ / nm	f	MB			RMS e-bohr
				Transition dipole moment			
				x	y	z	
MBC							
Ca ²⁺ -BC	S ₀ →S ₁	744.07	0.0003	0.0002	0.0394	-0.0834	0.092
	S ₀ →S ₂	743.49	0.0003	0.0013	0.0835	0.0395	0.092
	S₀→S₃	667.06	0.0174	-0.6181	0.0001	-0.0001	0.618
	S ₀ →S ₅	430.26	0.0001	0.0008	-0.0075	-0.0359	0.037
	S ₀ →S ₆	430.18	0.0001	-0.0001	0.0359	-0.0077	0.037
	S ₀ →S ₉	374.44	0.0009	0.0003	-0.0892	0.0523	0.103
	S ₀ →S ₁₀	374.36	0.0009	0.0003	-0.0518	-0.0892	0.103

TABLE S-II. NTO pictures and MO transitions corresponding to $\lambda_{\text{max}} / \text{nm}$ of the MB systems under study

System	$\lambda_{\text{max}} (f)$	Contribution, %	NTO		MO transitions, amplitude, contribution, %
			Hole	Particle	
B	168.035 (0.559)	50			HOMO-1 → LUMO 0.69 (47%)
					HOMO → LUMO+1 -0.69 (47%)
		50			HOMO-1 → LUMO+1 -0.16 (2%)
					HOMO → LUMO -0.16 (2%)
Li ⁺ -B	165.649 (0.522)	49			HOMO-1 → LUMO 0.62 (38%)
					HOMO → LUMO+1 -0.62 (38%)
		49			HOMO-1 → LUMO+1 0.33 (11%)
					HOMO → LUMO 0.33 (11%)
Na ⁺ -B	168.357 (0.528)	49			HOMO-1 → LUMO+2 0.70 (49%)
					HOMO → LUMO+1 -0.70 (49%)

TABLE S-II. Continued

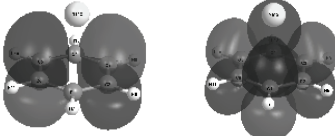
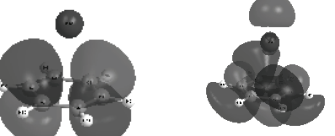

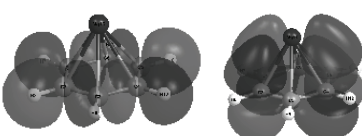
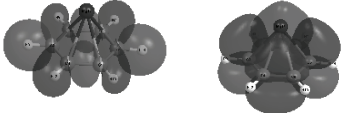
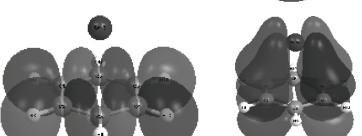
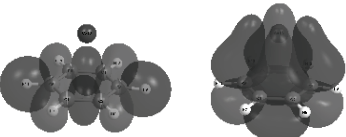
System	λ_{\max} (<i>f</i>)	NTO		MO transitions, amplitude, contribution, %
		Hole	Particle	
Na ⁺ -B	168.357 (0.528)	49		
K ⁺ -B	169.766 (0.344)	63		HOMO-1 → LUMO+3 0.57 (32%) HOMO → LUMO+1 0.57 (32%)
		34		HOMO → LUMO+5 0.46 (21%) HOMO-1 → LUMO+5 0.28 (8%) HOMO-1 → LUMO+1 0.15 (2%)
Mg ²⁺ -B	174.029 (0.011)	53		HOMO-2 → LUMO+1 0.73 (53%) HOMO-3 → LUMO+2 0.67 (45%)
		45		
Ca ²⁺ -B	173.551 (0.007)	49		HOMO-3 → LUMO+2 0.70 (49%) HOMO-2 → LUMO+1 0.70 (48%)
		48		

TABLE S-III. NTO pictures and MO transitions corresponding to λ_{\max} / nm of the MBC systems under study

