

SUPPLEMENTARY MATERIAL TO
Synthesis, characterization, antimicrobial screening and cytotoxic properties of Cu(II) and Zn(II) complexes with a bidentate hydroxylated 1,3-diaryl-2-propene-1-one ligand

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TABLE S-I. Preparation of complexes C1–C10

Metal chloride		Ligand*	Complex ^{a,b}	
Amount taken for synthesis, mg (0.25 mmol)	Molecular formula	Amount taken for synthesis, mg (0.50 mmol)	Molecular formula	Yield, %
43	C ₁₉ H ₁₁ O ₃ I ₃	334	CuC ₃₈ H ₂₀ O ₆ I ₆	58
43	C ₁₉ H ₁₂ O ₂ BrCl	194	CuC ₃₈ H ₂₂ O ₄ Br ₂ Cl ₂	60
43	C ₁₉ H ₁₂ O ₂ BrCl ₂	211	CuC ₃₈ H ₂₀ O ₄ Br ₂ Cl ₄	62
43	C ₁₉ H ₁₂ O ₂ BrCl ₂	211	CuC ₃₈ H ₂₀ O ₄ Br ₂ Cl ₄	54
43	C ₁₉ H ₁₃ O ₄ Br	193	CuC ₃₈ H ₂₄ O ₈ Br ₂	63
34	C ₁₉ H ₁₅ O ₂ Br	192	ZnC ₄₀ H ₂₈ O ₆ Br ₂	61
34	C ₁₉ H ₁₂ O ₂ Br ₂	216	ZnC ₃₈ H ₂₂ O ₄ Br ₄	59
34	C ₁₉ H ₁₃ O ₃ Br	185	ZnC ₃₈ H ₂₄ O ₆ Br ₂	57
34	C ₁₉ H ₁₂ O ₂ BrF	186	ZnC ₃₈ H ₂₂ O ₄ Br ₂ F ₂	58
34	C ₁₉ H ₁₃ O ₄ I	216	ZnC ₃₈ H ₂₄ O ₈ I ₂	62

* Physical data of ligand.¹

CHARACTERIZATION

Bis((2-((E)-3-(2-hydroxy-3,5-diiodophenyl)acryloyl)-4-iodonaphthalen-1-yl)oxy)copper(II) (C1). Yield: 58 %; brown solid; m.p.: >300 °C decomposition; FTIR (KBr, cm⁻¹): 3434.8 (s, OH), 1643.2 (m, C=O), 1433.2 (m, C=C), 1207.2 (m, CO), 558.3 (m, CuO); ¹H-NMR (500 MHz, DMSO, δ / ppm): 5.41 (2H, s), 7.11 (2H, d, J = 16 Hz), 7.74 (2H, d, J = 16 Hz), 7.76–8.57 (14H, m); ¹³C-NMR (125 MHz, DMSO, δ / ppm): 89.38, 100.22, 108.07, 113.10, 123.62, 136.50, 140.74, 147.81, 151.74, 164.62, 172.00, 183.31, 188.50; ESR: g_{||}: 2.2497, g_⊥: 2.0972.

Bis((4-bromo-2-((E)-3-(4-chlorophenyl)acryloyl)naphthalen-1-yl)oxy)copper(II) (C2). Yield: 60 %; brown solid; m.p.: >300 °C decomposition; FTIR (KBr, cm⁻¹): 3410.6 (s, OH), 1523.5 (m, C=O), 1491.2 (m, C=C), 1256.5 (m, C-O), 526.5 (m, Cu-O); ¹H-NMR (500 MHz, DMSO, δ / ppm): 7.28 (2H, d, J = 16 Hz), 7.60 (2H, d, J = 16 Hz), 7.76–8.36 (18H, m); ¹³C-

NMR (125 MHz, DMSO, δ / ppm): 92.83, 104.14, 109.80, 125.98, 134.46, 145.14, 148.91, 159.75, 174.20, 177.19.

Bis((4-bromo-2-((E)-3-(2,6-dichlorophenyl)acryloyl)naphthalen-1-yl)oxy)copper(II) (**C3**). Yield: 62 %; brown; m.p.: >300 °C decomposition; FTIR (KBr, cm^{-1}): 3463.7 (s, OH), 1635.1 (m, C=O), 1606.2 (m, C=C), 1255.7 (m, C-O), 599.4 (m, Cu-O).

Bis((4-bromo-2-((E)-3-(2,4-dichlorophenyl)acryloyl)naphthalen-1-yl)oxy)copper(II) (**C4**). Yield: 54 %; brown; m.p.: >300 °C decomposition; FTIR (KBr, cm^{-1}): 3463.8 (s, OH), 1633.8 (m, C=O), 1572.7 (m, C=C), 1251.2 (m, C-O), 601.9 (m, Cu-O); $^1\text{H-NMR}$ (500 MHz, DMSO, δ / ppm): 7.10 (2H, *d*, *J* = 16 Hz), 7.53 (2H, *d*, *J* = 16 Hz), 7.67-8.72 (16H, *m*); $^{13}\text{C-NMR}$ (500 MHz, DMSO, δ / ppm): 82.94, 92.21, 106.19, 108.86, 128.18, 131.01, 151.58, 156.77, 160.54, 169.18, 171.69, 180.17, 182.69, 189.28.

Bis((4-bromo-2-((E)-3-(3,4-dihydroxyphenyl)acryloyl)naphthalen-1-yl)oxy)copper(II) (**C5**): Yield: 63 %; brown; m.p.: >300 °C decomposition; FTIR (KBr, cm^{-1}): 3416.3 (s, OH), 1611.8 (m, C=O), 1585.5 (m, C=C), 1253.5 (m, C-O), 594.1 (m, Cu-O).

