



SUPPLEMENTARY MATERIAL TO
**Density functional theory calculation of lipophilicity for
organophosphate type pesticides**

FILIP VLAHOVIĆ¹, SAŠA IVANOVIĆ², MATIJA ZLATAR³ and MAJA GRUDEN^{4*}

¹University of Belgrade, Innovation center of the Faculty of Chemistry, Studentski trg 12–16, 11000 Belgrade, Serbia, ²University of Belgrade, Faculty of Veterinary Medicine, Department of Pharmacology and Toxicology, Bulevar oslobođenja 18, 11000 Belgrade Serbia,

³University of Belgrade, Institute of Chemistry, Technology and Metallurgy, Department of Chemistry, Njegoševa 12, 11000 Belgrade, Serbia and ⁴University of Belgrade, Faculty of Chemistry, Studentski trg 12–16, 11000 Belgrade, Serbia

J. Serb. Chem. Soc. 82 (12) (2017) 1369–1378

TABLE S-I. IUPAC names of all 22 OPs pesticides investigated in present study

Pesticide	Systematic name
Acephate	<i>N</i> -[Methoxy(methylsulfanyl)phosphoryl]acetamide
Aspon	<i>O,O,O',O'</i> -Tetrapropyl dithiopyrophosphate
Carbophenothion	<i>S</i> -[[4-Chlorophenyl]thio]methyl] <i>O,O</i> -diethyl phosphorodithioate
Chlorpyrifos	<i>O,O</i> -Diethyl <i>O</i> -3,5,6-trichloropyridin-2-yl phosphorothioate
Coumaphos	<i>O</i> -(3-Chloro-4-methyl-2-oxo-2H-chromen-7-yl) <i>O,O</i> -diethyl phosphorothioate
Crufomate	<i>N</i> -[(4- <i>tert</i> -Butyl-2-chlorophenoxy)methoxyphosphoryl]methanamine
Diazinon	<i>O,O</i> -Diethyl <i>O</i> -[2-isopropyl-6-methylpyrimidin-4-yl] phosphorothioate
Dichlorvos	2,2-Dichloroethenyl dimethyl phosphate
Dimethoate	<i>O,O</i> -Dimethyl <i>S</i> -[2-(methylamino)-2-oxoethyl] phosphorodithioate
Dioxathion	<i>S,S'</i> -1,4-Dioxane-2,3-diyli <i>O,O,O'</i> -tetraethyl bis(dithiophosphate)
Disulfoton	<i>O,O</i> -Diethyl <i>S</i> -[2-(ethylsulfanyl)ethyl] phosphorodithioate
Ethion	<i>O,O,O',O'</i> -Tetraethyl <i>S,S'</i> -methylene bis(phosphorodithioate)
Fenitrothion	<i>O,O</i> -Dimethyl <i>O</i> -(3-methyl-4-nitrophenyl)phosphorothioate
Fenthion	<i>O,O</i> -Dimethyl <i>O</i> -[3-methyl-4-(methylsulfanyl)phenyl] phosphorothioate
Fonofos	<i>O</i> -Ethyl <i>S</i> -phenyl ethylphosphonodithioate
Malathion	Diethyl 2-[(dimethoxyphosphorothiyl)sulfanyl]butanedioate
Methyl parathion	<i>O,O</i> -Dimethyl- <i>O</i> -4-nitrophenylphosphorothioate
Monocrotophos	Dimethyl (E)-1-methyl-2-(methylcarbamoyl)vinyl phosphate
Parathion	<i>O,O</i> -Diethyl <i>O</i> -4-nitrophenyl phosphorothioate

* Corresponding author. E-mail: gmaja@chem.bg.ac.rs

TABLE S-I. Continued

Pesticide	Systematic name
Phorate	<i>O,O</i> -Diethyl <i>S</i> -[(ethylsulfanyl)methyl] phosphorodithioate
Phosalone	6-Chloro-3-{[(diethoxyphosphinothioyl)sulfanyl]methyl}-1,3-benzoxazol-2-one
Temephos	<i>O,O,O',O'</i> -Tetramethyl <i>O,O'</i> -sulfanediylbis(1,4-phenylene) diphosphorothioate

