



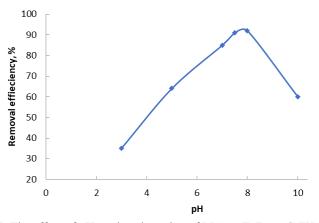
J. Serb. Chem. Soc. 84 (7) S252-S253 (2019)

JSCS-info@shd.org.rs • www.shd.org.rs/JSCS Supplementary material

## SUPPLEMENTARY MATERIAL TO Efficient removal of Malachite Green from aqueous solution by adsorption on carbon nanotubes modified with ZnFe<sub>2</sub>O<sub>4</sub> nanoparticles

SHIVA DEHGHAN ABKENAR<sup>1\*</sup>, MORASSA HASSANNEZHAD<sup>2</sup>, MORTEZA HOSSEINI<sup>3</sup> and MOHAMMAD REZA GANJALI<sup>2,4\*\*</sup>

<sup>1</sup>Department of Chemistry, Savadkooh Branch, Islamic Azad University, Savadkooh, Iran, <sup>2</sup>Center of Excellence in Electrochemistry, Faculty of Chemistry, University of Tehran, Tehran, Iran, <sup>3</sup>Department of Life Science Engineering, Faculty of New Sciences & Technologies, University of Tehran, Tehran, Iran and <sup>4</sup>BiosensorResearch Center, Endocrinology and Metabolism Molecular-Cellular Sciences Institute, Tehran University of Medical Sciences, Tehran, Iran



J. Serb. Chem. Soc. 84 (7) (2019) 701-712

Fig. S-1. The effect of pH on the adsorption of MG on  $ZnFe_2O_4/MWCNTs.$ 

<sup>\*\*\*\*</sup> Corresponding authors. E-mail: (\*)dehghan54@yahoo.com; (\*\*)ganjali@gmail.com

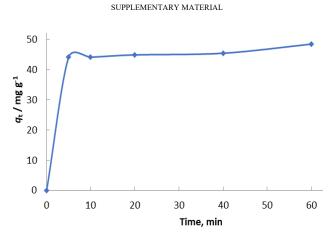


Fig. S-2. The effect of contact time on the adsorption of MG on  $ZnFe_2O_4/MWCNTs$ .

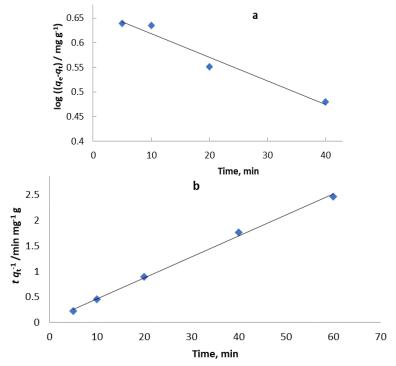


Fig. S-3. The pseudo-first-order kinetics (a) and the pseudo-second-order kinetics (b) for the adsorption MG on  $ZnFe_2O_4/MWCNTs.$ 

## S253