Bis((4-bromo-2-((E)-3-(4-fluorophenyl)acryloyl)naphthalen-1-yl)oxy)zinc(II) (**C9**). Yield: 58 %; brown; m.p.: >300 °C decomposition; FTIR (KBr, cm^{-1}): 3434.6 (s, OH), 1635.1 (m, C=O), 1547.5 (m, C=C), 1249.8 (m, C-O), 468.7 (m, Cu-O); $^1\text{H-NMR}$ (500 MHz, DMSO, δ / ppm): 6.88 (2H, *d*, *J* = 16Hz), 7.47 (2H, *d*, *J* = 16 Hz), 7.76-8.58 (18H, *m*); $^{13}\text{C-NMR}$ (500 MHz, DMSO, δ / ppm): 101.32, 117.34, 125.19, 126.92, 131.95, 141.22, 142.63, 152.84, 156.45, 159.91, 165.09, 170.12, 183.16, 185.20, 186.46; ESR: g_{\parallel} : 2.0509, g_{\perp} : 2.0066.

TABLE S-II. Crystal data and structure refinement

Complex	[Cu(C ₁₉ H ₁₀ O ₃ I ₃) ₂]	[Cu(C ₁₉ H ₁₀ O ₂ BrCl ₂) ₂]	Zn(C ₁₉ H ₁₂ O ₄ I) ₂
Empirical formula	Cu(C ₃₈ H ₂₀ O ₆ I ₆)	Cu(C ₃₈ H ₂₀ O ₄ Br ₂ Cl ₄)	ZnC ₃₈ H ₂₄ O ₈ I ₂
Formula weight	1397.53	905.73	927.78
Temperature, K	298	298	298
Crystal system	Orthorhombic	Orthorhombic	Orthorhombic
Space group	P b c a	P c a 21	P n m a
<i>a</i> / Å	16.8478	14.6387	22.2520
<i>b</i> / Å	12.4058	10.2508	17.6363
<i>c</i> / Å	27.8294	18.8246	11.3842
α / °	90	90	90
β / °	90	90	90
γ / °	90	90	90
Volume, Å ³	5816.64	2824.79	4467.65
Z	16	4	4
ρ_{calc} g / cm ⁻³	6.383	2.130	1.379
μ / cm ⁻¹	5.744	38.815	30.222
Crystallite size, nm	24.70794	29.28896	13.16591
Dislocation density, nm ⁻²	1.63805	1.16571	5.76897
Micro strain	12.93231	7.72544	19.39073

MTT Assay

In-vitro anticancer activity was evaluated by MTT assay. The growth inhibitory activity of complexes was determined against liver cancer cells (HepG2). Activity data is presented in Fig. 26.

