

SUPPLEMENTARY MATERIAL TO

**Application of the redox system of *Nocardia corallina* B-276 in the enantioselective biotransformation of ketones and alcohols**

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J. Serb. Chem. Soc. 85 (3) (2020) 279–290

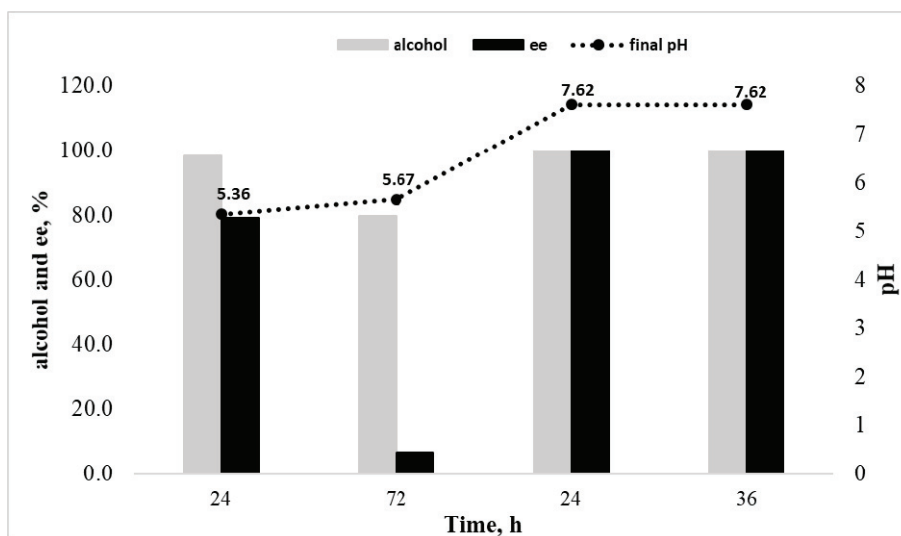


Fig. S-1. Reduction of ketone **1a** with *N. corallina* biomass, obtained at the final pH of the culture medium (5.36, 5.67 and 7.62).

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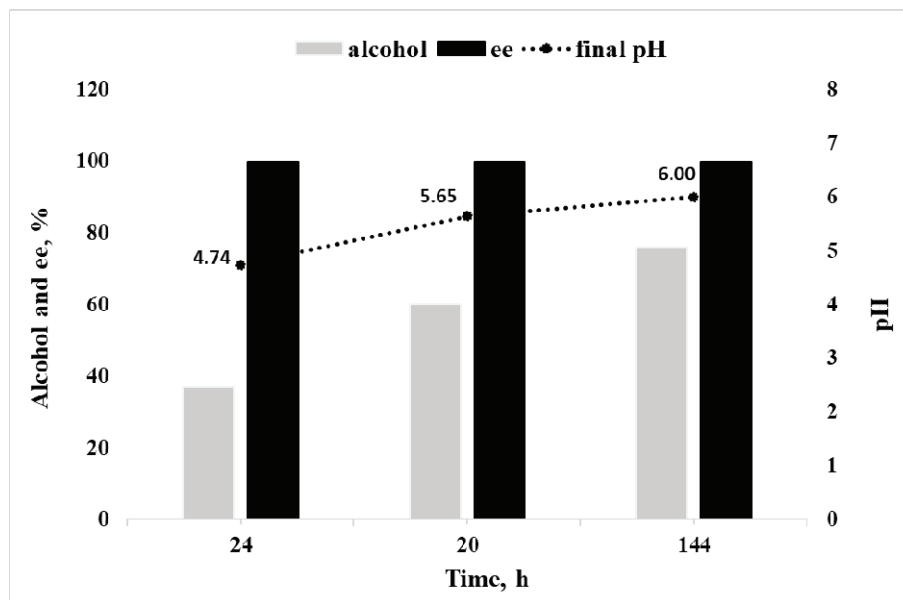


Fig. S-2. Biotransformation of **2a** with *N. corallina* biomass, obtained at the uncontrolled pH of the culture medium (4.74, 5.65 and 6.00).

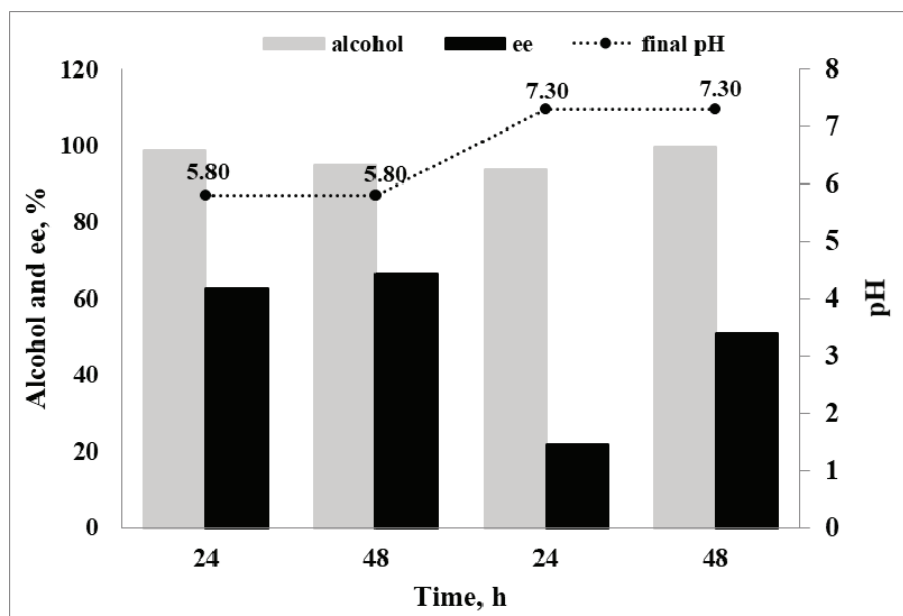
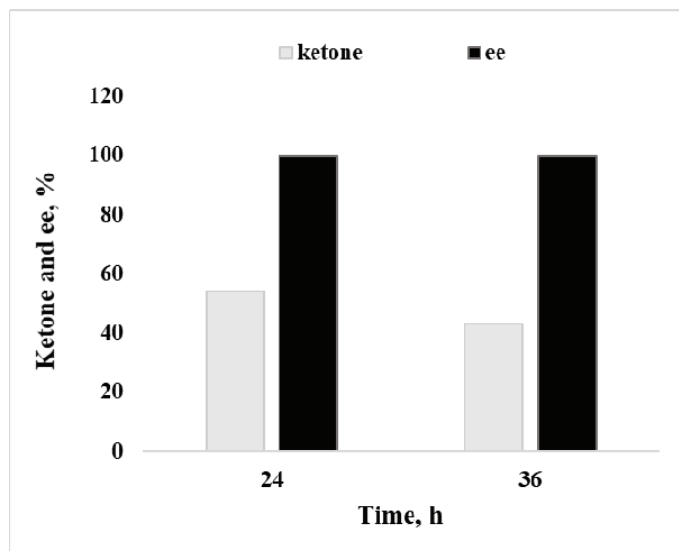
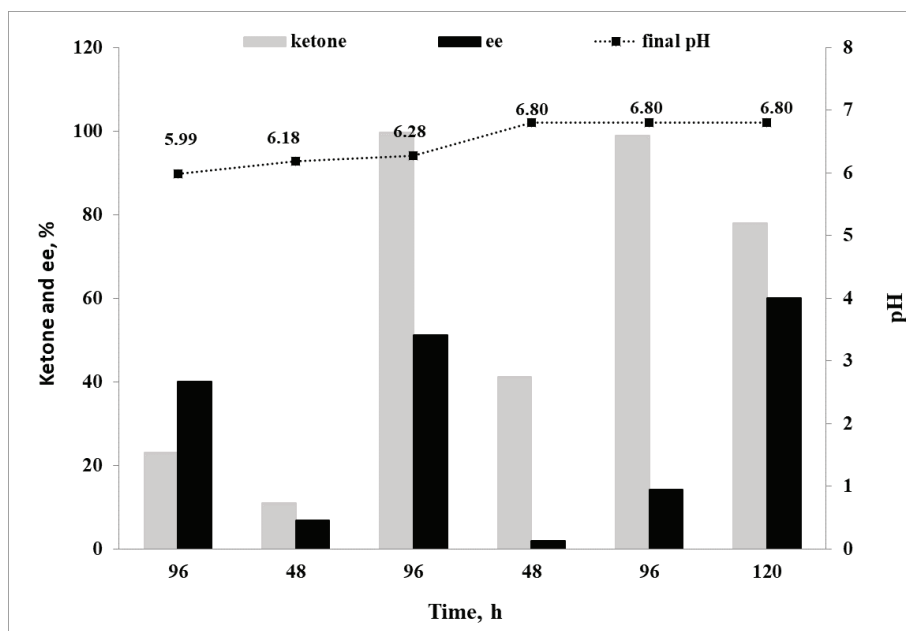
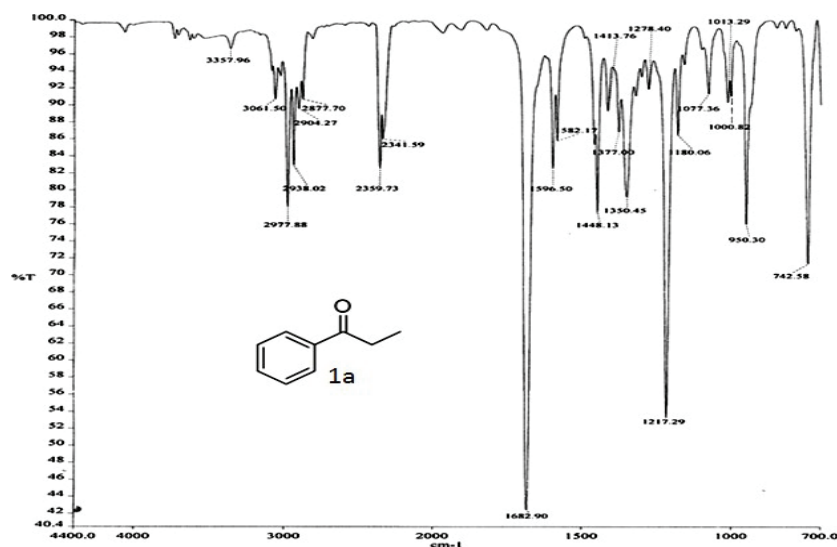
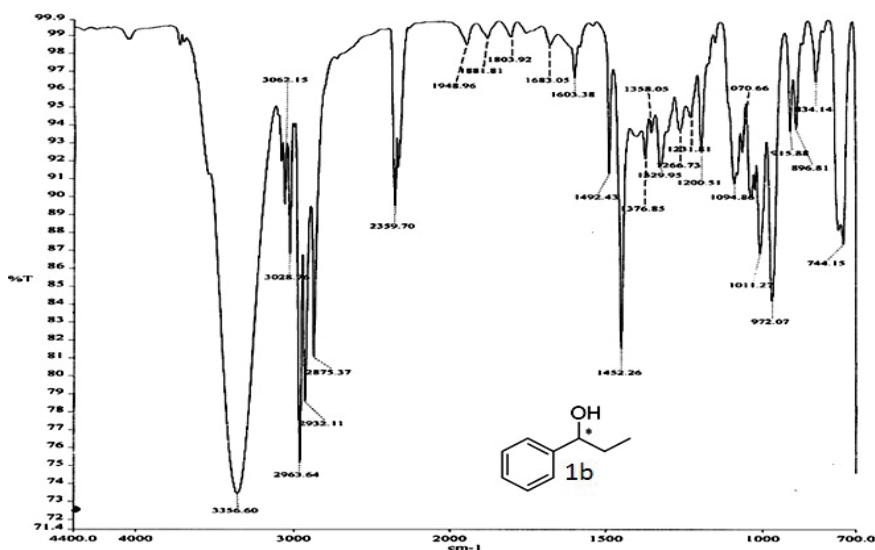


