

SUPPLEMENTARY MATERIAL TO

First cobalt complexes with methyl pyruvate semi/thiosemicarbazone – synthesis, physico-chemical and structural characterization

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ANALYTICAL AND SPECTRAL DATA FOR 1–4

[Co(Hmps)(H₂O)Cl₂] (1). Yield: 52 mg (85 %). Anal. Calc. for CoC₅H₁₁N₃O₄Cl₂: C, 19.56; H, 3.59; N, 13.69. Found: C, 19.29; H, 3.49; N, 13.35%. Conductivity [$\Lambda_M/\Omega^{-1} \text{ cm}^2 \text{ mol}^{-1}$]: 140 (in MeOH). $\mu_{\text{eff}} = 5.02 \mu_B$. Selected IR bands [$\tilde{\nu} / \text{cm}^{-1}$]: $\nu(\text{OH}, \text{NH}_2, \text{NH})$: 3414vs, 3305s, 3231ms, 3124ms; $\nu(\text{C}=\text{O})$: 1683vs, 1635s; $\nu(\text{C}=\text{N})$: 1583m.

[Co(Hmps)(H₂O)Br₂] (2). Yield: 90 mg (57 %). Anal. Calc. for CoC₅H₁₁N₃O₄Br₂: C, 15.17; H, 2.78; N, 10.61. Found: C, 15.08; H, 2.57; N, 10.55. Conductivity [$\Lambda_M/\Omega^{-1} \text{ cm}^2 \text{ mol}^{-1}$]: 170 (in MeOH). $\mu_{\text{eff}} = 4.98 \mu_B$. Selected IR bands [$\tilde{\nu} / \text{cm}^{-1}$]: $\nu(\text{OH}, \text{NH}_2, \text{NH})$: 3408vs, 3305vs, 3228s, 3126s; $\nu(\text{C}=\text{O})$: 1685vs, 1634vs; $\nu(\text{C}=\text{N})$: 1584m.

[Co(Hmpt)₂][CoCl₄]·2H₂O (3). Yield: 50 mg (83 %). Anal. Calc. for Co₂C₁₀H₂₂N₆O₆S₂Cl₄: C, 18.59; H, 3.43; N, 13.01; S, 9.92. Found: C, 18.21; H, 3.49; N, 12.87; S, 9.36%. Conductivity [$\Lambda_M/\Omega^{-1} \text{ cm}^2 \text{ mol}^{-1}$]: 222 (in MeOH). $\mu_{\text{eff}} = 4.41 \mu_B$. Selected IR bands

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25 $[\tilde{\nu}/\text{cm}^{-1}]$: $\nu(\text{OH})$: 3400ms, $\nu(\text{NH}_2, \text{NH})$: 3338s, 3261s, 3150vs; $\nu(\text{C=O})$: 1681vs; $\nu(\text{C=N})$:
 26 1621vs, 1605vs; $\nu(\text{C=S})$: 958w, 855ms.

27 $[\text{Co}(\text{Hmpt})_2]\text{Br}_2 \cdot \text{Me}_2\text{CO}$ (**4**). Yield: 68 mg (55 %). Anal. Calc. for $\text{CoC}_{13}\text{H}_{24}\text{N}_6\text{O}_5\text{S}_2\text{Br}_2$:
 28 C, 24.89; H, 3.83; N, 13.40; S, 10.22. Found: C, 24.68; H, 3.57; N, 13.55; S, 10.76%.
 29 Conductivity $[\Lambda_{\text{M}}/\Omega^{-1} \text{cm}^2 \text{mol}^{-1}]$: 180 (in MeOH). $\mu_{\text{eff}} = 4.36 \mu_{\text{B}}$. Selected IR bands $[\tilde{\nu}/\text{cm}^{-1}]$:
 30 $\nu(\text{NH}_2, \text{NH})$: 3431ms, 3261ms, 3112s; $\nu(\text{C=O})$: 1703m, 1672s; $\nu(\text{C=N})$: 1624s, 1606s; $\nu(\text{C=S})$:
 31 958w, 857w.