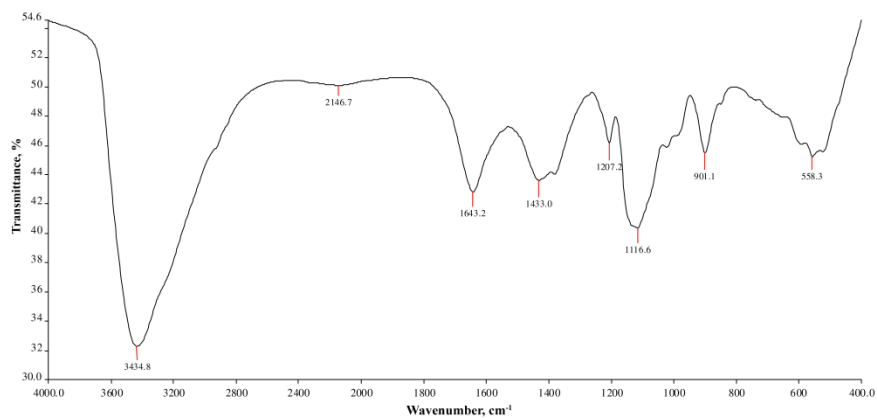


Figure S-1. FTIR Spectrum of complex C1

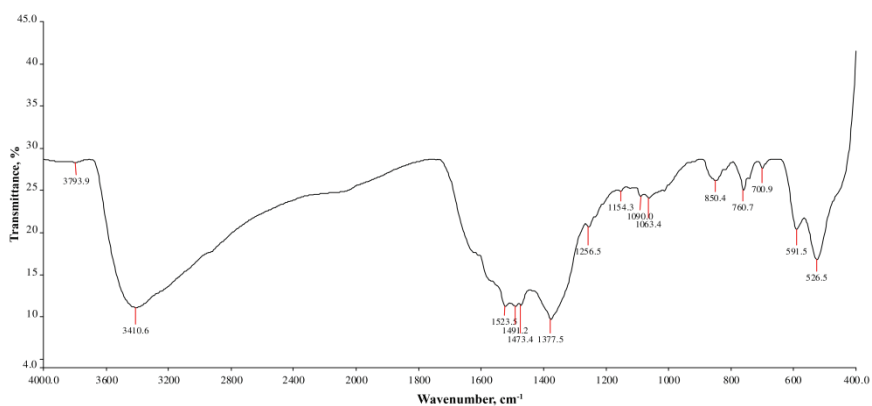


Figure S-2. FTIR Spectrum of complex C2

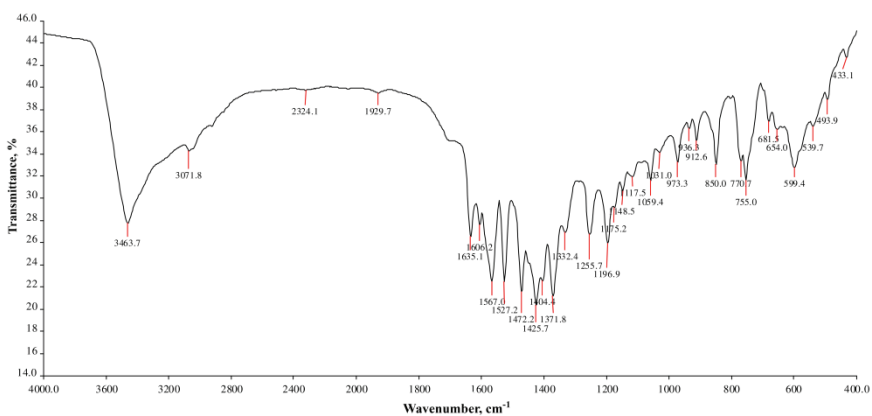


Figure S-3. FTIR Spectrum of complex C3

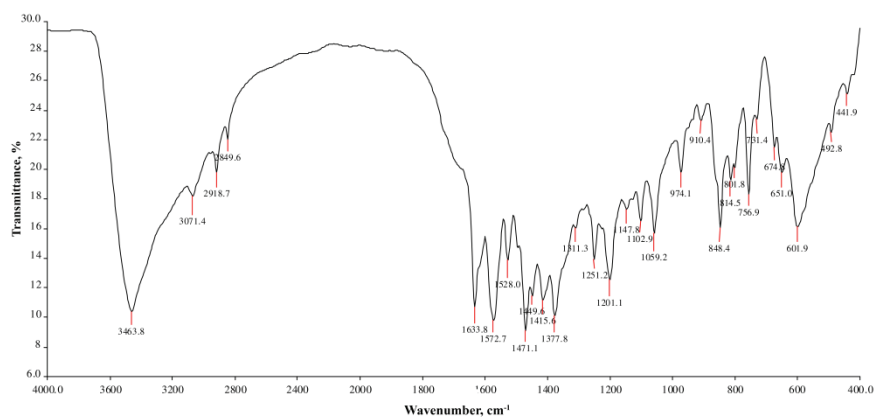


Figure S-4. FTIR Spectrum of complex C4