TABLE S-II. Calculated solvation free energy change of transfer from the gas phase to water phase ($\Delta G_{\text{solv(water)}} / \text{kcal mol}^{-1}$) and octanol phase ($\Delta G_{\text{solv(octanol)}} / \text{kcal mol}^{-1}$) under standard state conditions, and corresponding $\log K_{\text{ow}}$ values of examined OPs pesticide set at M062X/6-31+G(d,p) level of theory, with experimentally determined $\log P$

Organophosphate	$\Delta G_{\text{solv(water)}}$	$\Delta G_{\text{solv(octanol)}}$	$\log K_{\text{ow}}$	$\log P$ (exp.)	Ref.
Acephate	-17.7	-17.1	-0.4	-0.8	1
Aspon	-10.3	-19.3	5.8	6.0	1
Carbophenothion	-9.9	-17.0	5.2	5.3	1
Chlorpyrifos	-5.7	-13.0	5.4	5.0	2
Coumaphos	-12.2	-17.7	4.1	4.5	3
Crufomate	-11.9	-15.6	2.7	3.4	1
Diazinon	-8.8	-14.3	4.0	3.8	1
Dichlorvos	-7.3	-10.3	2.2	1.4	1
Dimethoate	-15.2	-16.1	0.7	0.8	1
Dioxathion	-15.2	-22.3	5.2	4.3	1
Disulfoton	-9.3	-16.2	5.1	4.0	1
Ethion	-10.9	-18.3	5.5	5.1	1
Fenitrothion	-7.9	-11.8	2.8	3.3	1
Fenthion	-7.7	-13.2	4.0	4.1	4
Fonofos	-11.3	-16.3	3.7	3.9	1
Malathion	-13.8	-17.4	2.7	2.4	1
Methyl parathion	-7.9	-11.3	2.5	2.9	1
Monocrotophos	-16.8	-15.7	-0.8	-0.2	1
Parathion	-8.4	-13.4	3.7	3.8	1
Phorate	-8.1	-14.0	4.3	3.6	1
Phosalone	-12.7	-17.8	3.8	4.4	1
Temephos	-11.1	-19.0	6.6	6.0	1

TABLE S-III. Calculated solvation free energy change of transfer from the gas phase to water phase ($\Delta G_{\text{solv(water)}/\text{kcal mol}^{-1}}$) and octanol phase ($\Delta G_{\text{solv(octanol)}/\text{kcal mol}^{-1}}$) under standard state conditions, and corresponding $\log K_{\text{ow}}$ values of examined OPs pesticide set at PBE/6-31+G(d,p) level of theory, with experimentally determined $\log P$

Organophosphate	$\Delta G_{\text{solv(water)}}$	$\Delta G_{\text{solv(octanol)}}$	$\log K_{\text{ow}}$	$\log P$ (exp.)	Ref.
Acephate	-15.5	-15.5	0.0	-0.8	1
Aspon	-8.7	-19.1	6.3	6.0	1
Carbophenothion	-9.3	-16.8	5.5	5.3	1
Chlorpyrifos	-5.2	-12.5	5.3	5.0	2
Coumaphos	-11.2	-16.9	4.1	4.5	3
Crufomate	-9.3	-14.8	4.0	3.4	1
Diazinon	-8.8	-14.6	3.8	3.8	1

TABLE S-III. Continued

Organophosphate	$\Delta G_{\text{solv(water)}}$	$\Delta G_{\text{solv(octanol)}}$	$\log K_{\text{ow}}$	$\log P$ (exp.)	Ref.
Dichlorvos	-5.9	-9.3	2.5	1.4	1
Dimethoate	-15.0	-16.2	0.9	0.8	1
Dioxathion	-14.4	-22.5	5.9	4.3	1
Disulfoton	-8.9	-16.1	5.3	4.0	1
Ethion	-10.4	-18.7	6.0	5.1	1
Fenitrothion	-6.9	-11.1	3.1	3.3	1
Fenthion	-7.1	-12.7	4.2	4.1	4
Fonofos	-9.9	-15.5	4.1	3.9	1
Malathion	-12.8	-16.7	2.9	2.4	1
Methyl parathion	-7.0	-10.6	2.6	2.9	1
Monocrotophos	-14.7	-15.2	0.4	-0.2	1
Parathion	-7.5	-12.7	4.2	3.8	1
Phorate	-7.8	-14.3	4.8	3.6	1
Phosalone	-11.9	-17.3	4.0	4.4	1
Temephos	-9.7	-18.3	7.6	6.0	1