32 Table S-I. Pertinent crystal and refinement details for **2**, **3**, and **4**.

	2	3	4
Chemical formula	$\text{C}_5\text{H}_{11}\text{Br}_2\text{CoN}_3\text{O}_4$	$\text{C}_{10}\text{H}_{22}\text{Co}_2\text{Cl}_4\text{N}_6\text{O}_6\text{S}_2$	$\text{C}_{13}\text{H}_{24}\text{Br}_2\text{CoN}_6\text{O}_5\text{S}_2$
M_{r}	395.92	646.11	627.23
Crystal system	Monoclinic	Orthorhombic	Triclinic
Space group	$P2_1/c$	$Pccn$	$P\bar{1}$
Temperature, K	294	294	294
$a/\text{\AA}$	7.8004(2)	14.8546(3)	10.3796(3)
$b/\text{\AA}$	13.1957(4)	19.6095(4)	10.9107(4)
$c/\text{\AA}$	11.9046(3)	18.0338(5)	12.5905(5)
$\alpha/^\circ$	90	90	101.435(3)
$\beta/^\circ$	94.357(3)	90	101.558(3)
$\gamma/^\circ$	90	90	115.576(3)
$V/\text{\AA}^3$	1221.82(6)	5253.1 (2)	1192.86(8)
Z	4	8	2
Radiation type	Mo $K\alpha$	Mo $K\alpha$	Mo $K\alpha$
μ/mm^{-1}	7.94	1.86	4.28
Crystal size, mm	$0.59 \times 0.24 \times 0.19$	$0.53 \times 0.22 \times 0.07$	$0.39 \times 0.25 \times 0.05$
Absorption correction	Analytical	Multi-scan	Multi-scan
$T_{\text{min}}, T_{\text{max}}$	0.037, 0.310	0.832, 1	0.465, 1
Measured reflections	13155	21982	18787
Independent reflections	2501	6243	5636
Observed reflections $[I > 2\sigma(I)]$	2213	4586	4619
R_{int}	0.026	0.031	0.024
$(\sin \theta/\lambda)_{\text{max}}/\text{\AA}^{-1}$	0.626	0.683	0.685
$R[F^2 > 2\sigma(F^2)]$	0.025	0.037	0.027
$wR(F^2)$	0.057	0.092	0.063
S	1.09	1.02	1.02
Parameters	153	257	286
Restraints	6	0	6
H-atom treatment	Mixed	Constrained	Mixed
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}} (\text{e \AA}^{-3})$	0.37, -0.40	0.64, -0.48	0.50, -0.51

34 TABLE S-II. Cremer & Pople puckering parameters for **3** and **4**

Ring	$Q_2 / \text{\AA}$	$\varphi_2 / ^\circ$	Pucker descriptor
3			
Co1–S1A–C3A–N2A–N1A	0.1979(16)	9.2(7)	Twisted on Co1–S1A
Co1–S1B–C3B–N2B–N1B	0.1230(18)	28.2(10)	Envelope on S1B
4			
Co1–S1A–C3A–N2A–N1A	0.3725(14)	7.2(3)	Envelope on Co1
Co1–S1B–C3B–N2B–N1B	0.1578(15)	19.7(7)	Twisted on Co1–S1B

35 TABLE S-III. Polyhedral distortion indices for **2–4**

	2	3	4
Average bond length / $\text{\AA}$	2.2766	2.2198	2.2149
Polyhedral volume / $\text{\AA}^3$	14.9428	13.7791	13.3587
Distortion index	0.07207	0.05141	0.05276
Quadratic elongation / $(^\circ)^2$	1.0415	1.0420	1.0591
Bond angle variance	112.4344	126.9278	175.4050

36 Distortion index, $D = \frac{1}{n} \sum_{i=1}^n \frac{|l_i - l_{av}|}{l_{av}}$, where l_i is the distance from the central atom to the i th coordinating atom, and
37 l_{av} is the average bond length.
38 Quadratic elongation, $\langle \lambda \rangle = \frac{1}{n} \sum_{i=1}^n \left(\frac{l_i}{l_0} \right)^2$, where l_i is the distance from the central atom to the i th coordinating atom,
39 and l_0 is the center-to-vertex distance of a regular polyhedron of the same volume.
40 Bond angle variance, $\sigma^2 = \frac{1}{m-1} \sum_{i=1}^m (\phi_i - \phi_0)^2$, where m is the number of bond angles within the polyhedron, ϕ_i is
41 the i th bond angle, and ϕ_0 is the ideal bond angle for a regular polyhedron (90° for an octahedron).