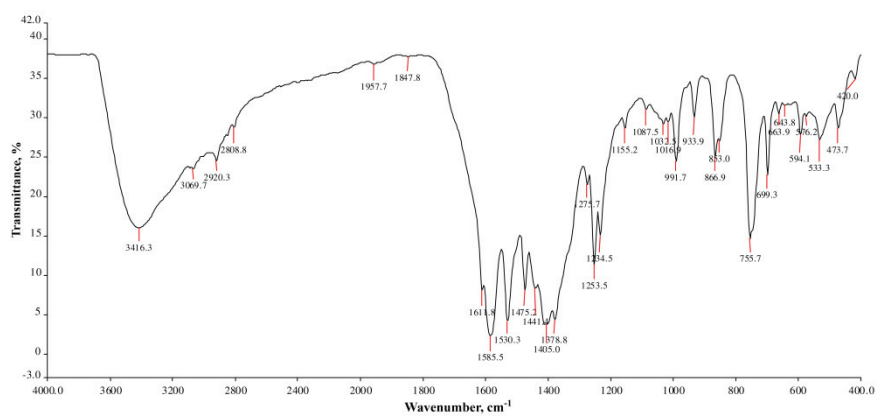


Figure S-5. FTIR Spectrum of complex C5

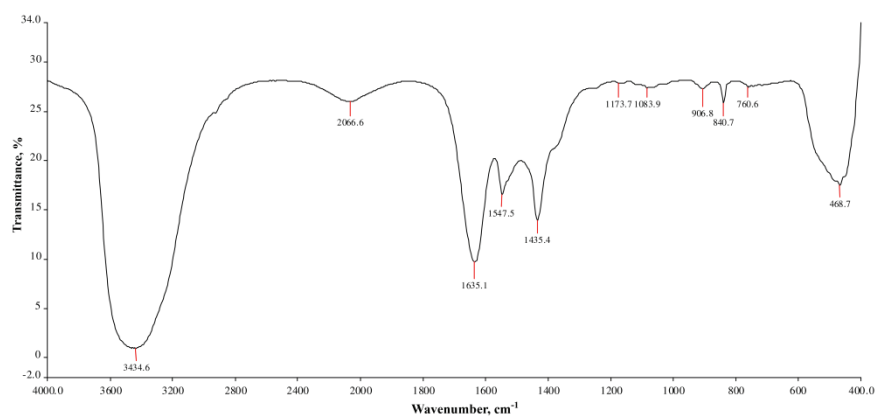


Figure S-6. FTIR Spectrum of complex C9

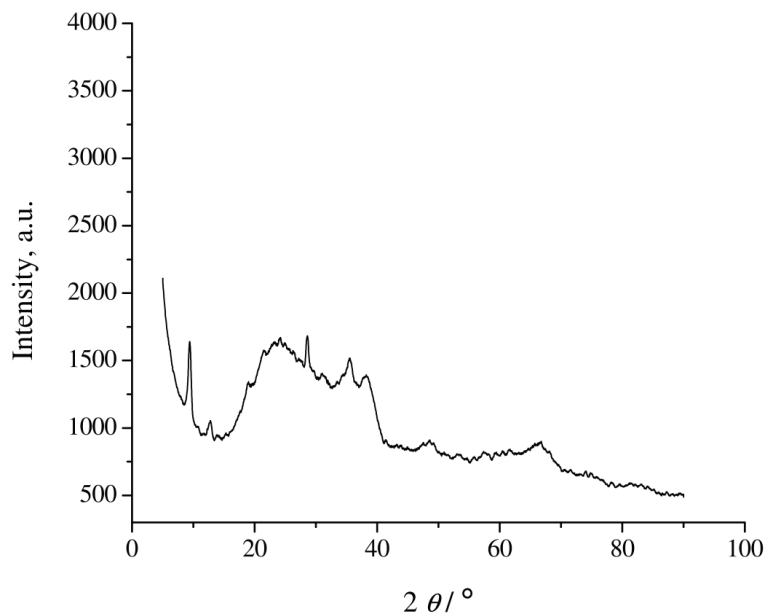


Figure S-7. XRD of complex C1

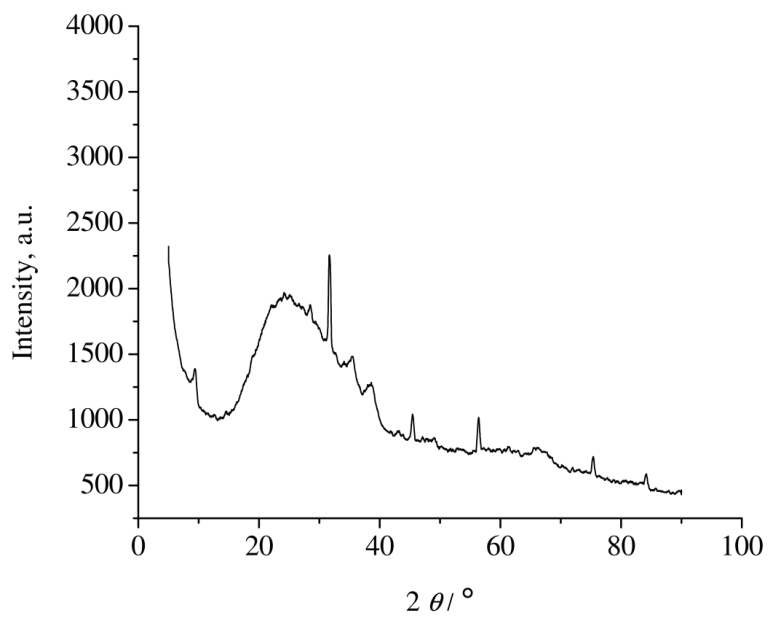


