

1

Supplementary material

2 **Influence of various cosolvents on the calcium oxide-catalyzed sunflower oil ethanolysis**

3 DUŠICA R. ĐOKIĆ-STOJANOVIĆ¹, ZORAN B. TODOROVIĆ², DRAGAN Z.

4 TROTER^{2*}, OLIVERA S. STAMENKOVIĆ² and VLADA B. VELJKOVIĆ²

5 *¹Zdravlj Actavis, Vlajkova 199, 16000 Leskovac, Serbia and ²Faculty of Technology,*

6 *University of Niš, Bulevar Oslobođenja 124, 16000 Leskovac, Serbia*

7 * Corresponding author. E-mail: drtroter@gmail.com

8

9

* Dragan Z. Troter. E-mail: drtroter@gmail.com

TABLE S-I. The properties of the employed cosolvents and ethanol

Cosolvent*	TEOA	DEOA	EG	MEK	HEX	TEA	Glyme	GLYC	THF	DIOX	ET
Chemical formula	C ₆ H ₁₅ NO ₃	C ₄ H ₁₁ NO ₂	C ₂ H ₆ O ₂	C ₄ H ₈ O	C ₆ H ₁₄	C ₆ H ₁₅ N	C ₄ H ₁₀ O ₂	C ₃ H ₈ O ₃	C ₄ H ₈ O	C ₄ H ₈ O ₂	C ₂ H ₆ O
Boiling point (°C)**	335.0	268.0	197.6	79.6	68.0	89.7	83.5	290.0	65.0	101.1	78.5
Melting point (°C)**	21.5	28.0	-13.0	-86.0	-95.0	-115.0	-58.0	19.0	-108.3	11.8	-114.1
Density (g/cm ³)											
at 25°C	1.12 ^b	—	1.11 ^b	0.799 ^b	0.656 ^b	0.724 ^b	0.865 ^b	1.257 ^b	0.88 ^b	1.029 ^b	0.787 ^b
at 20°C	1.1242 ^{c,e}	1.0966 ^{c,e}	1.1088 ^{c,e}	0.8054 ^c	0.6593 ^e	0.7275 ^c	0.8691 ^c	1.261 ^c	0.8892 ^c	1.0337 ^c	0.7893 ^c
Viscosity at 25°C (mPas)	652.576 ^b ; 609 ^c	109.5 (at 50 °C)	17.645 ^b ; 16.1 ^c	0.396 ^b ; 0.405 ^c	0.296 ^b ; 0.300 ^c	0.341 ^b ; 0.347 ^c	0.446 ^b	749.3 ^b ; 934 ^c	0.465 ^b ; 0.456 ^c	1.211 ^b ; 1.177 ^c	1.057 ^b ; 1.074 ^c
Partition coefficients	- 1.00 ^b ; - 1.43	- 1.36 ^{b,e}	0.29 ^{a,b,c,e,f}	3.90 ^{b,e}	1.45 ^{a,b,c} ; - 0.21 ^b	- 1.76 ^b	0.46	- 0.27 ^{b,e}	- 0.31 ^{b,e}	- 0.49 ^f	- 0.30 ^{a,c}
at 25°C (logP)	- 1.59 ^e ; (20 °C) ^{b,e}			4.0 ^{a,c,f}	1.64 ^e			(20 °C) ^a			
				- 2.3 ^f							
Refractive index at 25°C	1.4835 ^b	1.4747	1.4306 ^b	1.3764 ^b	1.3723 ^b	1.3980 ^b	1.3781 ^b ; 1.3770 ^c	1.4730 ^b	1.405 ^{b,c}	1.4202 ^b	1.3594 ^b
				(19 °C) ^b							
Dipole moment, D	3.57 ^{c,d} ; 3.48 ^f	0.85 ^b ; 2.8 ^c	2.31 ^b ; 2.36 ^c	2.76 ^b ; 2.779 ^c	0.00 ^b ; 0.08 ^f	0.66 ^{b,c}	1.71 ^b	4.21 ^b ; 2.56 ^c	1.63 ^{b,f} ; 1.75 ^c	0.00 ^{b,c} ; 0.45 ^f	1.69 ^{b,c}
				2.28 ^c	2.78 ^f						
Dielectric constant (20 °C)	29.36 (25 °C) ^{c,d}	25.75 ^c	41.4 ^c	18.56 ^c	1.8865 ^c	2.418 ^c	7.30 (23.7 °C) ^c	46.53 ^c	7.52 ^c	2.2189 ^c	25.3 ^c

*TEOA - triethanolamine, DEOA - diethanolamine, EG - ethylene glycol, MEK - methyl ethyl ketone, HEX - *n*-hexane, TEA - triethylamine, Glyme - ethylene glycol dimethyl ether, GLYC - glycerol, THF - tetrahydrofuran, DIOX - dioxane and ET - ethanol. **According to Material Safety Data Sheet. ^aref [1]; ^bref [2]; ^cref [3]; ^dref [4]; ^eref [5]; ^fref [6].

REFERENCES

1. J. Sangster, *J. Phys. Chem. Ref. Data* **18** (1989) 1111 (<https://doi.org/10.1063/1.555833>)
2. C. L. Yaws, *Chemical properties handbook: physical, thermodynamic, environmental, transport, safety, and health related properties for organic and inorganic chemicals*, McGraw-Hill, New York, USA, 1999
3. D. R. Lide, *CRC Handbook of Chemistry and Physics*, 84th ed., CRC Press LLC; 2003
4. J. A. Dean, *Lange's handbook of chemistry*, 15th ed., McGraw-Hill, New York, USA, 1999
5. D. Mackay, W. Y. Shiu, K.-C. Ma, S. C. Lee, *Handbook of physical-chemical properties and environmental fate for organic chemicals*, 2nd ed., CRC Press, Boca Raton, USA, 2006
6. Z. B. Todorović, O. S. Stamenković, I. S. Stamenković, J. M. Avramović, A. V. Veličković, I. B. Banković-Ilić, V. B. Veljković, *Fuel* **107** (2013) 493
(<https://doi.org/10.1016/j.fuel.2012.11.049>)