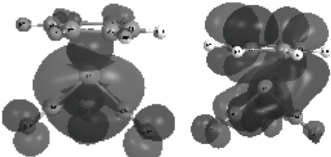
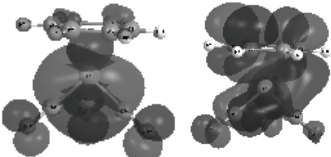
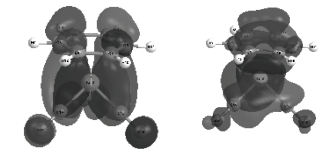
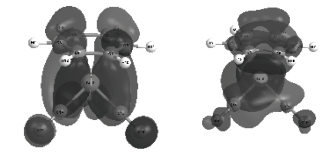
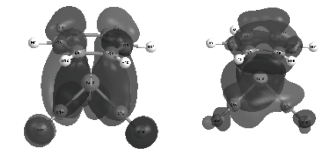
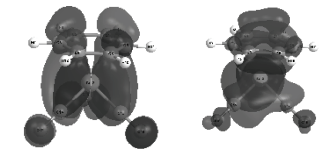
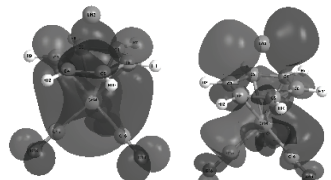
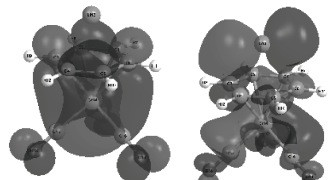
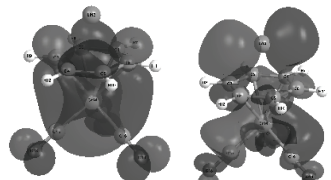
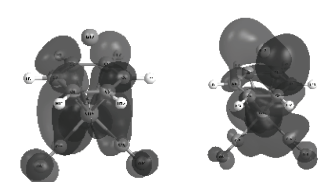
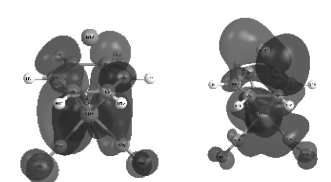
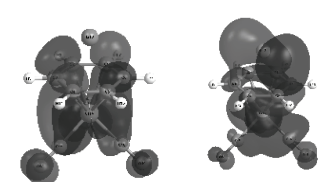
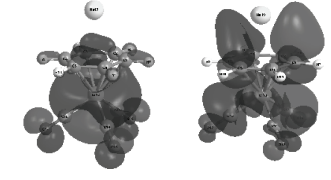
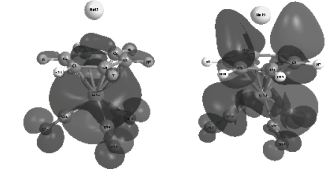
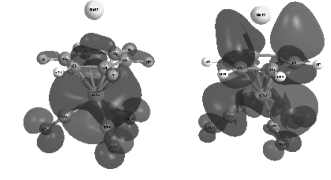
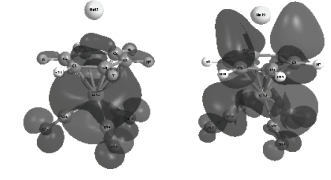
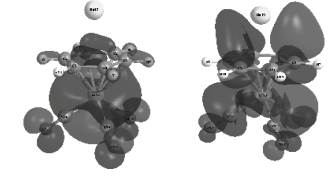
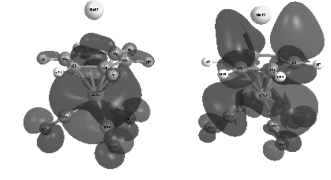


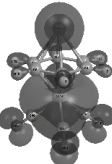
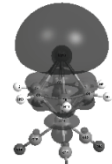

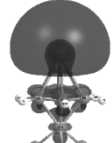
System	λ_{\max} (<i>f</i>)	NTO		MO transitions, amplitude, contribution, %
		Hole	Particle	
BC	328.925 (0.002)	71		HOMO-2 → LUMO+1 0.71 (51%)
				HOMO-2 → LUMO+2 -0.37 (14%)
		10		HOMO-1 → LUMO+3 0.28 (8%)
				HOMO → LUMO+2 -0.27 (7%)
				HOMO-2 → LUMO+13 -0.18 (3%)
				HOMO-1 → LUMO+12 0.15 (2%)
Li ⁺ -BC	352.210 (0.003)	49		HOMO → LUMO+5 0.45 (21%)
				HOMO → LUMO+4 -0.45 (20%)
				HOMO-1 → LUMO+4 -0.44 (19%)
		43		HOMO-1 → LUMO+4 -0.41 (17%)
				LUMO+5 HOMO → -0.22 (5%)
				LUMO+12 HOMO-1 → -0.20 (4%)
Na ⁺ -BC	351.840 (0.002)	77		HOMO-2 → LUMO+3 0.66 (43%)
				HOMO-2 → LUMO+11 0.31 (10%)
				HOMO-2 → LUMO+1 0.27 (7%)
				HOMO-2 → LUMO+6 0.27 (7%)
				HOMO → LUMO+4 0.23 (5%)
				HOMO-1 → LUMO+3 0.23 (5%)