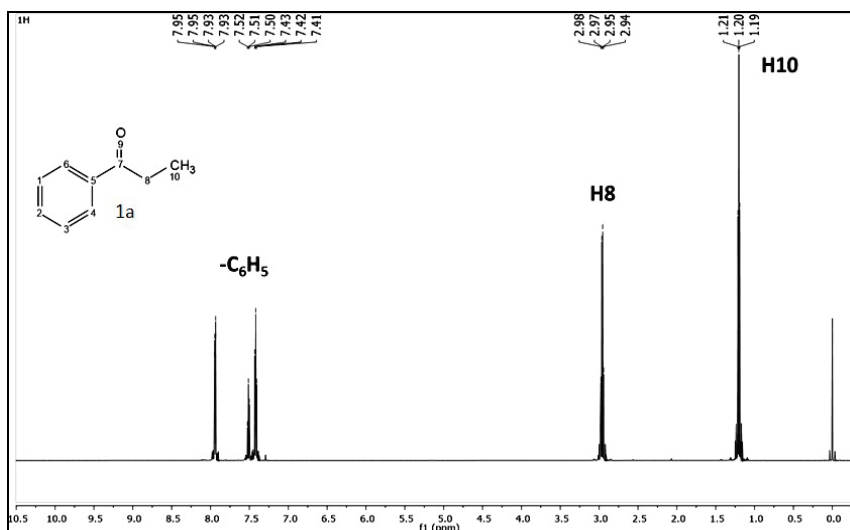
Fig. S-3. Biotransformation of **3a** with *N. corallina* biomass obtained at the uncontrolled pH of the culture medium (5.80 and 7.30).

Fig. S-4. Biotransformation of **1b** with *N. corallina* biomass, final pH 7.62.Fig. S5. Biotransformation of **2b** with *N. corallina* biomass obtained at the uncontrolled pH of the culture medium (5.99, 6.18, 6.28 and 6.80).

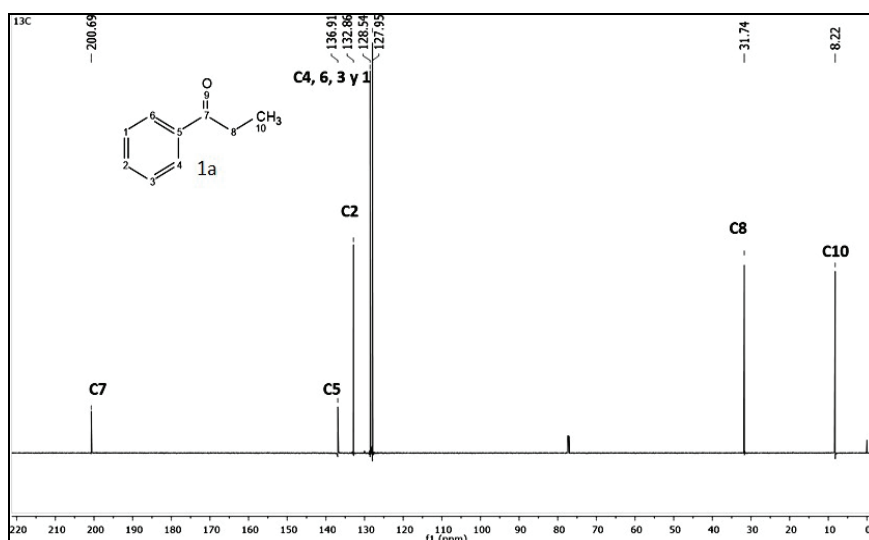
Fig. S-6. IR spectrum of 1-phenyl-1-propanone (**1a**).Fig. S-7. IR spectrum of 1-phenylpropan-1-ol (**1b**).

*The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of 1-phenyl-1-propanone (**1a**)*

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 1.20 (3H, *t*,  $J = 7.3$  Hz,  $\text{CH}_3$ ), 2.96 (2H, *q*,  $J = 7.2$  Hz,  $\text{CH}_2$ ), 7.42 (2H, *t*,  $J = 7.7$  Hz, CH), 7.51 (1H, *t*,  $J = 7.4$  Hz, CH), 7.94 (2H, *dd*,  $J = 8.4$  & 1.3 Hz, CH).

Figure S-8. <sup>1</sup>H-NMR spectrum of 1-phenyl-1-propanone (**1a**), in CDCl<sub>3</sub>.

<sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>, δ/ ppm): 8.4 (C10), 31.7 (C8), 127.9/128.5 (C4, 3, 6 & 1), 132.9 (C2), 136.9 (C5), 200.7 (C7).

Fig. S-9. <sup>13</sup>C-NMR spectrum of 1-phenyl-1-propanone (**1a**), in CDCl<sub>3</sub>.

The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of *l*-phenylpropan-1-ol (**1b**)

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 0.77 (3H, *t*,  $J = 7.5$  Hz,  $\text{CH}_3$ ), 1.61 (2H, *ddt*,  $J = 2.1, 13.6$  &  $6.9$  Hz,  $\text{CH}_2$ ), 3.99 (1H, *s*, OH), 4.41 (1H, *t*,  $J = 6.7$  Hz, CH), 7.13–7.19 (5H, *dt*,  $J = 12.8$  &  $6.9$  Hz, CH).

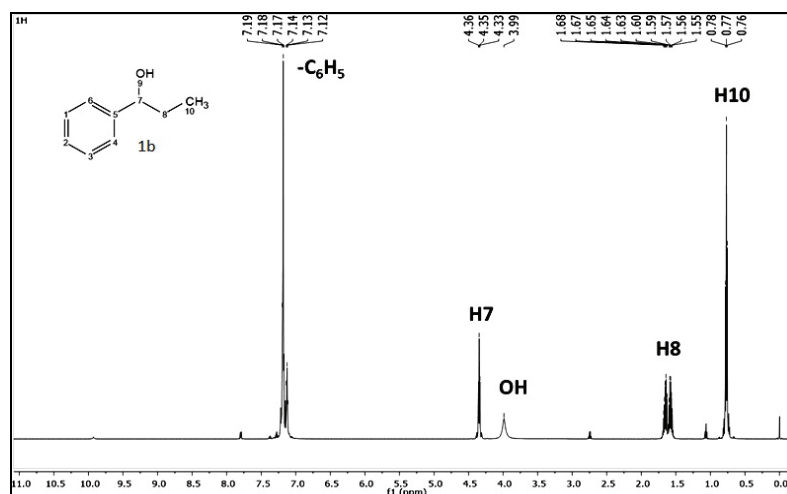


Fig. S-10.  $^1\text{H}$ -NMR spectrum of *l*-phenylpropan-1-ol (**1b**), in  $\text{CDCl}_3$ .

$^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm):  $\delta$  10.2 (C10), 31.9 (C8), 75.7 (C7), 126.2 (C2), 127.3 (C6 & C4), 128.3 (C1 & C3) and 144.9 (C5) ppm.

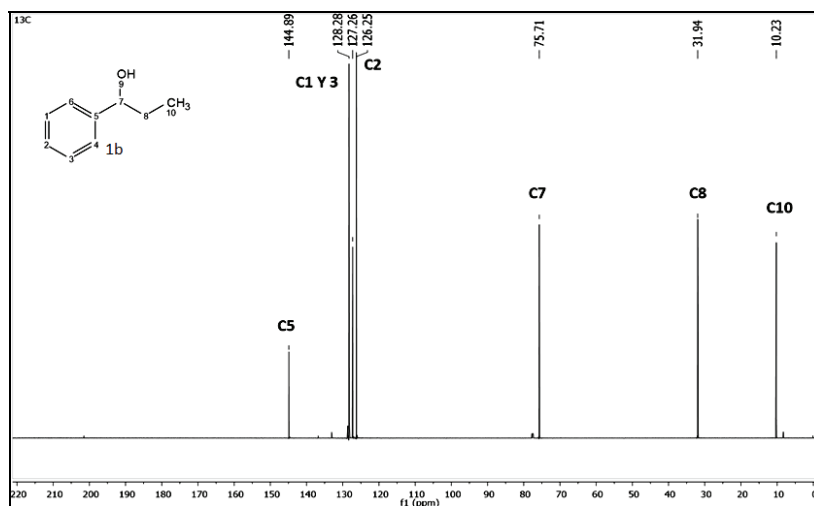


Fig. S-11.  $^{13}\text{C}$ -NMR spectrum of *l*-phenylpropan-1-ol (**1b**), in  $\text{CDCl}_3$ .

The GC chromatograms of **1a** and **1b** with retention times of 4.39 and 5.63 min, respectively under conditions reported in the Experimental Section.



Fig. S-12. GC chromatograms of 1-phenyl-1-propanone (**1a**) and 1-phenylpropan-1-ol (**1b**).

The HPLC chromatogram of a mixture of **1a** & *rac*-(**1b**), separately **1a** and (*S*)-**1b** & (*R*)-**1b** with retention times of: (20.4, 11.4 and 9.6 min, respectively, column OB-H and 7.7, 14.9 and 12.7 min, respectively, column OD) with conditions reported in the Experimental Section.