Figure S-8. XRD of complex C2

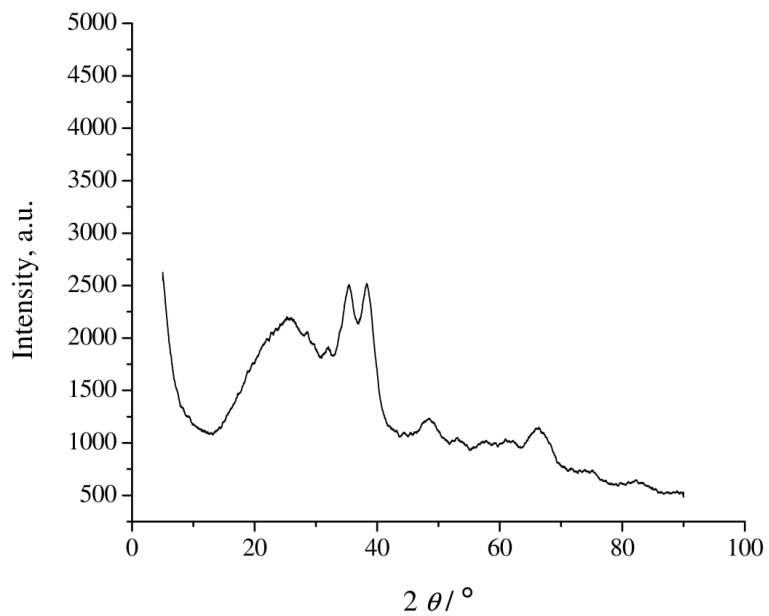


Figure S-9. XRD of complex C4

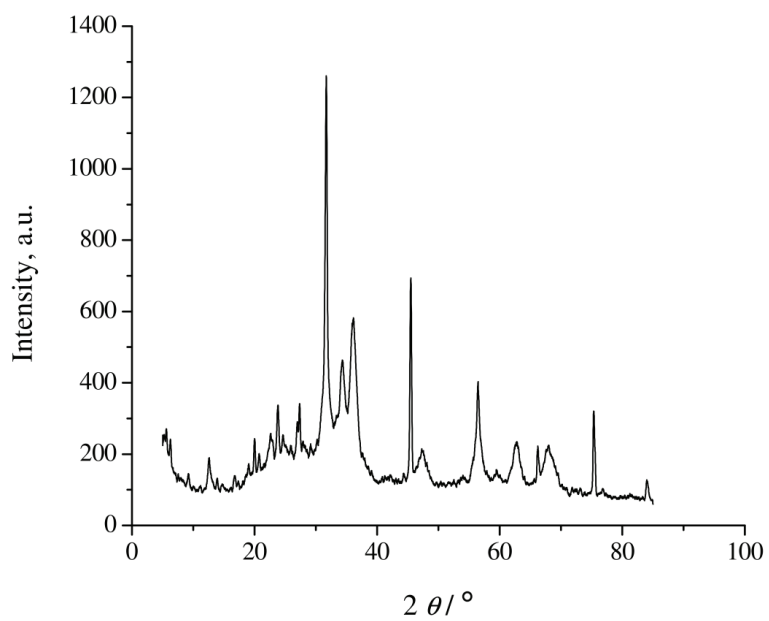


Figure S-10. XRD of complex C9

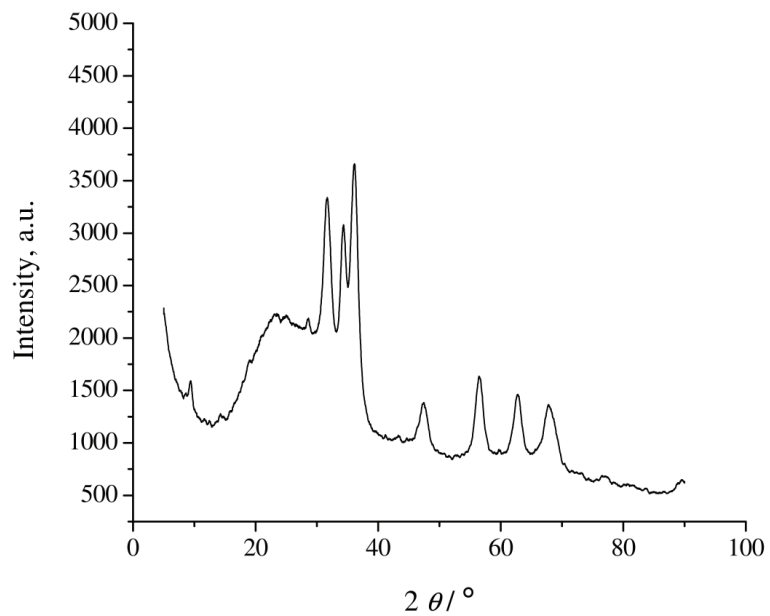


Figure S-11. XRD of complex C10

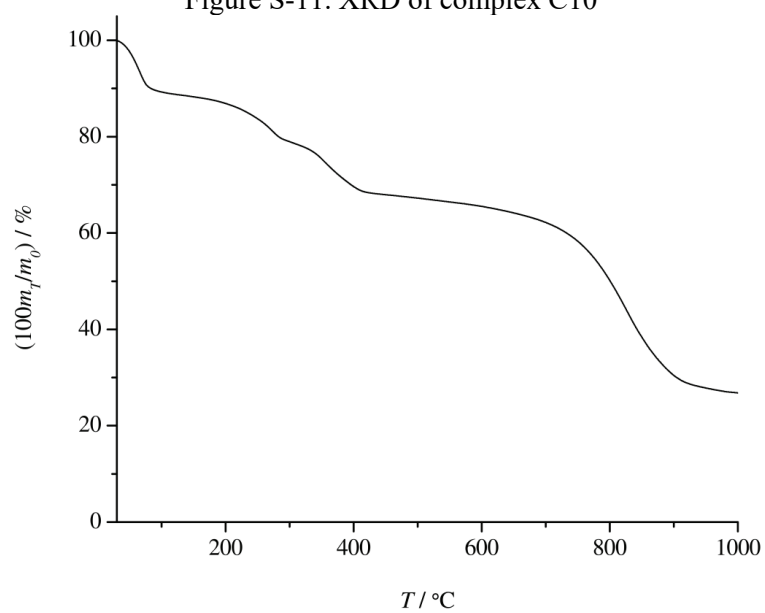


Figure S-12. TGA of complex C1

