



SUPPLEMENTARY MATERIAL TO  
**Quantitative structure–property relationship studies for the  
prediction of the vapor pressure of volatile organic compounds**

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TABLE S-I. The data set and the corresponding experimental and predicted values of  
log ( $p_v$ /Pa) for the training and validation sets

ID	Name	log ( $p_v$ /Pa) (Exp)	log ( $p_v$ /Pa) (Pred)	$X0_{sol}$	$SpPosA\_H2$	$GATS2e$	$Hy$	$Hat (h_{ii})$	$\epsilon_{istd}$
1	1,1,2-Trichloroethane	3.48	3.3841	5.784	0.541	1.078	-0.359	0.105	-0.4647
2	Iodomethane	4.72	4.7638	3.5	0.5	0.97	-0.315	0.101	0.211
3	Iodoethane	4.12	4.631	4.207	0.515	1.071	-0.528	0.072	2.3469
4	1-Bromobutane	3.72	3.63	5.121	0.57	0.95	-0.719	0.03	-0.3867
5	Hexan-2-one	2.71	3.1166	5.699	0.56	0.534	-0.802	0.118	2.014
6	2-Methylheptane	3.41	3.662	6.406	0.576	1.317	-0.946	0.108	1.2283
7	2,3,4-Trimethylpentane	3.53	3.6423	6.732	0.548	1.128	-0.946	0.145	0.5827
8	1,1-Dichloroethane	4.47	4.1496	4.577	0.5	0.74	-0.431	0.1	-1.5394
9	Ethylcyclohexane	3.12	3.264	5.82	0.633	1.348	-0.946	0.136	0.7362
10	1-Propoxypropane	3.95	4.1052	5.536	0.588	1.641	-0.802	0.163	0.8319
11	1,2-Dichloroethene	4.63	4.188	4.414	0.55	1.25	-0.431	0.054	-1.9698
12	2-Bromopropane	4.42	4.4842	4.577	0.5	0.801	-0.646	0.12	0.3188
13	3,3-Dimethylpentane	4.02	3.9226	5.914	0.55	1.076	-0.936	0.073	-0.4478
14	Toluene	3.42	3.2118	5.113	0.632	1.016	-0.936	0.132	-1.0569
15	1,4-Xylene	3.12	2.9552	5.983	0.61	0.893	-0.946	0.085	-0.7732
16	Ethylbenzène	3.12	2.9638	5.82	0.633	1.084	-0.946	0.118	-0.7736
17	1,2-Xylene	2.42	2.9182	5.983	0.613	0.893	-0.946	0.09	2.3562
18	Ethanol	3.87	3.8322	2.707	0.515	1.026	0.638	0.261	-0.2442
19	Methylcyclopentane	4.25	4.2475	4.406	0.599	1.299	-0.921	0.109	-0.0122
20	2-Methyl-2-propanol	3.7	3.6991	4.5	0.482	0.728	0.132	0.173	-0.0051
21	2-Propylenitrile	4.15	4.0264	3.414	0.55	0.47	-0.646	0.23	-0.751
22	Pent-1-ene	4.96	4.7238	4.121	0.57	1.313	-0.898	0.111	-1.1554
23	Dichloromethane	4.75	4.7186	3.707	0.515	1.232	-0.264	0.094	-0.1493

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TABLE S-I. Continued

ID	Name	$\log(p_v/\text{Pa})$ (Exp)	$\log(p_v/\text{Pa})$ (Pred)	$X_{0sol}$	$SpPosA\_H2$	$GATS2e$	$Hy$	$Hat(h_{ii})$	$\epsilon_{istd.}$
24	Butyronitrile	3.12	3.4784	4.121	0.57	0.412	-0.719	0.211	2.099
25	Propan-1-ol	3.41	3.384	3.414	0.55	0.956	0.323	0.182	-0.144
26	Butan-1-ol	2.82	3.0135	4.121	0.57	0.926	0.132	0.158	1.0282
27	2-Methylpropan-1-ol	3.12	3.2801	4.284	0.541	0.912	0.132	0.118	0.7935
28	1,2-Dichlorobenzene	2.13	1.8814	6.983	0.613	0.642	-0.71	0.232	-1.5148
29	2,2,4-Trimethylpentane	3.79	3.7334	6.784	0.53	1.034	-0.946	0.185	-0.3156
30	Pentan-2-one	3.31	3.4448	4.992	0.556	0.531	-0.767	0.115	0.6638
31	2,4-Dimethylpentane	4.1	4.1299	5.862	0.545	1.183	-0.936	0.089	0.141
32	(1-Methylethyl)benzene	2.82	2.6799	6.69	0.619	1.026	-0.954	0.124	-0.7011
33	Propan-2-ol	3.75	3.8153	3.577	0.5	0.859	0.323	0.15	0.3419
34	1-Chloropropane	4.66	4.324	3.914	0.55	0.934	-0.646	0.069	-1.5356
35	2,2-Dimethylbutane	4.63	4.4817	5.207	0.523	1.005	-0.921	0.104	-0.717
36	pentamethylene	4.63	4.907	3.536	0.585	1.4	-0.898	0.195	1.5733
37	2-Methyl-2-butene	4.72	4.6084	4.284	0.541	0.963	-0.898	0.097	-0.5337
38	propyl formate	4.02	3.5854	4.828	0.57	0.906	-0.614	0.034	-1.8789
39	Diethyl ether	4.87	4.9728	4.121	0.57	1.726	-0.719	0.211	0.6022
40*	Dibromomethane	3.76	4.3039	4.707	0.515	1.272	-0.264	0.107	2.3622
41*	1,3-Dichlorobenzene	2.12	1.8198	6.983	0.618	0.642	-0.71	0.237	-1.4105
42*	3-Pentanone	3.33	3.3708	4.992	0.562	0.531	-0.767	0.113	0.1776
43*	Ethyl acetate	4.12	3.5608	4.992	0.556	0.799	-0.614	0.04	-2.342
44*	2-Butanone	4.11	3.8998	4.284	0.541	0.534	-0.719	0.136	-0.9284
45*	Hexane	4.29	4.5053	4.828	0.57	1.382	-0.921	0.085	0.9237
46*	Pentane	4.85	4.7773	4.121	0.57	1.36	-0.898	0.116	-0.3174
47*	1,2-Dichloropropane	3.83	3.7066	5.284	0.541	0.964	-0.539	0.042	-0.5175
48*	Heptane	3.76	3.9942	5.536	0.588	1.398	-0.936	0.082	1.0032
49*	1,2-dichloroethane	4.03	4.0937	4.414	0.55	1.167	-0.431	0.041	0.2668
50*	1-Bromopropane	4.12	4.1189	4.414	0.55	0.956	-0.646	0.039	-0.0048
51*	Ethyl acetate	2.42	2.8565	5.983	0.618	0.893	-0.946	0.098	1.886

\*Test compounds