TABLE S-III. Continued

System	λ_{\max} (<i>f</i>)	NTO		MO transitions, amplitude, contribution, %		
		Contri- bution, %	Hole	Particle		
K ⁺ -BC	339.759 (0.001)	70			HOMO-3 →	0.56 (31%)
					LUMO+2	
					HOMO-3 →	0.39 (15%)
					LUMO+1	
					HOMO-3 →	0.39 (15%)
					LUMO+5	
					HOMO-3 →	0.21 (4%)
LUMO+14						
HOMO-2 →	-0.20 (4%)					
LUMO+3						
HOMO-2 →	0.17 (3%)					
LUMO+2						
HOMO-2 →	0.99 (98%)					
LUMO						
Mg ²⁺ - -BC	792.337 (0.048)	98				
Ca ²⁺ - BC	667.052 (0.017)	99			HOMO-2 →	1.00 (99%)
					LUMO	

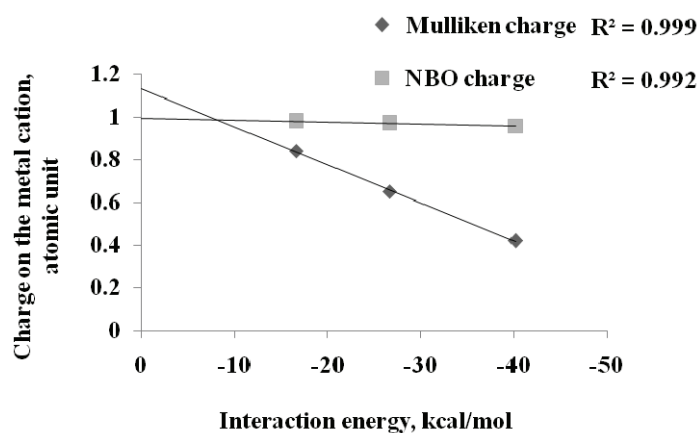


Fig. S-1. Interaction energy, kcal* mol⁻¹, vs. charge on the monovalent metal cations, atomic unit in the MB systems.

* 1 kcal = 4184 J

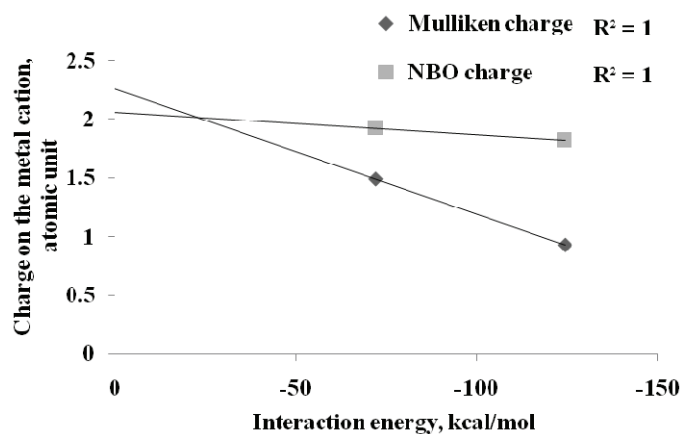


Fig. S-2. Interaction energy, kcal mol⁻¹, vs. charge on the divalent metal cations, atomic unit, in the MB systems.