42 TABLE S-IV. Structural parameters (Å, °) of the complexes **3** and **4**, as well as the ligand Hmpt

	[Co(Hmpt) ₂][CoCl ₄]·2H ₂ O (3)	[Co(Hmpt) ₂]Br ₂ ·Me ₂ CO (4)	Hmpt
Co1–O1A	2.1803(19)	2.1472(16)	
Co1–O1B	2.1878(19)	2.1857(15)	
Co1–N1A	2.092(2)	2.1009(16)	
Co1–N1B	2.077(2)	2.0751(16)	
Co1–S1A	2.3931(8)	2.3879(7)	
Co1–S2B	2.3888(8)	2.3924(6)	
O1A–C1A	1.217(3)	1.227(3)	1.208(4)
O1B–C1B	1.220(3)	1.219(3)	
O2A–C1A	1.311(3)	1.310(3)	1.337(4)
O2B–C1B	1.311(3)	1.304(3)	
C1A–C2A	1.496(4)	1.491(3)	1.506(6)
C1B–C2B	1.487(4)	1.499(3)	
N1A–C2A	1.281(3)	1.280(3)	1.285(4)
N1B–C2B	1.290(3)	1.275(3)	
N1A–N2A	1.354(3)	1.350(2)	1.367(4)
N1B–N2B	1.349(3)	1.357(2)	
N2A–C3A	1.361(3)	1.367(3)	1.363(4)
N2B–C3B	1.359(3)	1.354(3)	
S1A–C3A	1.688(3)	1.698(2)	1.684(4)
S1B–C3B	1.698(3)	1.703(2)	
N3A–C3A	1.315(3)	1.305(3)	1.325(5)
N3B–C3B	1.314(3)	1.307(3)	

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44 TABLE S-V. Hydrogen-bond geometry (Å, °) in complexes **3** and **4**

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>	Symmetry operation on <i>A</i>
3					
N2A–H2A···Cl1	0.86	2.47	3.185(2)	141	
N2B–H2B···Cl2 ⁱ	0.86	2.44	3.259(2)	159	<i>x</i> +1/2, <i>y</i> –1/2, – <i>z</i> +1
N3A–H3A···Cl1	0.86	2.58	3.271(3)	138	
N3A–H3B···Cl3 ⁱⁱ	0.86	2.38	3.232(3)	171	– <i>x</i> +1, – <i>y</i> +1, – <i>z</i> +1
N3B–H3D···Cl2 ⁱ	0.86	2.66	3.442(3)	152	<i>x</i> +1/2, <i>y</i> –1/2, – <i>z</i> +1
N3B–H3D···O1A ⁱⁱⁱ	0.86	2.57	3.079(3)	119	– <i>x</i> +1, – <i>y</i> , – <i>z</i> +1
N3B–H3C···Cl4 ^{iv}	0.86	2.39	3.244(3)	176	<i>x</i> , – <i>y</i> +1/2, <i>z</i> –1/2
4					
N2A–H2A···Br1	0.86(2)	2.53(2)	3.3221(18)	155(2)	
N3A–H3A···Br1	0.85(2)	2.50(2)	3.290(2)	154(3)	
N3A–H3B···Br2 ⁱ	0.86(2)	2.50(2)	3.350(2)	173(3)	– <i>x</i> +1, – <i>y</i> +1, – <i>z</i> +1
N2B–H2B···Br2	0.84(2)	2.46(2)	3.2664(17)	160(2)	
N3B–H3D···Br2	0.85(2)	2.65(2)	3.427(2)	153(2)	
N3B–H3C···Br1 ⁱⁱ	0.85(2)	2.56(2)	3.405(2)	176(2)	– <i>x</i> +1, – <i>y</i> +2, – <i>z</i> +1

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