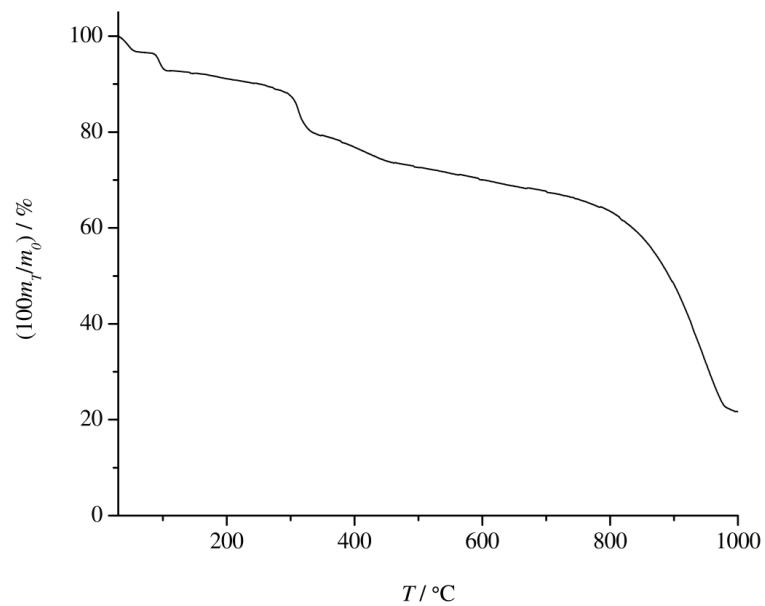


Figure S-13 TGA of complex C3

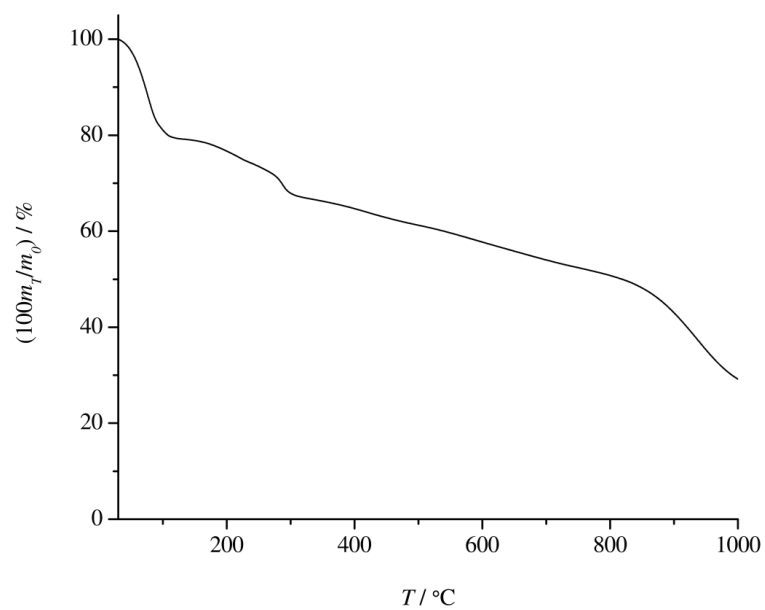


Figure S-14. TGA of complex C4

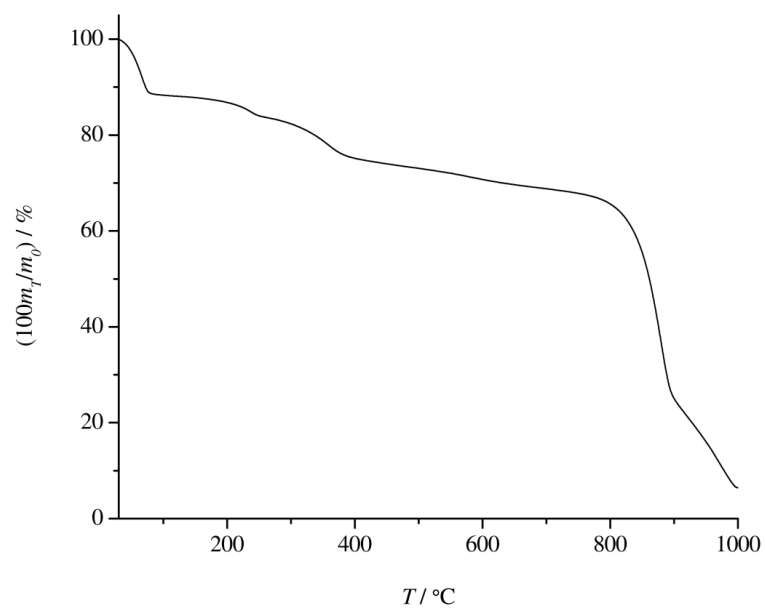


Figure S-15. TGA of complex C9

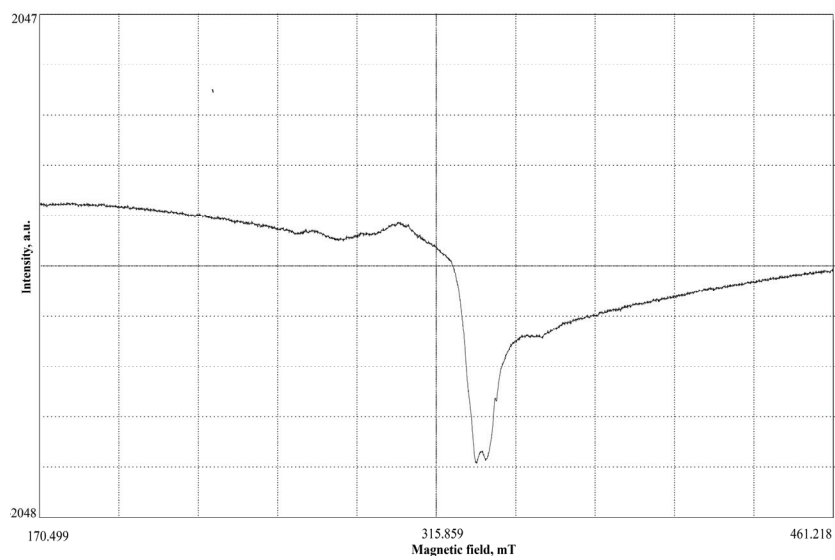


Figure S-16. ESR of complex C1