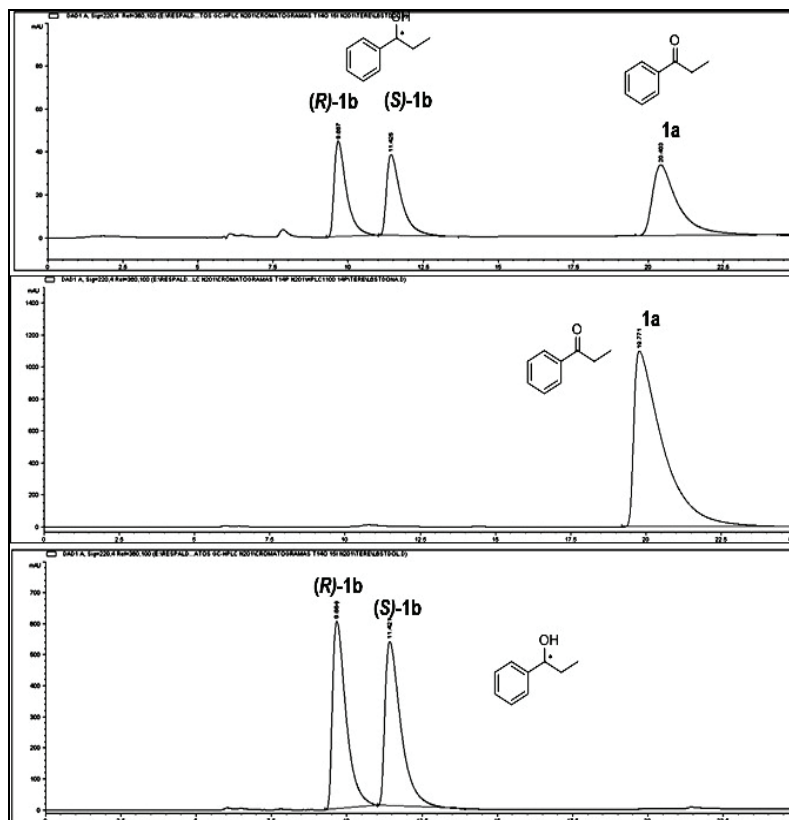


Fig. S-13. HPLC chromatogram (OB-H column) of mixture of 1-phenyl-1-propanone (**1a**) & *rac*-1-phenylpropan-1-ol (**1b**), separately **1a** and (*S*)-**1b** & (*R*)-**1b**.



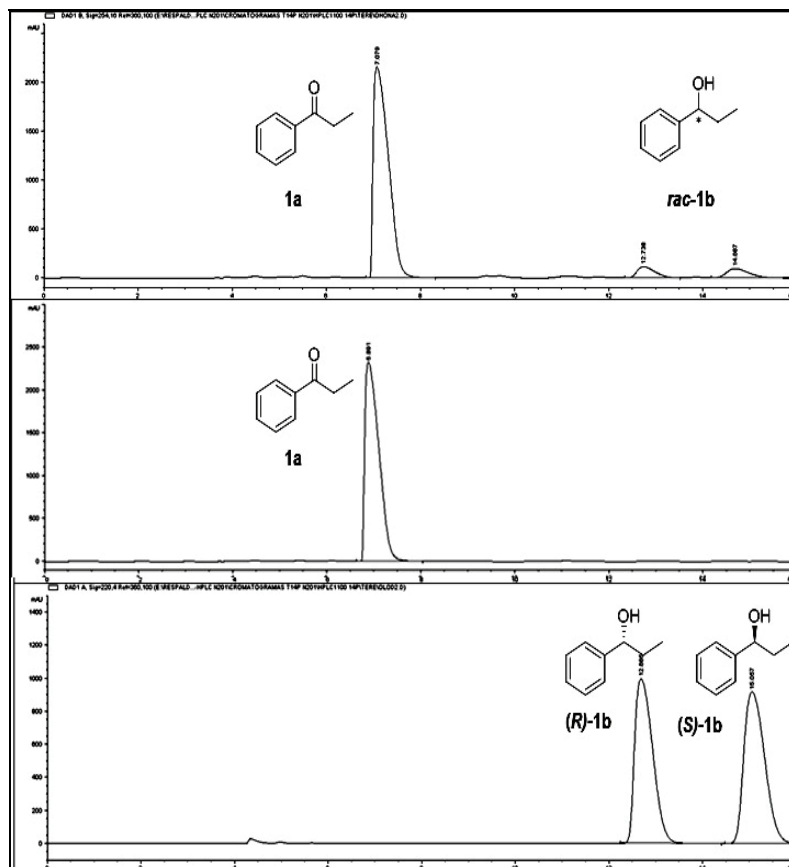
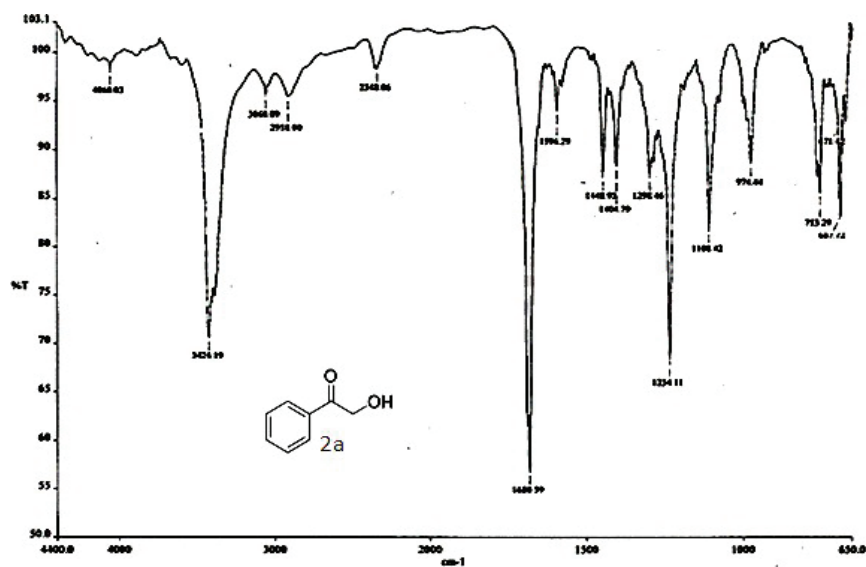
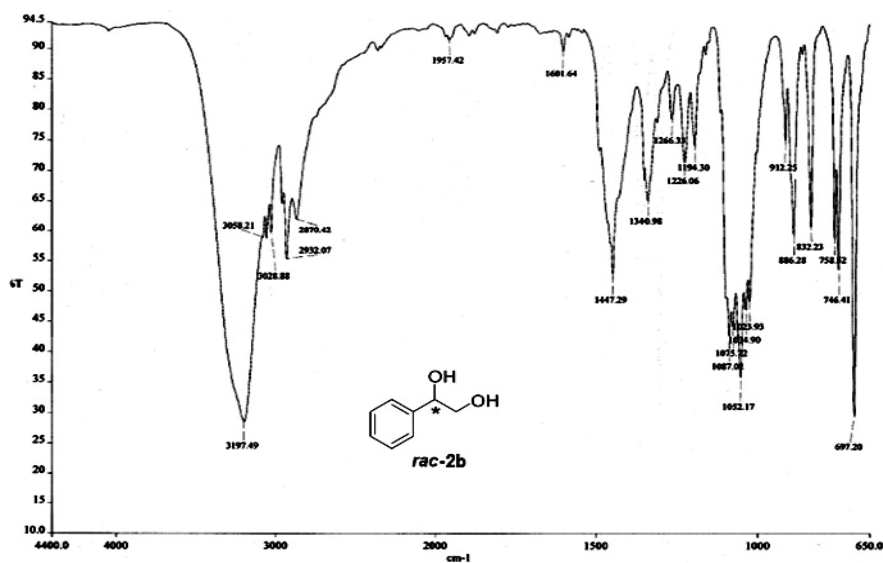


Fig. S-14. HPLC chromatogram (OD column) of a mixture of 1-phenyl-1-propanone (**1a**) & *rac*-1-phenylpropan-1-ol (**1b**), separately **1a** and (*S*)-**1b** & (*R*)-**1b**.

Fig. S-15. IR spectrum of 2-hydroxy-1-phenylethanone (**2a**).Fig. S-16. IR spectrum of 1-phenyl-1,2-ethanediol (**2b**).

The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of 2-hydroxy-1-phenylethanone (**2a**)

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 4.9 (2H, *m*,  $\text{CH}_2$ ), 5.4 (1H, *s*, OH), 7.5 (2H, *dd*,  $J = 11.9$  & 3.9 Hz, CH), 7.63 (1H, *t*,  $J = 7.7$  Hz, CH), 7.93 (2H, *dd*,  $J = 8.3$  & 1.1 Hz, CH).

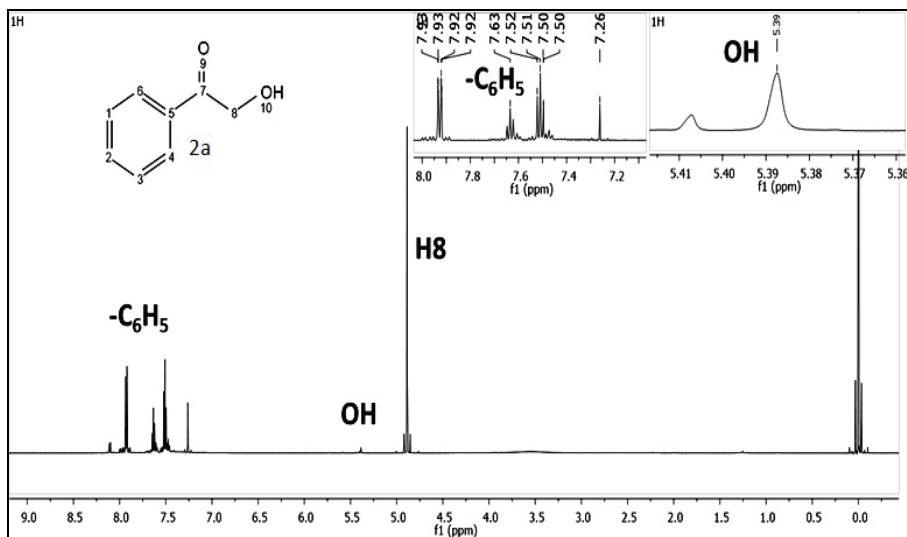


Fig. S-17.  $^1\text{H}$ -NMR spectrum of 2-hydroxy-1-phenylethanone (**2a**) in  $\text{CDCl}_3$ .

$^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 65.5 (C8), 127.7/128.4 (C4 & C1), 128.9/129.5 (C6 & C1), 133.6 (C2), 198.4 (C7).

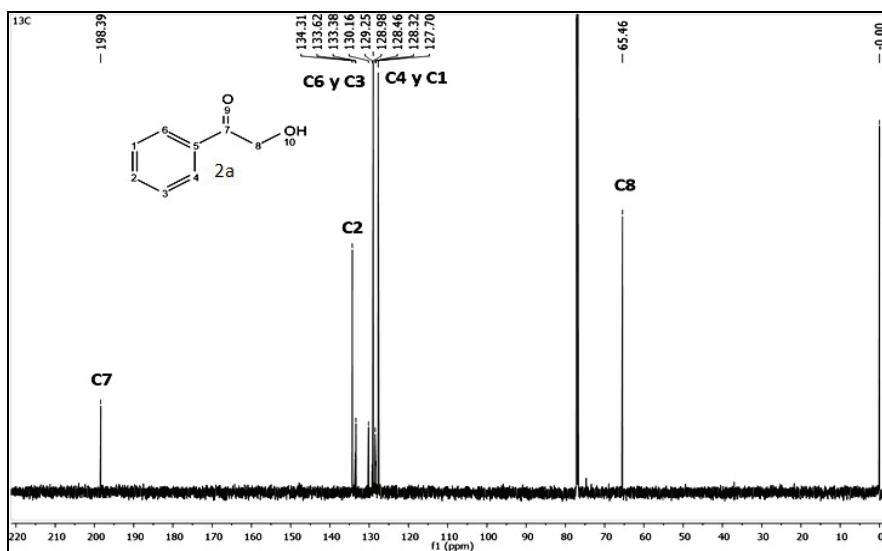


Fig. S-18.  $^{13}\text{C}$ -NMR spectrum of 2-hydroxy-1-phenylethanone (**2a**), in  $\text{CDCl}_3$ .