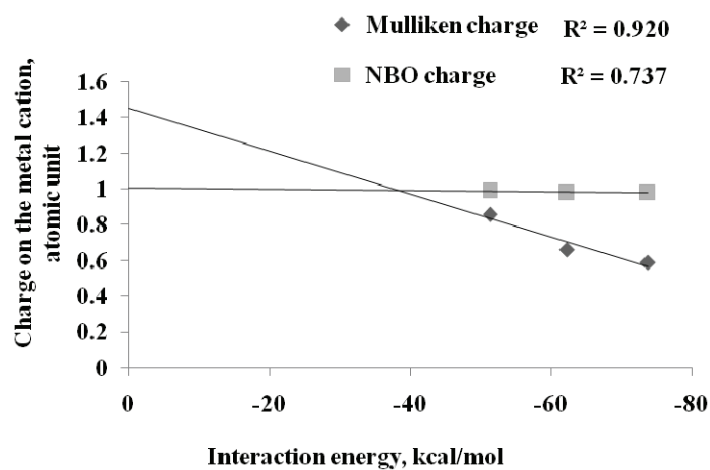


Fig. S-3. Interaction energy, kcal mol⁻¹, vs. charge on the monovalent metal cations, atomic unit, in the MBC systems.

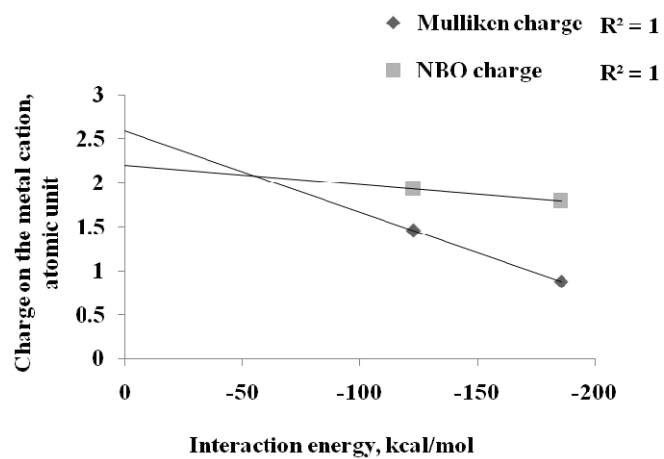


Fig. S-4. Interaction energy, kcal mol^{-1} , vs. charge on the divalent metal cations, atomic unit, in the MBC systems.

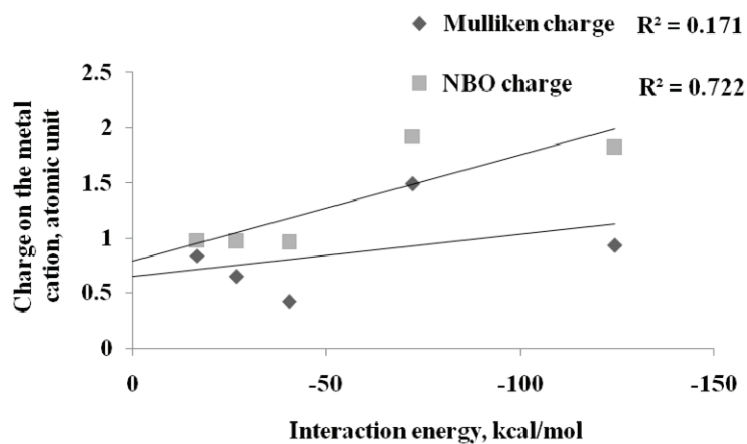


Fig. S-5. Interaction energy, kcal mol^{-1} , vs. charge on the metal cation, atomic unit, in the MB systems.

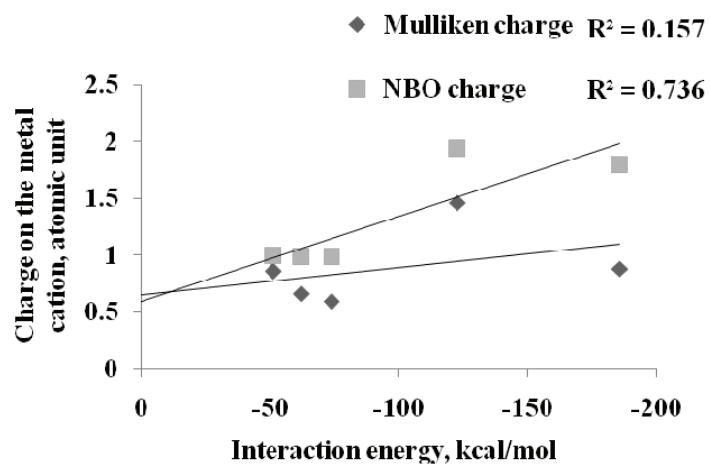


Fig. S-6. Interaction energy, kcal mol⁻¹, vs. charge on the metal cation, atomic unit, in the MBC systems.