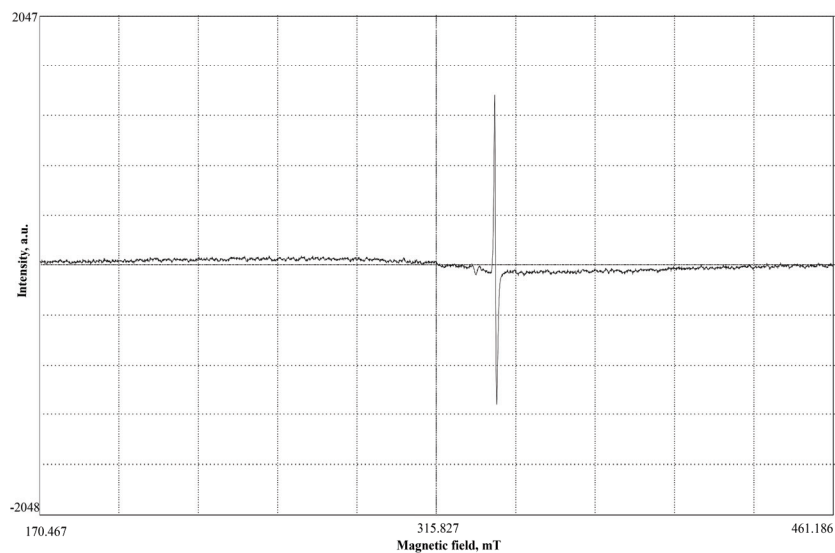


Figure S-17. ESR of complex C9

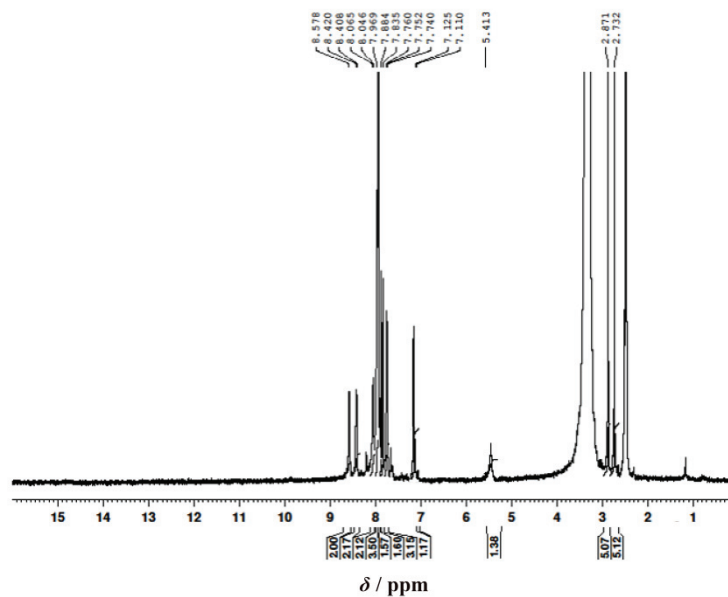
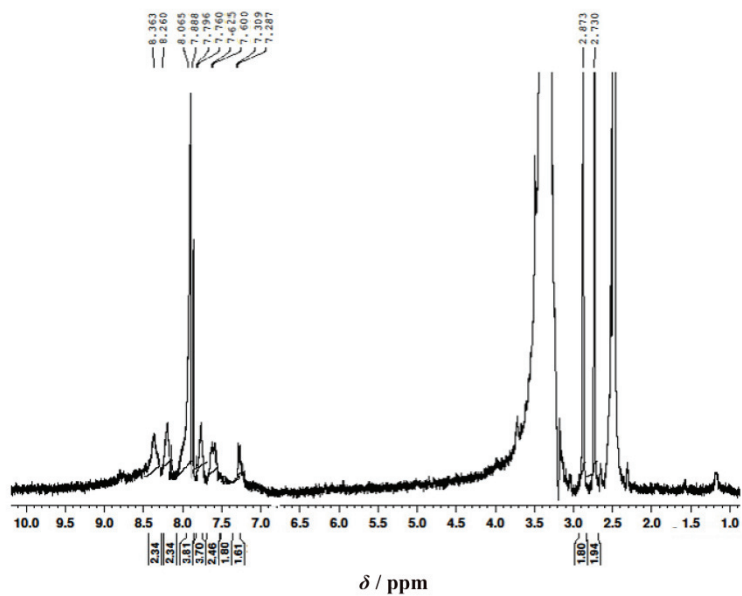
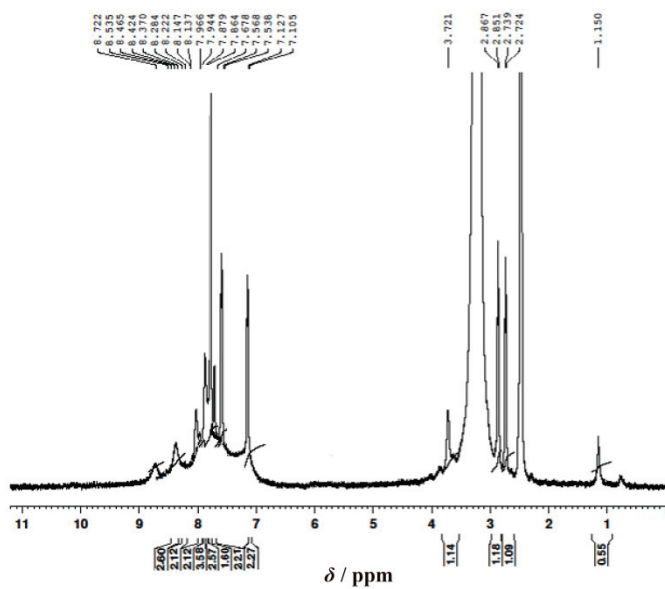


Figure S-18. ¹H NMR spectrum of complex C1

Figure S-19. ¹H NMR spectrum of complex C2Figure S-20. ¹H NMR spectrum of complex C4