The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of 1-phenyl-1,2-ethanediol (**2b**)

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 2.3 (1H, s, OH), 3.2 (1H, s, OH), 3.49–3.78 (2H, m,  $\text{CH}_2$ ) 4.8 (1H, dd,  $J = 8.4$  & 3.3 Hz, CH), 7.09–7.48 (5H, m, CH).

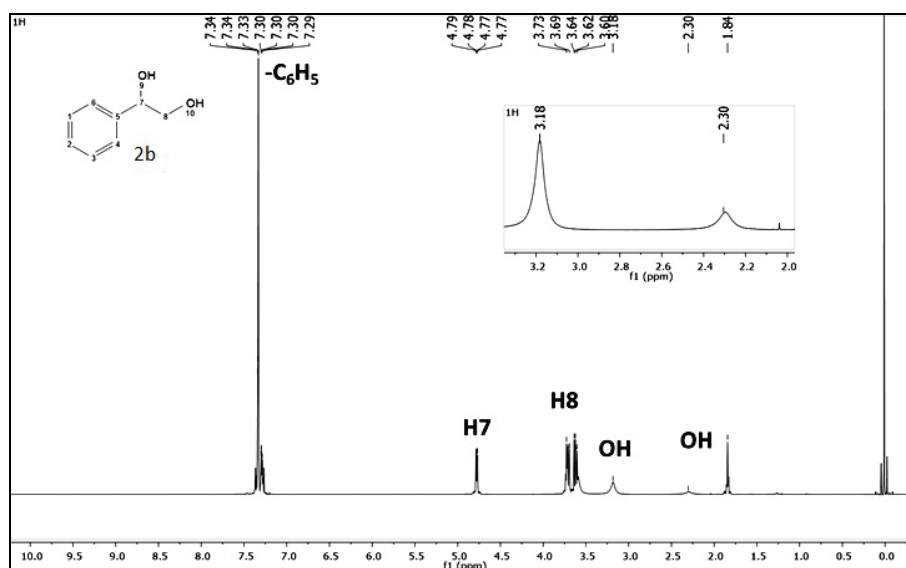


Fig. S-19.  $^1\text{H}$ -NMR spectrum of 1-phenyl-1,2-ethanediol (**2b**) in  $\text{CDCl}_3$ .

$^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$   $\delta$  / ppm.): 67.95/68.04 (C8), 74.7 (C7), 126.07 (C4 & C6), 127.9 & 128.5 (C1, 2 & C3), 140.5 (C5).

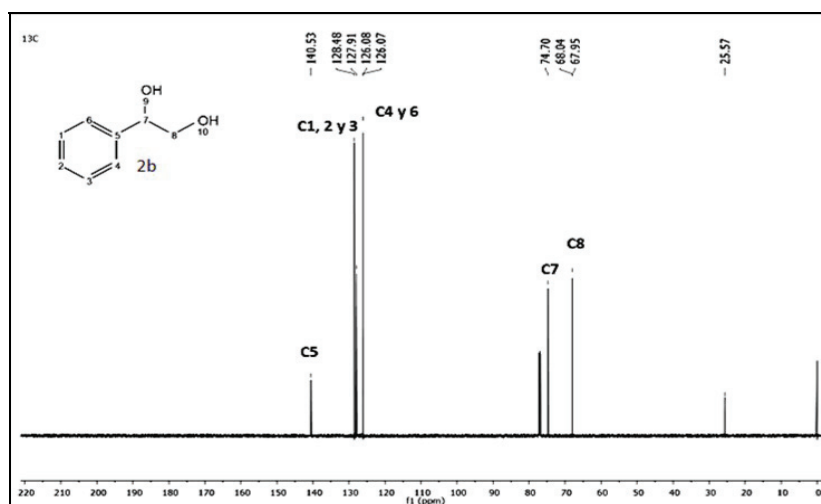


Fig. S-20.  $^{13}\text{C}$ -NMR spectrum of 1-phenyl-1,2-ethanediol (**2b**), in  $\text{CDCl}_3$ .

The GC chromatograms of **2a** and **2b** with retention times of 4.90 and 5.79 min, respectively, with conditions reported in the Experimental Section.

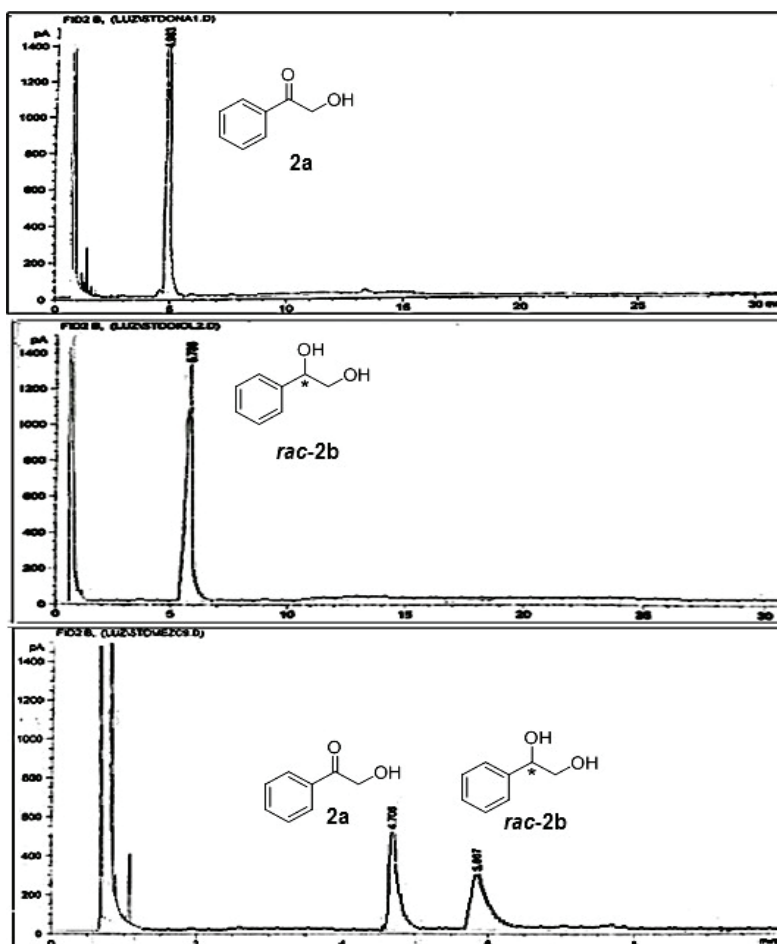


Fig. S-21. GC chromatograms of 2-hydroxy-1-phenylethanone (**2a**) and 1-phenyl-1,2-ethanediol (**2b**) and a mixture of **2a** and *rac-2b*.

The HPLC chromatograms of mixtures of **2a**, (*S*)-**2b** and (*R*)-**2b** with retention times of 17.12, 11.38 and 9.04 min, respectively, with conditions reported in the Experimental Section.

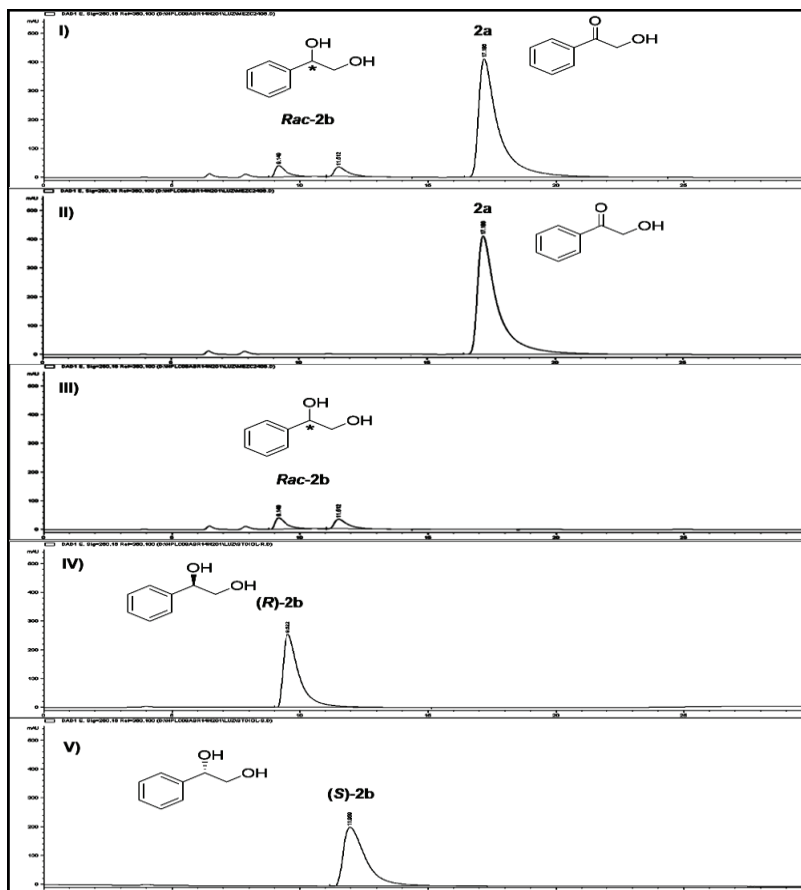


Fig. S-22. HPLC chromatogram of: I) mixture of 2-hydroxy-1-phenylethanone (**2a**) and *rac*-1-phenyl-1,2-ethanediol (*rac*-**2b**); II) 2-hydroxy-1-phenylethanone (**2a**); III) *rac*-1-phenyl-1,2-ethanediol (*rac*-**2b**); IV) (*R*)-1-phenyl-1,2-ethanediol (*R*-**2b**); V) (*S*)-1-phenyl-1,2-ethanediol (*S*-**2b**)

The HPLC chromatograms of mixtures of **2a**, (*S*)-**2b** and (*R*)-**2b** with retention times of 28.9, 18.9 and 15.1 min, respectively, under the conditions reported in the Experimental Section.