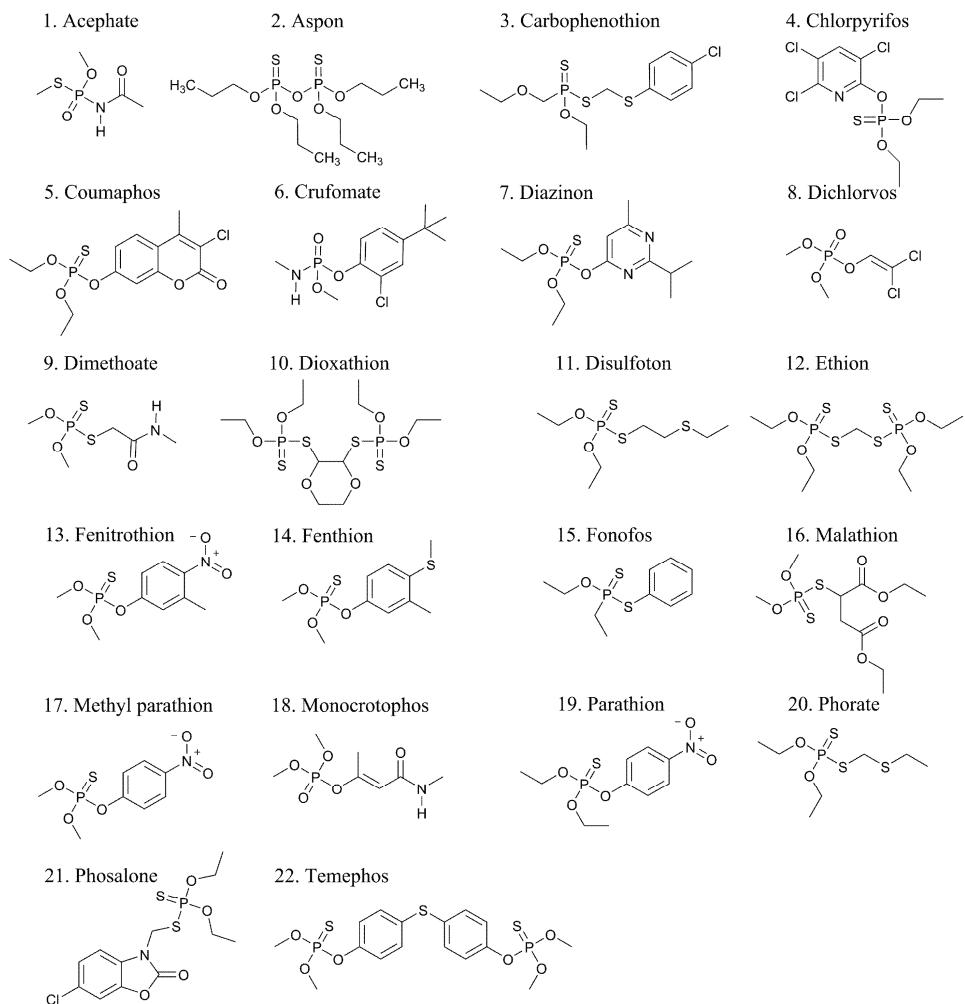


Fig. S-1. The structures of 22 OPs pesticides presented in this work.

REFERENCES

1. Y. C. Martin, *J. Med. Chem.* **39** (1996) 1189
2. J. Sangster, *J. Phys. Chem. Ref. Data* **18** (1989) 1111
3. C. T. Garten, J. R. Trabalka, *Environ. Sci. Technol.* **17** (1983) 590
4. B. T. Bowman, W. W. Sans, *J. Environ. Sci. Health, B* **18** (1983) 667.