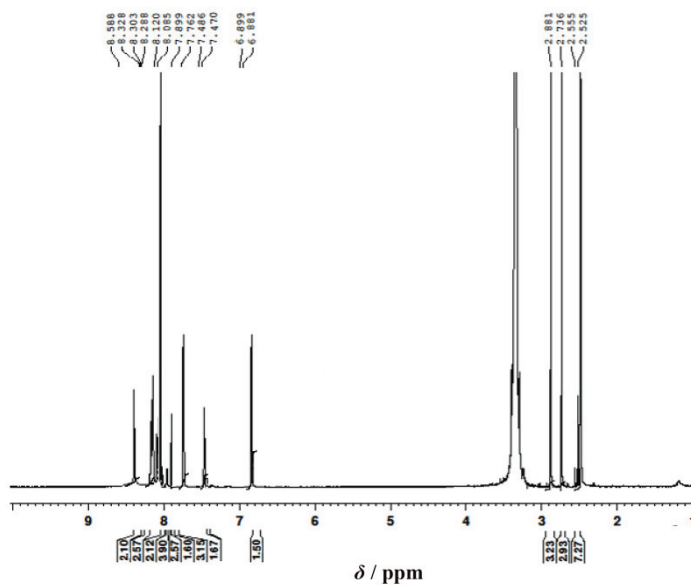


Figure S-21. ¹H NMR spectrum of complex C9

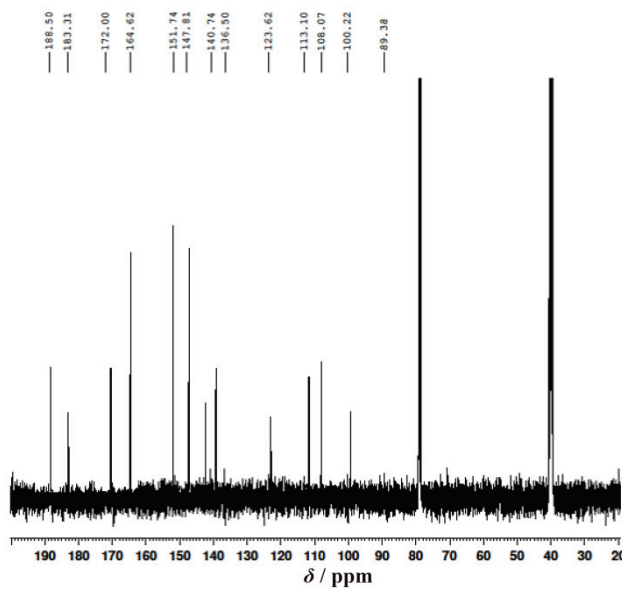
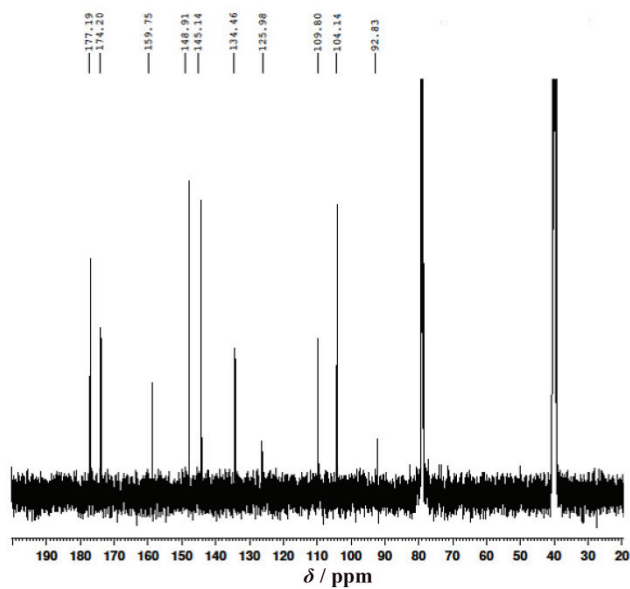
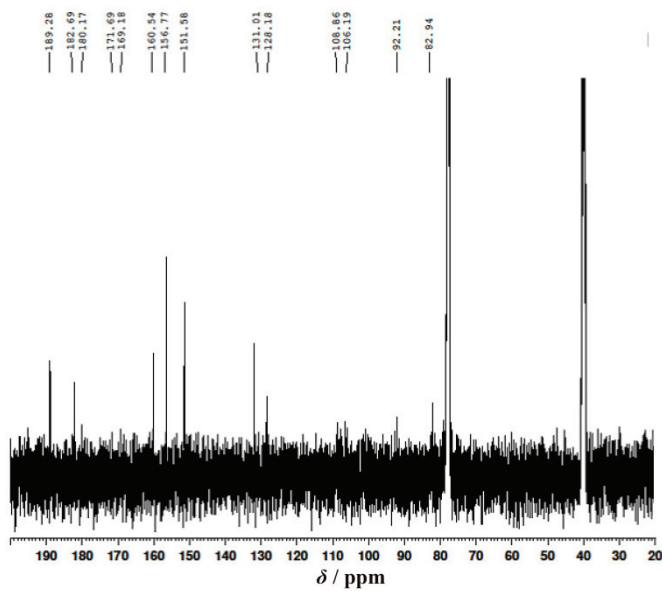


Figure S-22. ¹³C NMR spectrum of complex C1

Figure S-23. ¹³C NMR spectrum of complex C2Figure S-24. ¹³C NMR spectrum of complex C4

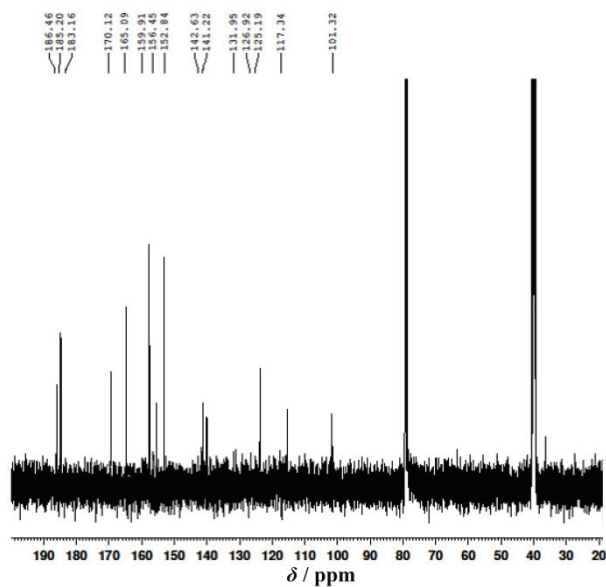
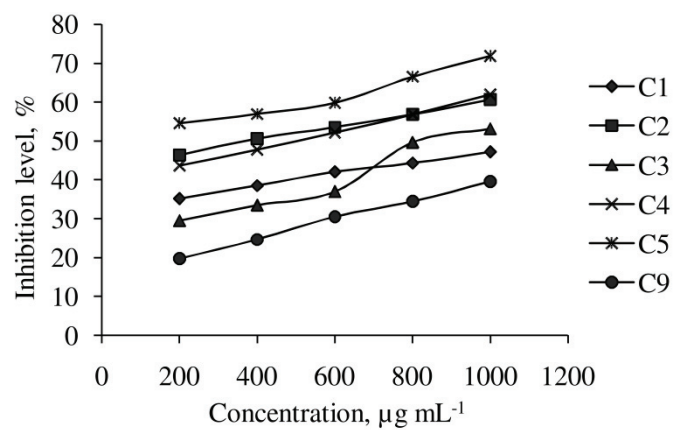
Figure S-25. ^{13}C NMR spectrum of complex C9

Figure S-26. Graph of sample concentration against the inhibition level

REFERENCE

1. P. Patil, P. A. Khan, S. Zangade, *Curr. Chem. Lett.* **9** (2020) 183–198 (http://www.growing-science.com/ccl/Vol9/ccl_2020_5.pdf)