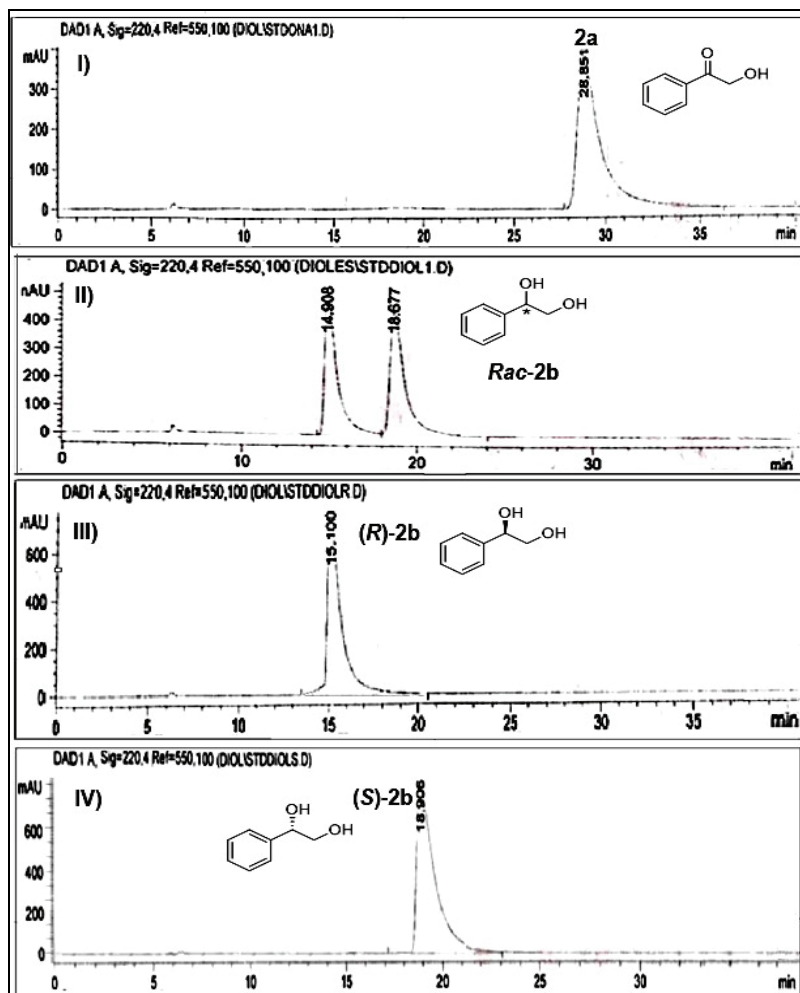
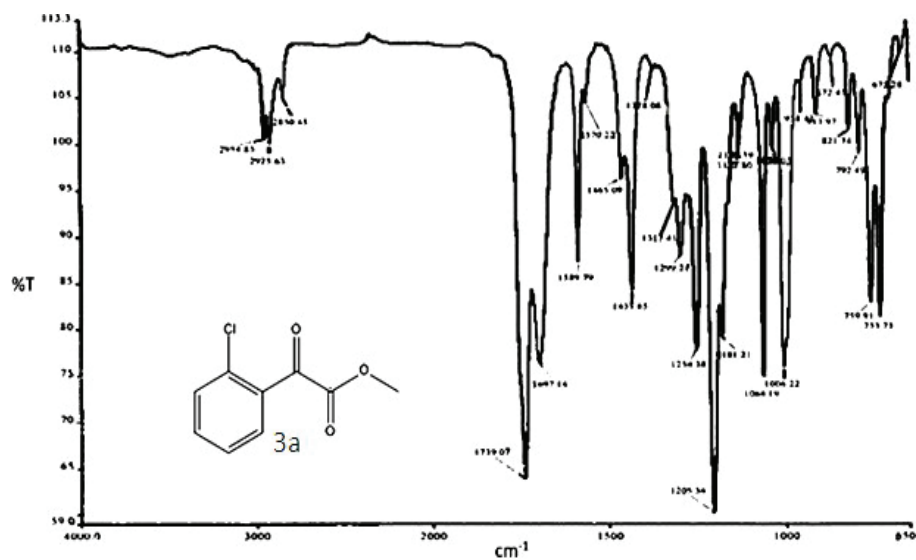
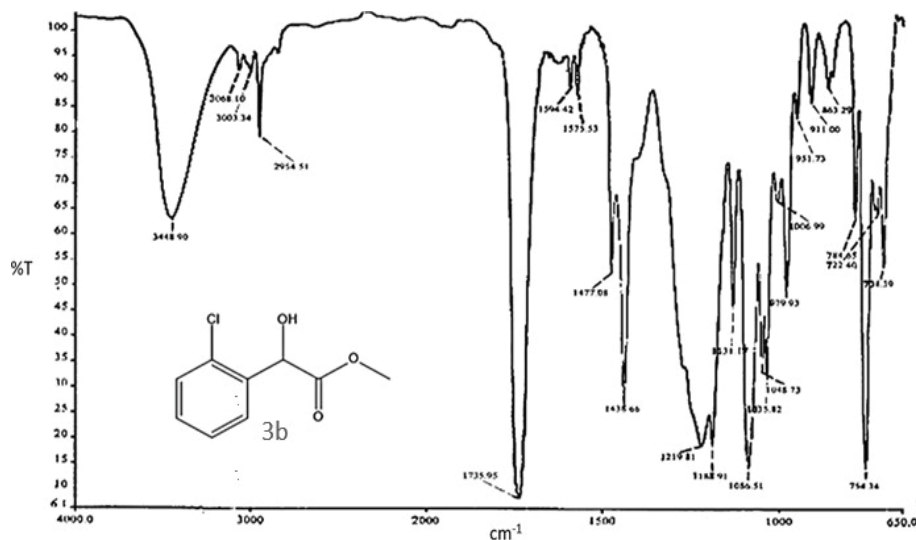


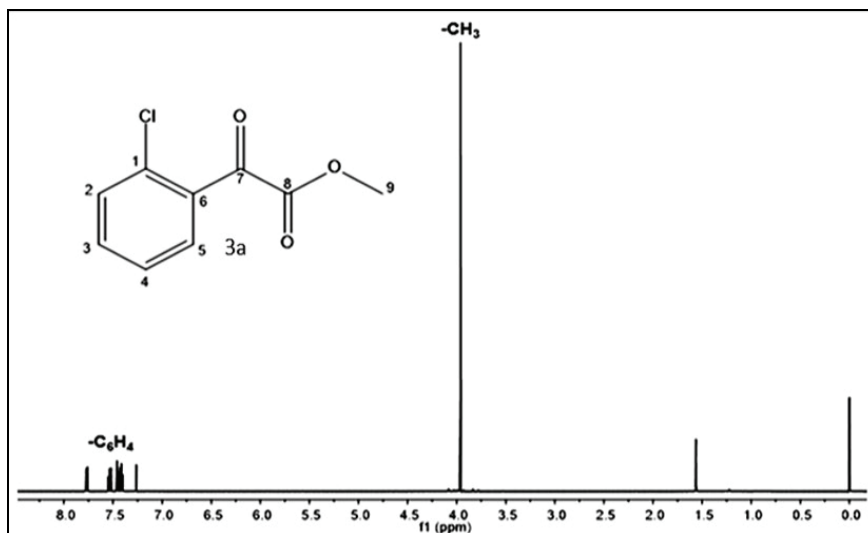
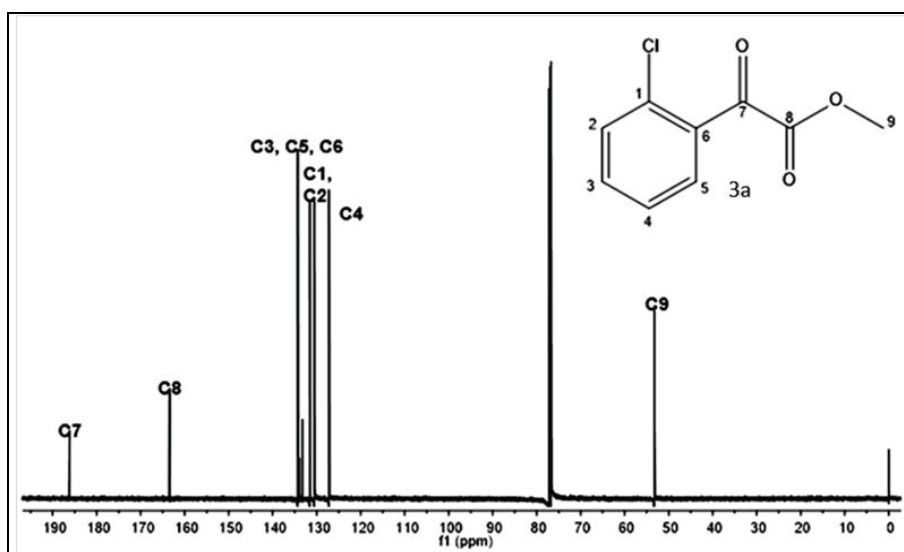
Fig. S-23. HPLC chromatogram of: I) 2-hydroxy-1-phenylethanone (**2a**); II) *rac*-1-phenyl-1,2-ethanediol (*rac-2b*); III) (*R*)-1-phenyl-1,2-ethanediol (*R-2b*); IV) (*S*)-1-phenyl-1,2-ethanediol (*(S)-2b*).

Fig. S-24. IR spectrum of methyl (2-chlorophenyl)(oxo)acetate (**3a**).Fig. S-25. IR spectrum of methyl (2-chlorophenyl)(hydroxy)acetate (**3b**).

The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of methyl (2-chlorophenyl)(oxo)acetate (**3a**)

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 3.96 (3H, s), 7.43 (2H, m), 7.53 (1H, ddd,  $J=8.0, 7.4$  &  $1.7$  Hz), 7.77 (1H, dd,  $J=7.7$  &  $1.7$  Hz).



Fig. S-26.  $^1\text{H}$ -NMR spectrum of methyl (2-chlorophenyl)(oxo)acetate (**3a**), in  $\text{CDCl}_3$ Fig. S-27.  $^{13}\text{C}$ -NMR spectrum of methyl (2-chlorophenyl)(oxo)acetate (**3a**), in  $\text{CDCl}_3$ .

$^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 53.2 (C9), 127.2 (C4), 130.5 (C2), 131.6 (C5), 133.2 (C6), 133.9 (C1), 134.3 (C3), 163.4 (C8), 186.2 (C7).

*The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of methyl (2-chlorophenyl)(hydroxy)acetate (**3b**)*

$^1\text{H}$ -NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 3.6 (1H, *d*,  $J = 5.1$  Hz, OH), 3.8 (3H, *s*,  $\text{CH}_3$ ), 5.6 (1H, *d*,  $J = 5.1$  Hz, CH) 7.3 (2H, *m*), 7.4 (2H, *m*).

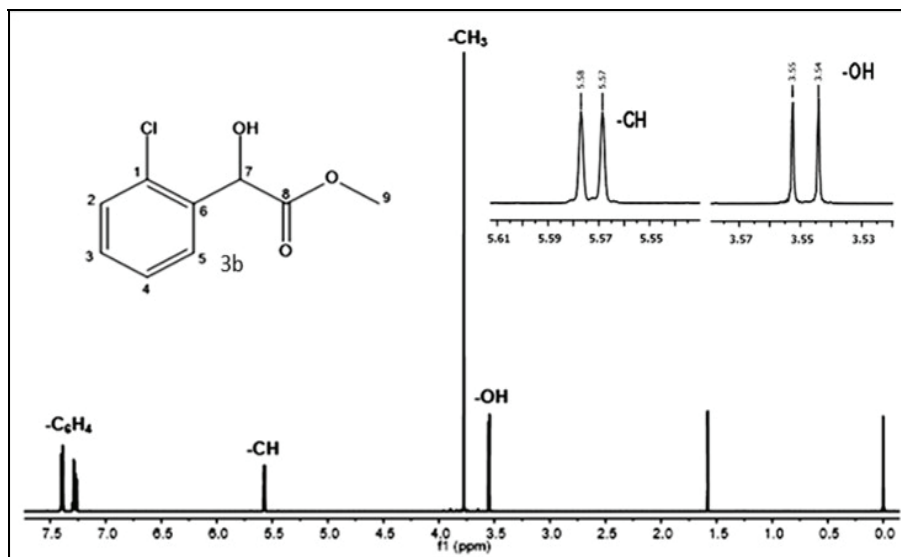


Fig. S-28.  $^1\text{H}$ -NMR spectrum of methyl (2-chlorophenyl)(hydroxy)acetate (**3b**), in  $\text{CDCl}_3$ .

$^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$ ,  $\delta$  / ppm): 53.2 (C9), 70.3 (C7), 127.1 (C4), 128.8 (C2), 129.7 (C5), 129.9 (C3), 133.41 (C1), 135.9 (C6), 173.7 (C8).

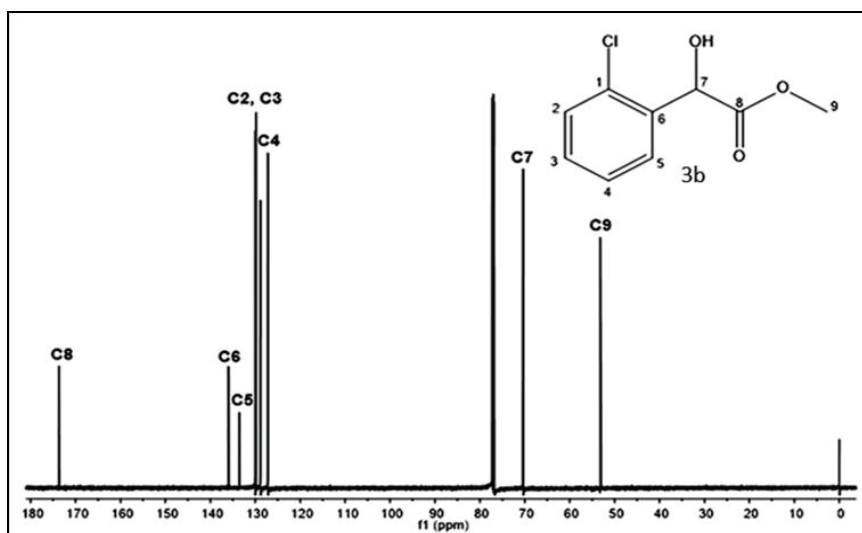


Fig. S-29.  $^{13}\text{C}$ -NMR spectrum of methyl (2-chlorophenyl)(hydroxy)acetate (**3b**), in  $\text{CDCl}_3$ .

The GC chromatograms of **3a** and **3b** with retention times of 4.6 and 6.1 min, respectively, under conditions reported in the Experimental Section.

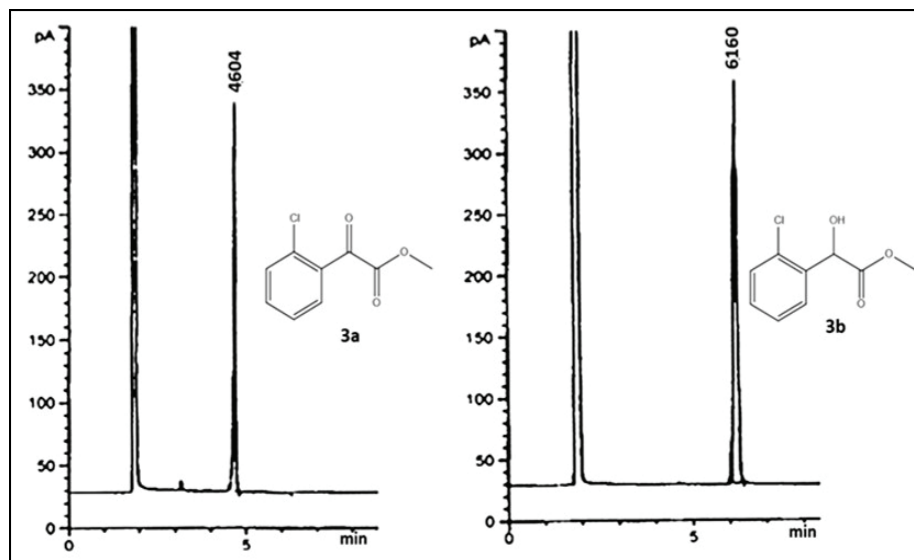


Fig. S-30. GC chromatograms of methyl (2-chlorophenyl)(oxo)acetate (**3a**) and methyl (2-chlorophenyl)(hydroxy)acetate (**3b**).

The HPLC chromatogram of the mixture of **3a**, (*S*)-**3b** and (*R*)-**3b** with retention times of 7.0, 8.9 and 10.2 min, respectively, under conditions reported in the Experimental Section.

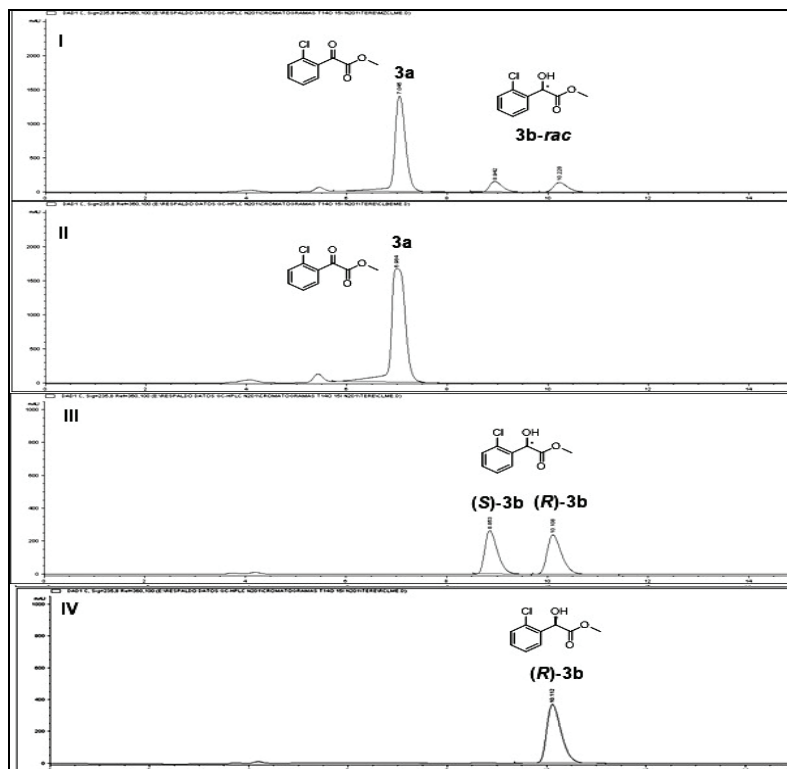


Fig. S-31. HPLC chromatogram of: I) mixture of methyl (2-chlorophenyl)oxoacetate (**3a**) and *rac*-methyl (2-chlorophenyl)(hydroxy)acetate (*rac*-**3b**); II) methyl (2-chlorophenyl)oxoacetate (**3a**); III) *rac*-methyl (2-chlorophenyl)(hydroxy)acetate (*rac*-**3b**) and IV) (*R*)-methyl (2-chlorophenyl)(hydroxy)acetate (*R*-**3b**).

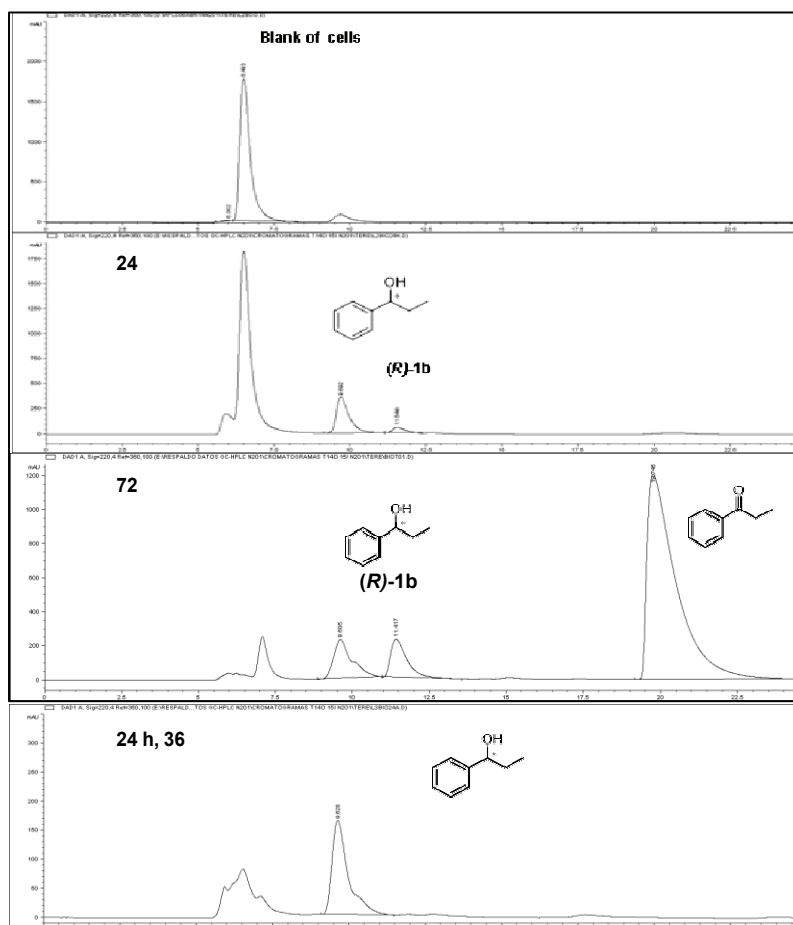


Fig. S-32. HPLC chromatogram of Fig. 2. Reduction of ketones **1a** with *N. corallina* biomass, final pH (5.36, 5.67 and 7.62).

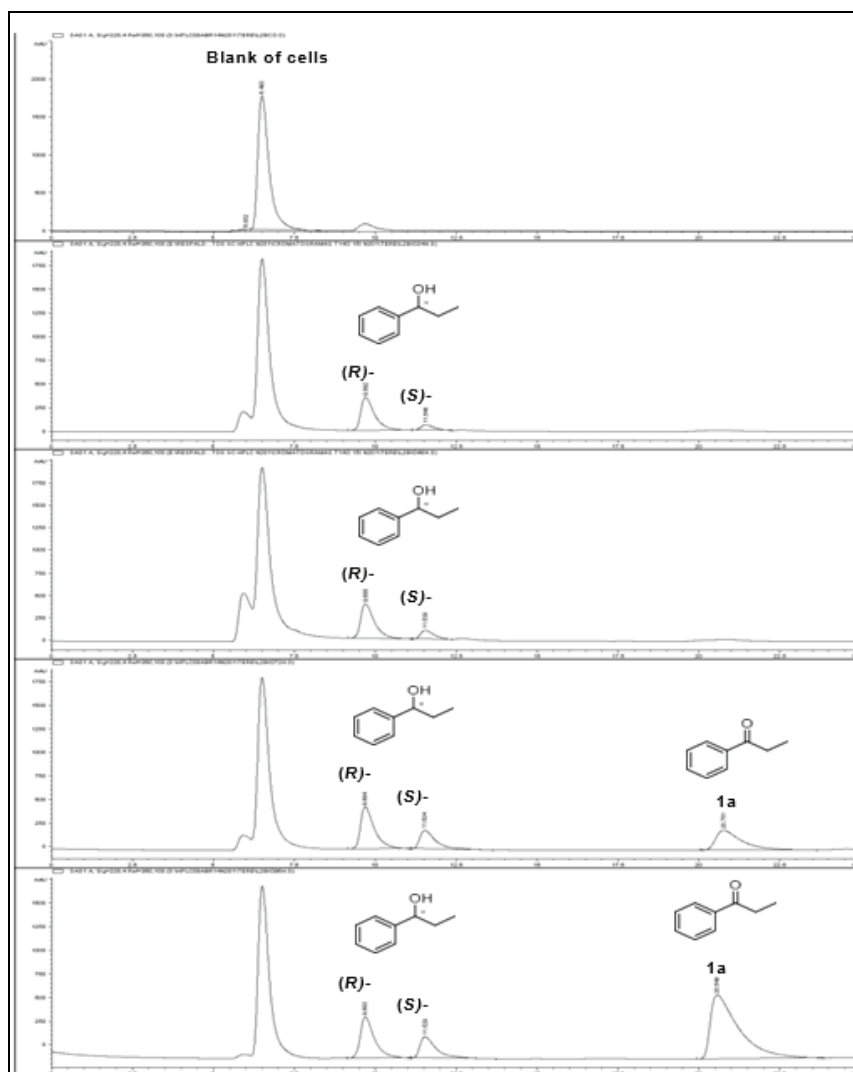


Fig. S-33. HPLC chromatogram of Fig. 3. Biotransformation of **1a** with *N. corallina* biomass, final pH 5.36.

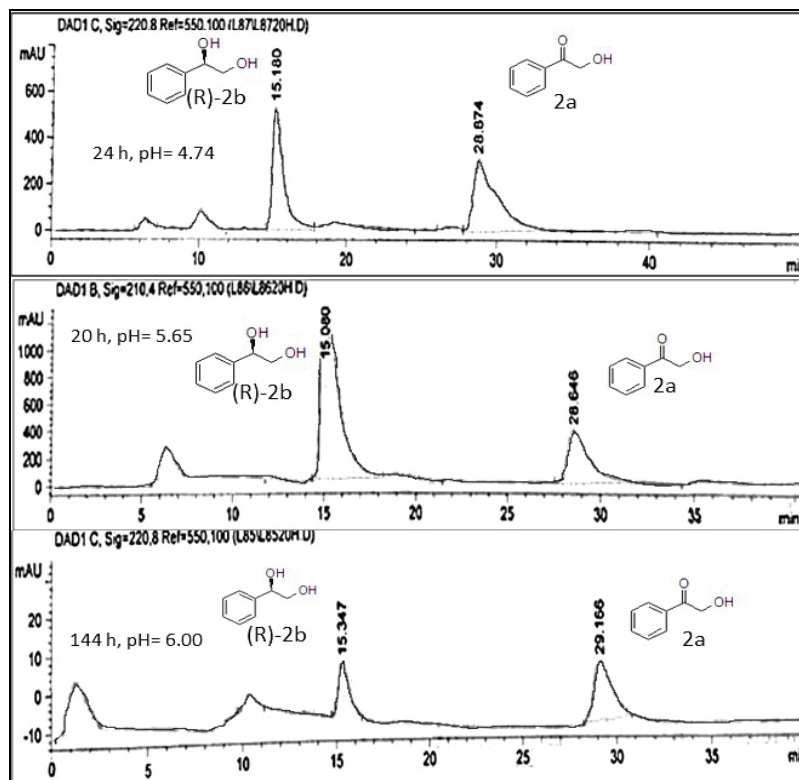


Fig. S-34. HPLC chromatogram of Fig. 4. Biotransformation of 2a with *N. corallina* biomass, for different final pH values and times.

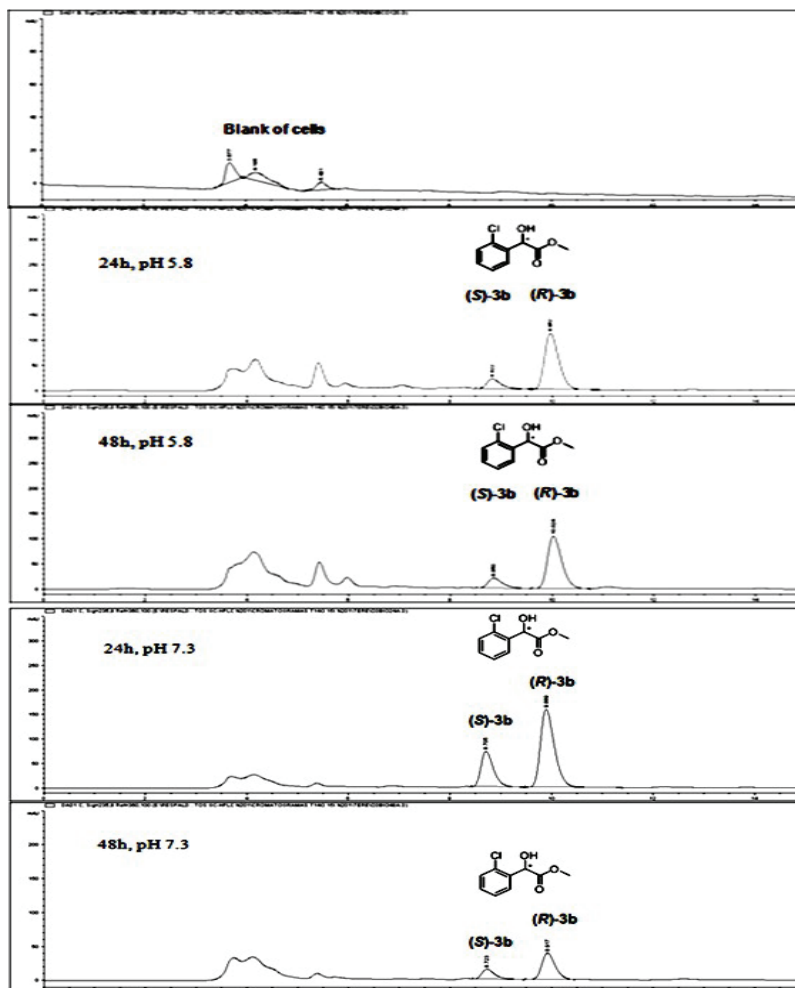


Fig. S-35. HPLC chromatogram of Fig. 5. Biotransformation of **3a** with *N. corallina* biomass, final pH 5.8 and 7.3.



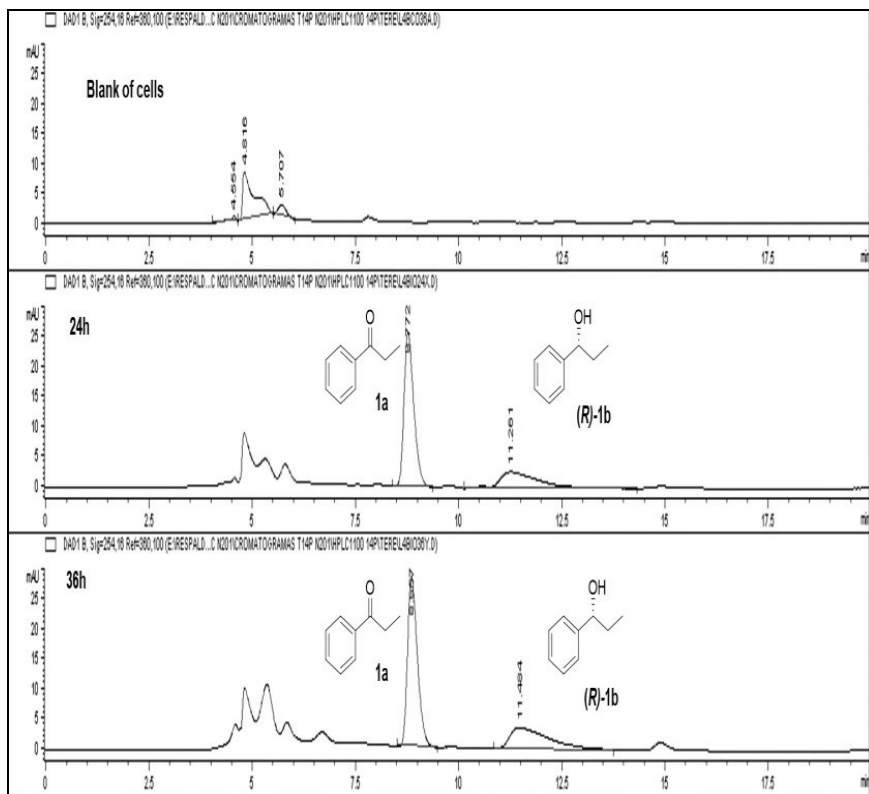


Fig. S-36. HPLC chromatogram of Fig. 6. Biotransformation of **1b** with *N. corallina* biomass, final pH 7.62.

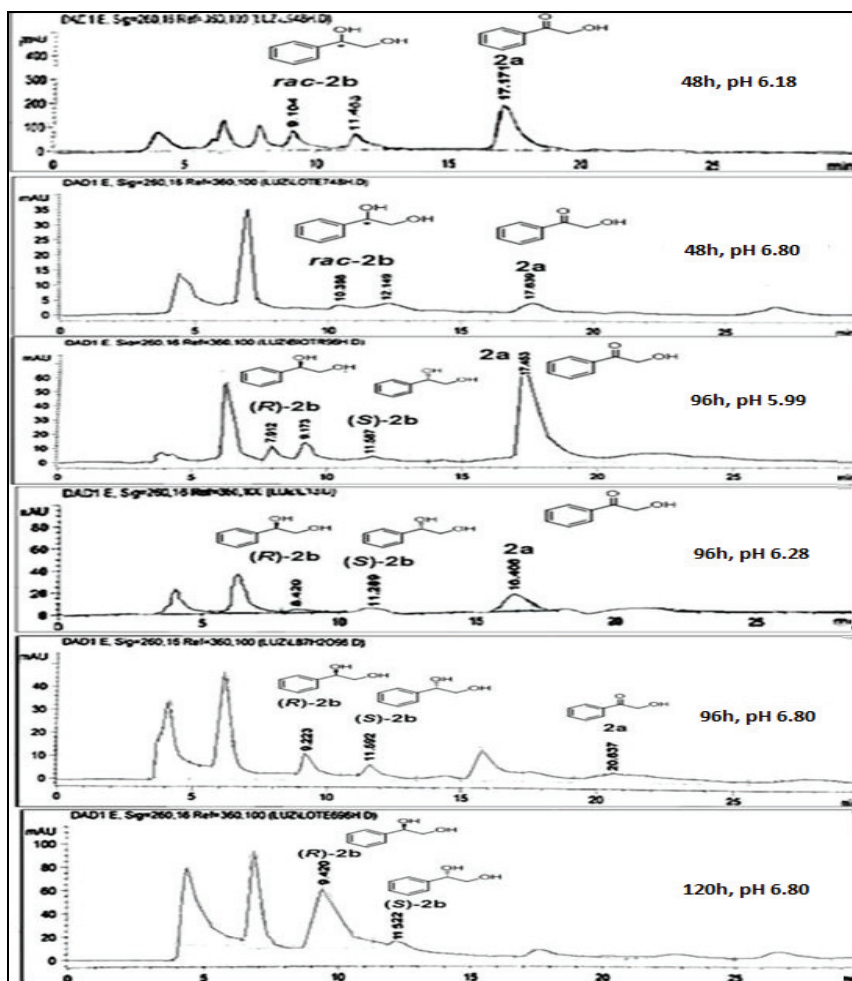


Fig. S-37. HPLC chromatogram of Fig. 7. Biotransformation of **2b** with *N. corallina* biomass, at different final pH values and times.

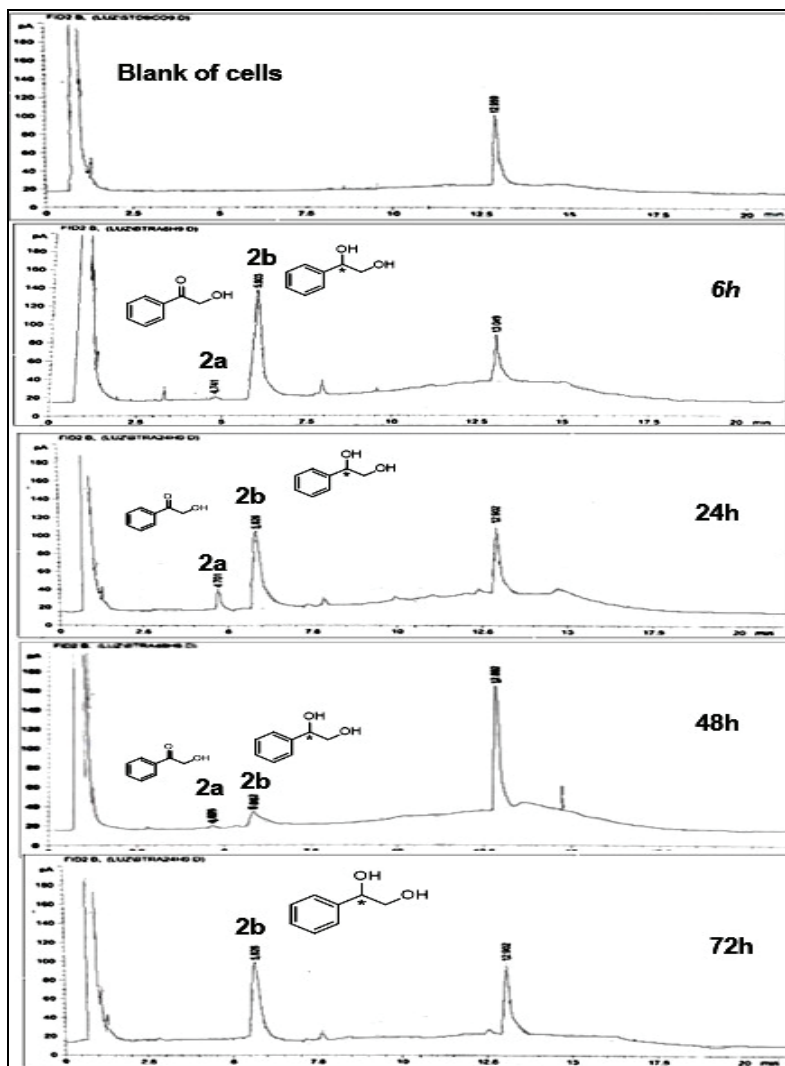


Fig. S-38. GC chromatogram of Fig. 8. Biotransformation of **2b** with *N. corallina* biomass, final pH 5.9.

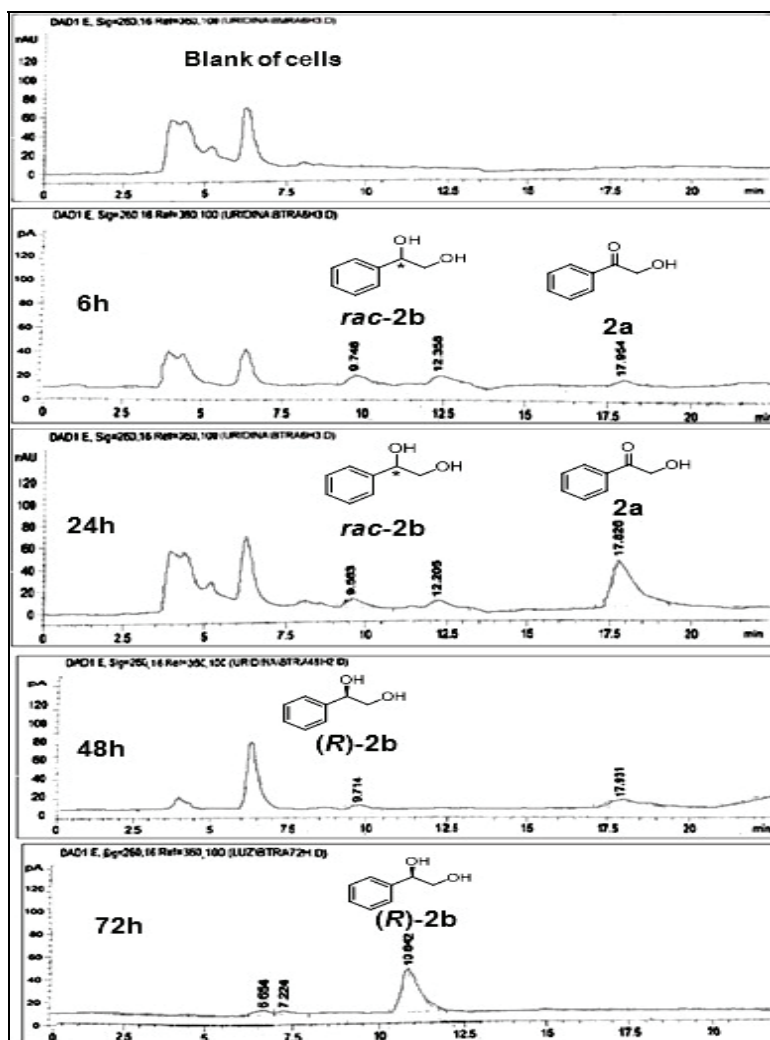


Fig. S-39. HPLC chromatogram of Fig. 8. Biotransformation of **2b** with *N. corallina* biomass, final pH 5.9.

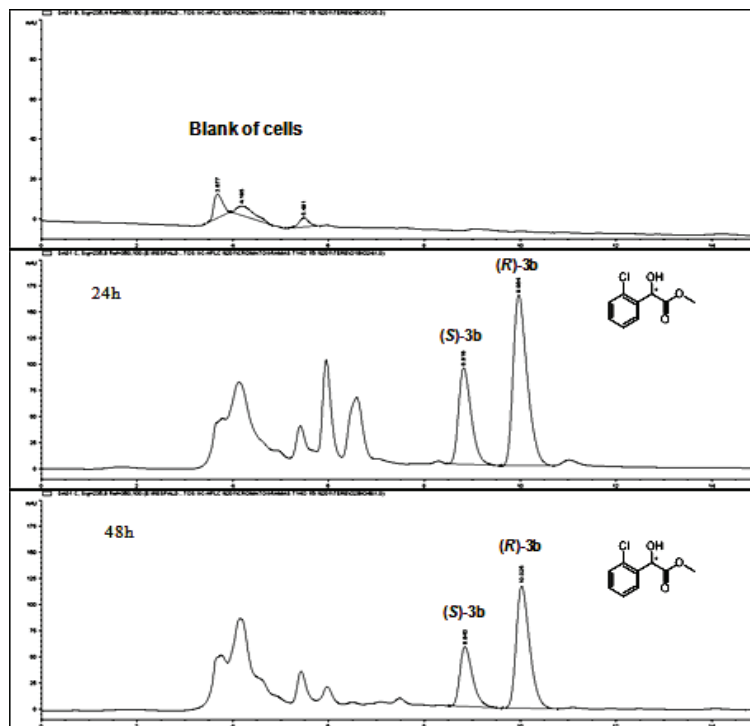


Fig. S-40. HPLC chromatogram of Fig. 9. Biotransformation of **3b** with *N. corallina* biomass, final pH 5.9.

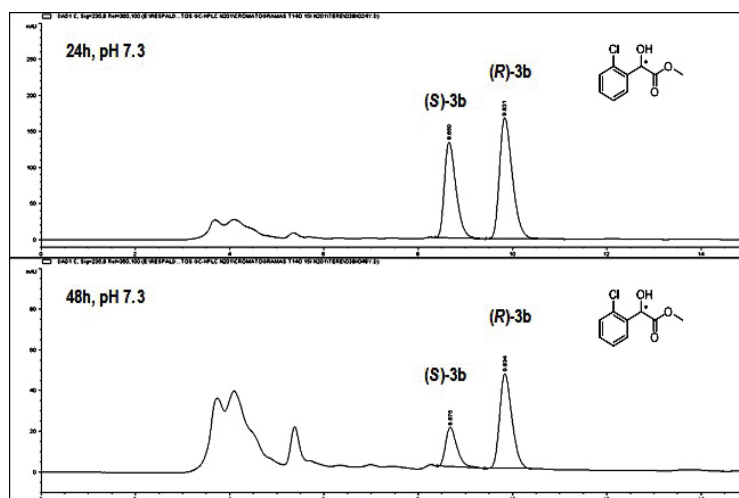


Fig. S-41. HPLC chromatogram of Fig. 9. Biotransformation of **3b** with *N. corallina* biomass, final pH 7.3.